

Article

The Conversion Polymorphism of Perovskite Phases in the BiCrO₃–BiFeO₃ System

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Abstract: Perovskite-type materials containing Bi³⁺ cations at A sites are interesting from the viewpoints of applications and fundamental science as the lone pair of Bi³⁺ cations often stabilizes polar, ferroelectric structures. This can be illustrated by a lot of discoveries of different new functionalities in bulk and thin films of BiFeO₃ and its derivatives. In this work, we investigated solid solutions of BiCr_{1-x}Fe_xO₃ with 0.1 ≤ x ≤ 0.4 prepared by a high-pressure (HP) method and post-synthesis annealing at ambient pressure (AP). HP-BiCr_{1-x}Fe_xO₃ modifications with 0.1 ≤ x ≤ 0.3 were mixtures of two phases with space groups C2/c and Pbam, and the amount of the C2/c phase decreased with increasing x. The amount of the C2/c phase was also significantly decreased in AP-BiCr_{1-x}Fe_xO₃ modifications, and the C2/c phase almost disappeared in AP-BiCr_{1-x}Fe_xO₃ with 0.2 ≤ x ≤ 0.3. Fundamental, strong reflections of HP-BiCr_{1-x}Fe_xO₃ and AP-BiCr_{1-x}Fe_xO₃ were almost unchanged; on the other hand, weak superstructure reflections were different and showed clear signs of strong anisotropic broadening and incommensurate positions. These structural features prevented us from determining their room-temperature structures. On the other hand, HP-BiCr_{1-x}Fe_xO₃ and AP-BiCr_{1-x}Fe_xO₃ showed high-temperature structural phase transitions to the GdFeO₃-type Pnma modification at T_{srt} = 450 K (x = 0.1), T_{srt} = 480 K (x = 0.2), T_{srt} = 510 K (x = 0.3), and T_{srt} = 546 K (x = 0.4). Crystal structures of the GdFeO₃-type Pnma modifications of all the samples were investigated by synchrotron powder X-ray diffraction. Magnetic properties of HP-BiCr_{1-x}Fe_xO₃ and AP-BiCr_{1-x}Fe_xO₃ were quite close to each other (HP vs. AP), and the x = 0.2 samples demonstrated negative magnetization phenomena without signs of the exchange bias effect.

Keywords: multiferroics; doped BiFeO₃; BiCrO₃; high-pressure synthesis; irreversible transformations; conversion polymorphism; incommensurate structures



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1. Introduction

Bi³⁺ cations have a lone electron pair similar to Pb²⁺ cations and, therefore, are often considered as a replacement of toxic lead in ferroelectric and piezoelectric materials. BiFeO₃ perovskite has been known since the late 1950s [1–4]. However, it has received tremendous interest in the 2000s [5,6] after the discovery that thin films of BiFeO₃ can have good ferroelectric properties and are effective and model multiferroic materials [7–9]. Since then, thousands of publications have been devoted to BiFeO₃ perovskite and its derivatives [6]. A lot of fundamental science and practical discoveries have recently been made using BiFeO₃ perovskite [10–14], for example, related to manipulations of charge domain walls, chiral spin transport, and spin cycloids, the formation of skyrmion lattices, and so on.

BiFeO₃ crystallizes in space group R3c below its ferroelectric transition temperature T_{FE} = 1100 K, and it shows a number of high-temperature structural transitions [6]. Its first transition at T_{FE} is to a GdFeO₃-type Pnma phase [15], not to a centrosymmetric pair

(space group $R\bar{3}c$) of space group $R3c$ as one would expect. The ferroelectricity of BiFeO_3 originates from the activity of the lone electron pair of Bi^{3+} cations [6]. The pure antiferromagnetic (AFM) transition occurs at the Néel temperature $T_N = 643$ K and originates from superexchange interactions between magnetic Fe^{3+} cations [6]. The absence of any net ferromagnetic (FM) moments in BiFeO_3 originates from long-period incommensurate spin ordering, which averages the total net moment to zero [6]. BiFeO_3 perovskite is the only compound among other simple BiMO_3 perovskites that can be prepared in bulk form under ambient pressure (AP) conditions.

BiCrO_3 perovskite has been known since the middle of the 1960s [16–18], but it received renewed interest in the 2000s [19–31] as a counterpart to BiFeO_3 . It crystallizes in space group $C2/c$ at room temperature (RT) and shows a structural phase transition to a GdFeO_3 -type $Pnma$ modification above $T_{\text{str}} = 420$ K [23,24]. BiCrO_3 demonstrates an AFM transition at $T_N = 112$ K to a G-type AFM structure with small spin canting (and, therefore, with the appearance of weak FM properties) [23–26,30,31], and there is a spin-reorientation transition near 72–75 K with an increased weak FM moment. BiCrO_3 perovskite can only be stabilized in bulk form under high-pressure (HP) synthesis conditions [32].

Other simple BiMO_3 perovskites have been studied less because they need an HP high-temperature method for their preparation in bulk form [32]. It is interesting that there are many structural variations in simple BiMO_3 perovskites [27,32]: samples with $M = \text{Sc}$, Cr , and Mn crystallize in space group $C2/c$; $M = \text{Al}$ and Fe samples—in polar space group $R3c$; the $M = \text{Co}$ sample—in polar space group $P4mm$; the $M = \text{Ni}$ sample—in space group $P\bar{1}$; the $M = \text{Ga}$ sample—in a pyroxene-type (non-perovskite) $Pcca$ structure; and the $M = \text{Rh}$ sample—in space group $Pnma$ [32]. Therefore, different $\text{BiM}_{1-x}\text{M}_2\text{O}_3$ solid solutions can show complex structural variations and transformations as a function of composition x and temperature. Such solid solutions have received a tremendous amount of interest in the literature [33–51]. For example, the appearance of polar $R3c$ phases was observed in the solid solution system between centrosymmetric BiCrO_3 and BiGaO_3 [47,50]. In the solid solutions between BiFeO_3 and BiCrO_3 , one composition has been investigated a lot, namely $\text{Bi}_2\text{FeCrO}_6$, after the first-principle predictions [52] that this composition (with ordered arrangements of Fe^{3+} and Cr^{3+} cations) should show large ferroelectric polarization and large magnetization [53–58]. In particular, thin films of $\text{Bi}_2\text{FeCrO}_6$ have attracted a lot of attention from the application point of view as photovoltaic materials [57,58]. The number of investigations focusing on properties of bulk $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions is very limited because such solid solutions need an HP method for their preparation [59–64], and the majority of such studies are focused on studies of Fe-rich samples [60–62].

Therefore, in this work we investigated samples from the Cr-rich side of bulk $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions with $0.1 \leq x \leq 0.4$. We found structural phase transitions to the $Pnma$ modification with systematic changes in the phase transition temperature. We found that the as-synthesized samples do not transform back into their initial state after cooling from high temperatures, therefore realizing the conversion polymorphism phenomenon [41]. Weak incommensurate structural modulations were found in $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions with $0.1 \leq x \leq 0.4$. Detailed magnetic properties are reported.

2. Results and Discussion

All $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ samples contained small amounts of $\text{Bi}_2\text{O}_2\text{CO}_3$ impurity. The $x = 0.1$ and 0.2 samples additionally contained very small amounts of Cr_2O_3 impurity; the amount of Cr_2O_3 impurity was at the background level in the $x = 0.3$ and 0.4 samples. HP- $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ samples with $x = 0.1$, 0.2 , and 0.3 consisted of two perovskite-type modifications: one modification had the $C2/c$ symmetry as found in undoped BiCrO_3 [24], and the second modification had the PbZrO_3 -type structure (space group $Pbam$ (No. 55)

and a $\sqrt{2}a_p \times 2\sqrt{2}a_p \times 2a_p$ superstructure, where a_p (≈ 3.95 Å) is the parameter of the cubic perovskite subcell [62,65,66]). The amount of the BiCrO_3 -type $C2/c$ modification rapidly decreased with increasing x from about 75% ($x = 0.1$) to about 50% ($x = 0.2$) and about 5% ($x = 0.3$) in the HP samples (Figures 1 and 2). The amount of the BiCrO_3 -type $C2/c$ modification also rapidly decreased with increasing x from about 40% ($x = 0.1$) to trace amounts ($x = 0.2$) to undetectable amounts ($x = 0.3$) in the AP samples (Figures 1 and 2). Both HP- and AP- $\text{BiCr}_{0.6}\text{Fe}_{0.4}\text{O}_3$ contained only a perovskite phase with the PbZrO_3 -type structure (Figure 2b).

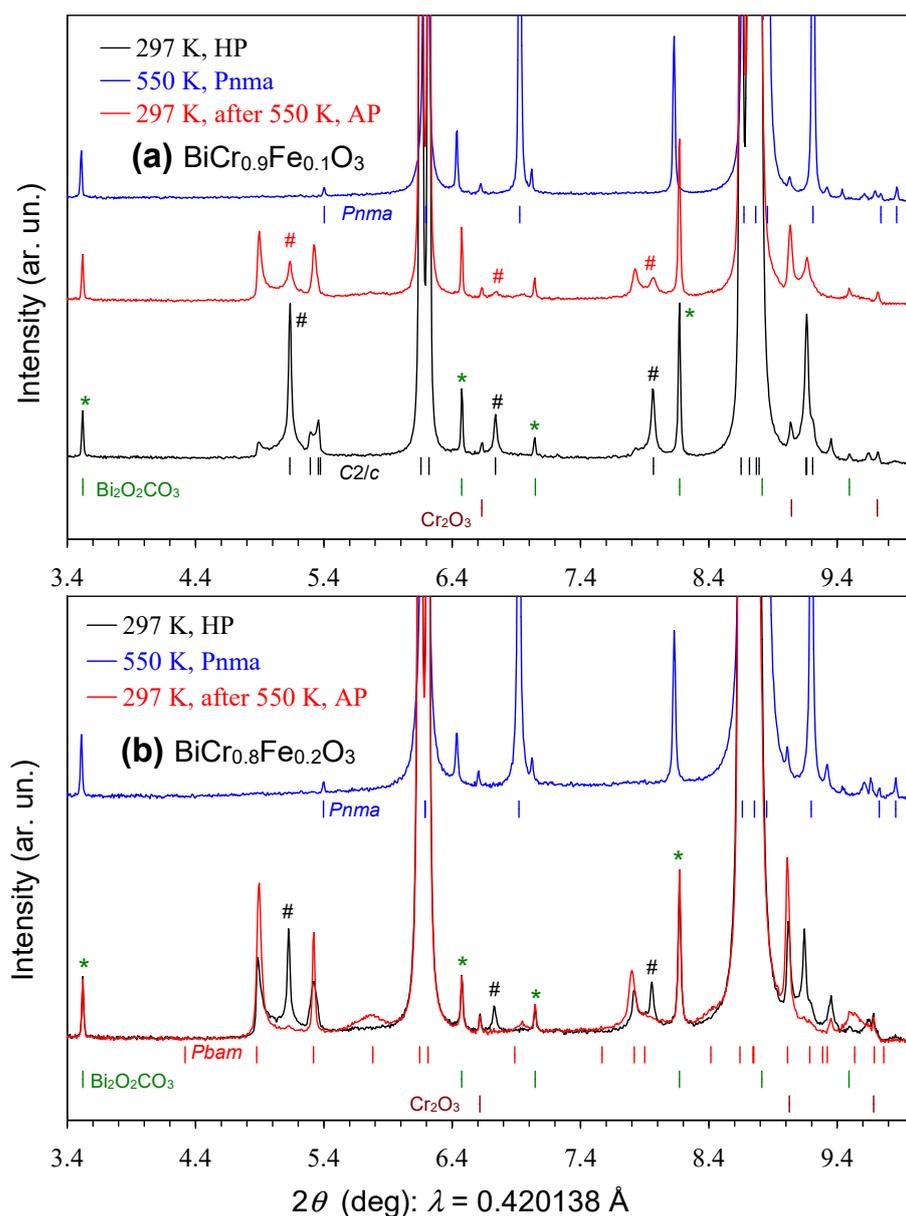


Figure 1. Magnified fragments of experimental synchrotron powder X-ray diffraction patterns of (a) $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$ and (b) $\text{BiCr}_{0.8}\text{Fe}_{0.2}\text{O}_3$. Patterns for the as-synthesized HP modifications at $T = 297$ K are shown by black lines, for the $Pnma$ modifications at $T = 550$ K—by blue lines, and the AP modifications at $T = 297$ K—by red lines. The tick marks show possible Bragg reflection positions for the main perovskite phases ($C2/c$, $Pnma$, and $Pbam$) and $\text{Bi}_2\text{O}_2\text{CO}_3$ and Cr_2O_3 impurities. The characteristic reflections of the $C2/c$ modification are additionally marked by black/red octothorps. The characteristic reflections of the $\text{Bi}_2\text{O}_2\text{CO}_3$ impurity are additionally marked by green stars.

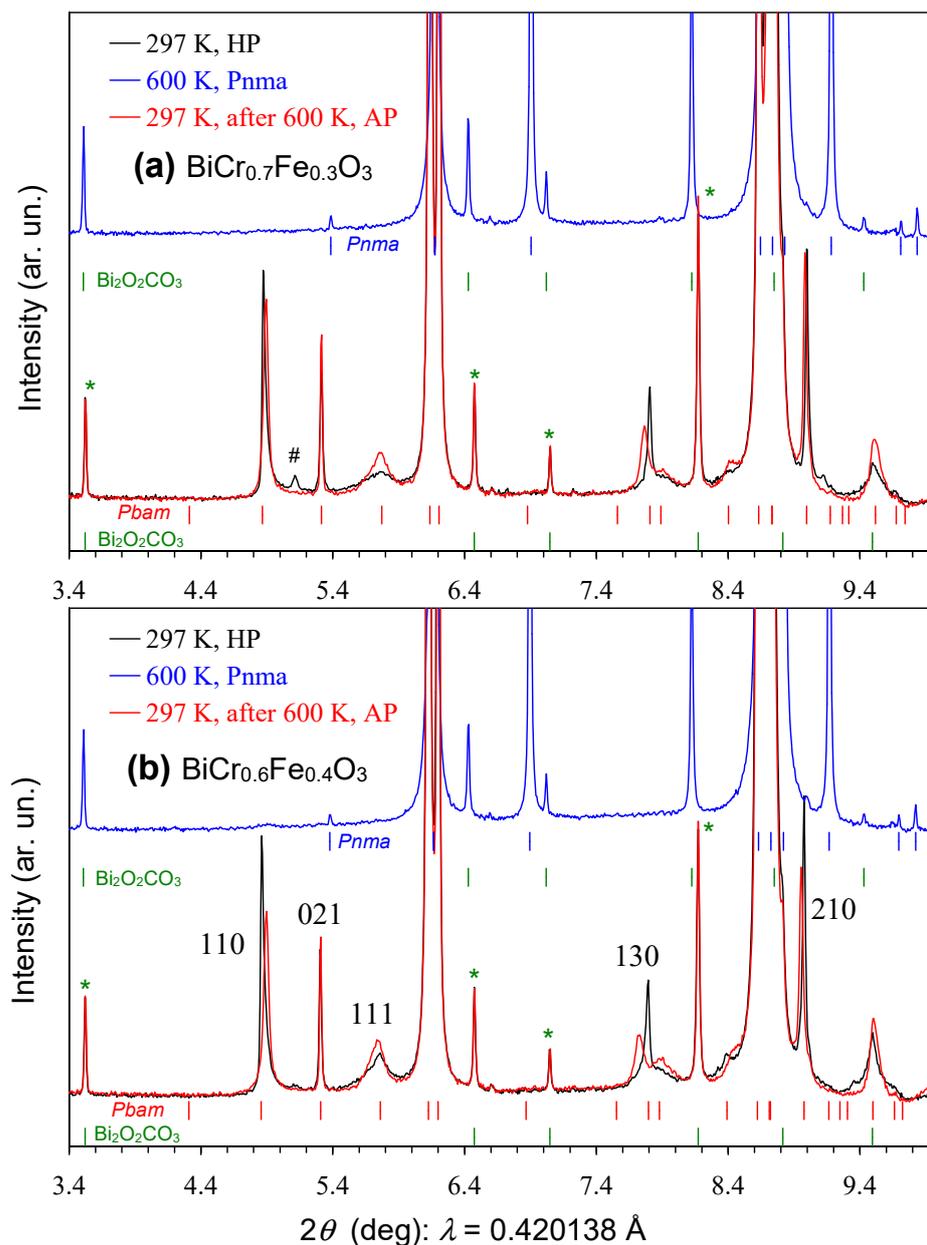


Figure 2. Magnified fragments of experimental synchrotron powder X-ray diffraction patterns of (a) $\text{BiCr}_{0.7}\text{Fe}_{0.3}\text{O}_3$ and (b) $\text{BiCr}_{0.6}\text{Fe}_{0.4}\text{O}_3$. Patterns for the as-synthesized HP modifications at $T = 297$ K are shown by black lines, for the $Pnma$ modifications at $T = 600$ K—by blue lines, and the AP modifications at $T = 297$ K—by red lines. The tick marks show possible Bragg reflection positions for the main perovskite phases ($Pnma$ and $Pbam$) and $\text{Bi}_2\text{O}_2\text{CO}_3$ impurity. The characteristic reflections of the $C2/c$ modification are marked by black octothorps. The characteristic reflections of the $\text{Bi}_2\text{O}_2\text{CO}_3$ impurity are additionally marked by green stars. (hkl) indices of some superstructure reflections of the $Pbam$ -related modification are given.

Figure 3 shows the results of the DSC measurements. All samples demonstrated a reversible phase transition with sharp peaks on the DSC heating curves at $T_{\text{str}} = 450$ K ($x = 0.1$), 480 K ($x = 0.2$), 510 K ($x = 0.3$), and 546 K ($x = 0.4$), where T_{str} stands for the structural phase transition temperature. The first heating curve was slightly different from the second and third heating curves; all cooling curves were almost identical. The small difference between the heating curves can be explained by the fact the HP modification transforms to the AP modification after the first DSC run. The synchrotron XRD data (Figures 1 and 2; blue curves) clearly showed that the DSC anomalies corre-

spond to a structural phase transition to the $Pnma$ modification. With $T_{\text{str}} = 420$ K for undoped BiCrO_3 [23,24], T_{str} almost linearly increases with increasing x in the $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions with $0.0 \leq x \leq 0.4$. All the samples showed comparable enthalpies (about 7.0–7.9 J/g) of the structural phase transition (Figure 3). However, a systematic increase in enthalpy with increasing x can be seen.

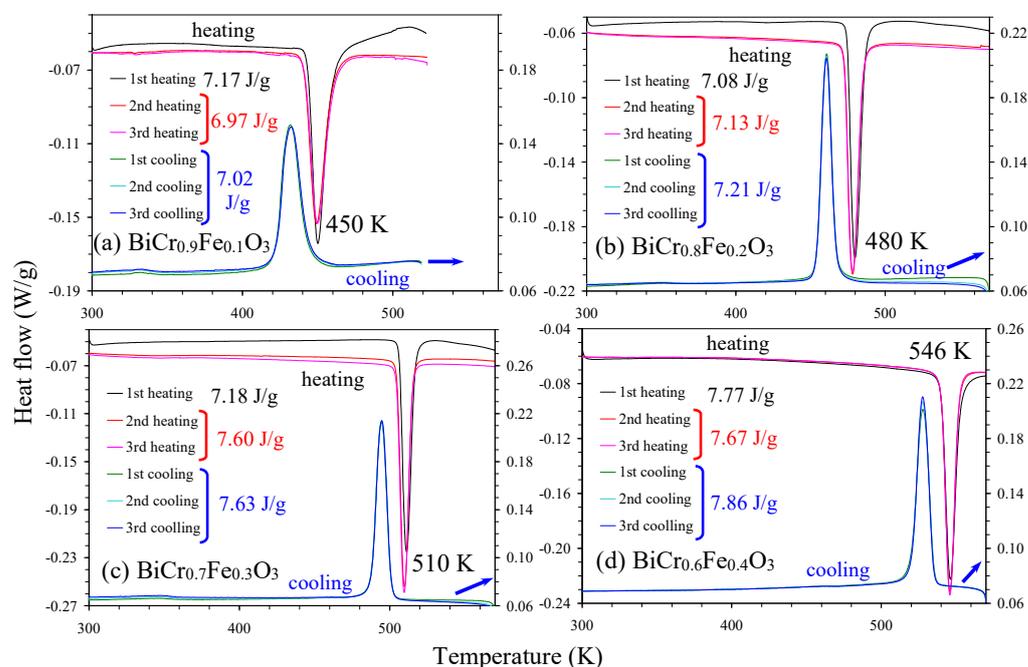


Figure 3. Differential scanning calorimetry (DSC) curves of (a) $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$, (b) $\text{BiCr}_{0.8}\text{Fe}_{0.2}\text{O}_3$, (c) $\text{BiCr}_{0.7}\text{Fe}_{0.3}\text{O}_3$, and (d) $\text{BiCr}_{0.6}\text{Fe}_{0.4}\text{O}_3$ on heating (the left-hand axes) and cooling (the right-hand axes). Three DSC runs are given for each sample. Temperatures of peak positions on the heating curves are given. The peak areas (in J/g) are also given for the first heating curve, for the second and third heating curve (an average value), and for the first, second, and third cooling curves (an average value).

The $Pnma$ modification of all the samples had sharp reflections without any anisotropic broadening or asymmetry. Therefore, structure parameters of the $Pnma$ modifications could be readily refined from synchrotron XRD data. Refined structural parameters of the $Pnma$ modifications for all the samples are summarized in Table 1, and Figure 4 shows fragments of experimental, calculated, and difference synchrotron XRPD data after the Rietveld fit at 550 K for the $x = 0.1$ sample as an example.

Figure 5 shows the compositional dependence of the lattice parameters and unit cell volume of the $Pnma$ modifications at 550 K (Table S1 provides numerical data). Nearly linear increases in all parameters were observed with the increase in the Fe content in agreement with the larger ionic radius of Fe^{3+} cations ($r_{\text{VI}} = 0.645$ Å) in comparison to Cr^{3+} cations ($r_{\text{VI}} = 0.615$ Å) [67], confirming the formation of the solid solutions.

On the other hand, superstructure reflections of the PbZrO_3 -type $Pbam$ modifications in all the samples showed strong anisotropic broadening, and some reflections showed asymmetry from the high-angle side of the reflections (Figures 1 and 2 and Figure S1). For example (Figure 2b), the (021) reflection had the same width as the fundamental, main reflections, and it was symmetrical. The (111) reflection was very broad, and it was nearly symmetrical. The (110) reflection had intermediate broadening, and it showed asymmetry. Moreover, some reflections of the AP modification showed noticeable shifts from their expected, commensurate positions (for example, the (110), (130), and (210) reflections). Such shifts suggest the presence of incommensurate modulations. Such shifts were also present

in the HP modifications, but the shifts were less pronounced. Therefore, the presence of incommensurate modulations and significant anisotropic broadening prevented us from determining the precise structural parameters of the HP and AP modifications at RT.

Table 1. Structure parameters of $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ at high temperatures from synchrotron powder X-ray diffraction data.

x	0.1	0.2	0.3	0.4
T (K)	550	550	600	600
a (Å)	5.55595 (3)	5.56458 (3)	5.57383 (2)	5.58329 (2)
b (Å)	7.77499 (6)	7.78492 (6)	7.80159 (3)	7.81360 (3)
c (Å)	5.44100 (3)	5.44552 (3)	5.45751 (2)	5.46389 (2)
V (Å ³)	235.037 (3)	235.899 (3)	237.318 (2)	238.366 (2)
ρ_{cal} (g/cm ³)	8.743	8.721	8.680	8.653
x (Bi)	0.04309 (6)	0.04275 (8)	0.04304 (7)	0.04281 (8)
z (Bi)	0.99574 (14)	0.99616 (23)	0.99618 (18)	0.99613 (21)
B_{iso} (Bi) (Å ²)	1.173 (7)	1.364 (11)	1.528 (10)	1.643 (11)
B_{iso} (Cr/Fe) (Å ²)	0.55 (2)	0.72 (3)	0.76 (2)	0.79 (3)
x (O1)	0.4800 (9)	0.4871 (12)	0.4852 (11)	0.4860 (12)
z (O1)	0.0823 (10)	0.0747 (13)	0.0847 (12)	0.0824 (13)
B_{iso} (O1) (Å ²)	0.62 (13)	0.10 (17)	0.88 (16)	0.55 (18)
x (O2)	0.2915 (9)	0.2921 (15)	0.2945 (11)	0.2994 (13)
y (O2)	0.0385 (6)	0.0413 (11)	0.0374 (8)	0.0381 (10)
z (O2)	0.7068 (9)	0.7071 (15)	0.7030 (11)	0.7041 (13)
B_{iso} (O2) (Å ²)	1.06 (10)	2.4 (2)	1.51 (13)	1.97 (17)
R_{wp} (%)	5.48	7.18	6.39	7.09
R_{p} (%)	4.07	5.28	4.81	5.18
R_{i} (%)	3.18	4.17	4.26	4.16
R_{f} (%)	2.51	3.81	4.35	4.56
Impurities:				
Bi ₂ O ₂ CO ₃	1.1 wt. %	0.8 wt. %	1.6 wt. %	1.6 wt. %
Cr ₂ O ₃	0.6 wt. %	0.9 wt. %	–	–

Crystal system: orthorhombic. Space group: *Pnma* (No. 62); $Z = 4$. Source: synchrotron powder X-ray diffraction ($\lambda = 0.420138$ Å). d -space range used in the refinements: 0.497–11.563 Å. Fractional coordinates: Bi: $4c$ ($x, 0.25, z$), Cr/Fe: $4b$ ($0, 0, 0.5$), O1: $4c$ ($x, 0.25, z$), and O2: $8d$ (x, y, z). Occupation factors, g , of the Bi and O sites are 1. The occupation factor of the Cr/Fe site is mixed based on the nominal compositions.

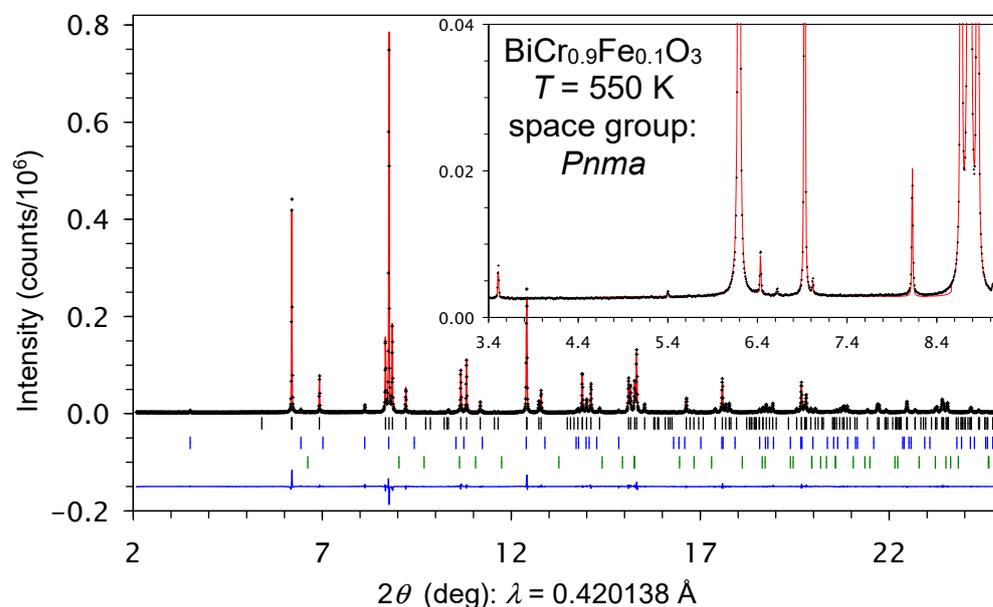


Figure 4. Fragments (between 2° and 25°) of experimental (black crosses), calculated (red line), and difference (blue line at the bottom) synchrotron powder X-ray diffraction patterns of $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$ at $T = 550$ K in the *Pnma* modification. The tick marks show possible Bragg reflection positions for the main phase (black) and $\text{Bi}_2\text{O}_2\text{CO}_3$ (blue) and Cr_2O_3 (green) impurities from top to bottom. The inset shows a magnified fragment.

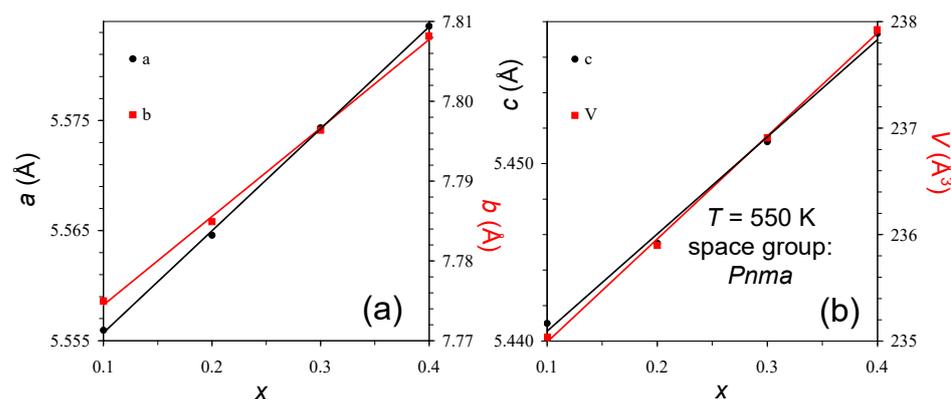


Figure 5. Compositional dependence of the lattice parameters of the $Pnma$ modifications of the $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions at $T = 550$ K. (a) The a and b lattice parameters, (b) the c lattice parameter and unit cell volume. Data on heating are used for $x = 0.1, 0.2,$ and 0.3 and on cooling—for $x = 0.4$.

BiFeO_3 has an incommensurate AFM structure, but BiFeO_3 has a well-defined commensurate crystal structure [6]. Incommensurate structural modulations were only observed in the rare-earth-doped samples, $\text{Bi}_{1-x}\text{R}_x\text{FeO}_3$, where R^{3+} is a rare-earth element, and at very limited compositional regions [40,68–73]. To the best of our knowledge, incommensurate structural modulations found in $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ samples, especially in the AP modifications, have never been observed before in the only-transition-metal-doped $\text{BiFe}_{1-x}\text{M}_x\text{O}_3$, where M is a transition metal element. Electron diffraction will be essential to understand incommensurate structural modulations of $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ samples.

Figure 6a shows the temperature dependence of the lattice parameters of $\text{BiCr}_{0.6}\text{Fe}_{0.4}\text{O}_3$ on heating and cooling (Table S2 provides numerical data). Drastic changes in the fundamental perovskite reflections were observed during the transition from the $Pbam$ modification to the $Pnma$ modification. The unit cell volume also drops by -1.4% at 550 K, where the two phases coexist, suggesting the first-order structural phase transition. It is interesting that the a lattice parameter of the $Pnma$ modification decreased with increasing temperature, suggesting anisotropic thermal expansion. All other parameters of the $Pnma$ modification and all parameters of the $Pbam$ modification increased with increasing temperature as expected. The fundamental lattice parameters of the HP and AP modifications were very close to each other at RT (Figure 6). However, the superstructure reflections were slightly different as discussed above. Therefore, we can discuss the different modifications, namely the HP and AP modifications, and the observation of the conversion polymorphism phenomenon [41] in the $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ system similar to the conversion polymorphism phenomenon in the $\text{BiFe}_{1-x}\text{Sc}_x\text{O}_3$ system [41] and the $\text{BiFe}_{1-x}\text{Mn}_x\text{O}_3$ system [36,49].

Because of the presence of some structural differences between the HP and AP modifications, we investigated the effects of the structural differences on the magnetic properties. Temperature-dependent magnetization curves of the HP and AP modifications are shown in Figures 7–10, and the results of M versus H measurements are shown in Figures 11–14 (Figures S2–S5 show magnified parts of Figures 11–14). The HP and AP modifications of $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$ showed very similar magnetic properties with one detectable magnetic transition at $T_N = 100$ K despite their different phase compositions (a different ratio of the $Pbam$ and $C2/c$ phases). Undoped BiCrO_3 has $T_N = 112$ K and shows a spin reorientation transition at $T_{N2} = 72$ K [23]. No signs of a spin reorientation transition were found in $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$. A noticeable weak FM moment appeared below $T_N = 100$ K at $H = 100$ Oe and $H = 10$ kOe (Figure 7). The transition with the weak FM moment with $T_N = 100$ K could correspond to the behavior of the $C2/c$ phase. The $Pbam$ phase has a much weaker FM moment (see properties of the $x = 0.3$ and 0.4 samples); therefore, its contribution to magnetic properties could be hidden in the $x = 0.1$ sample.

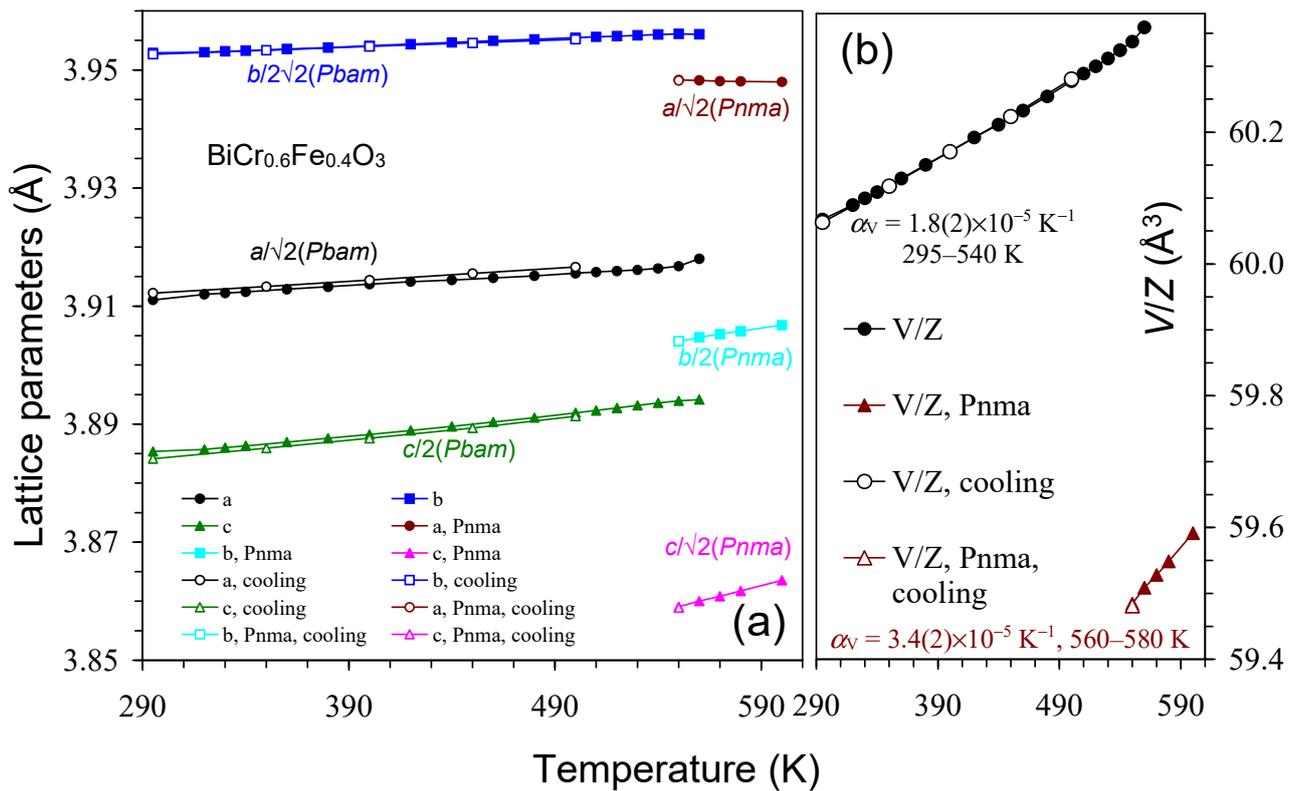


Figure 6. (a) The temperature dependence of the normalized lattice parameters of $\text{BiCr}_{0.6}\text{Fe}_{0.4}\text{O}_3$ on heating (full symbols) and cooling (empty symbols). (b) The temperature dependence of the normalized unit cell volumes (V/Z). α_V is the volumetric coefficient of thermal expansion.

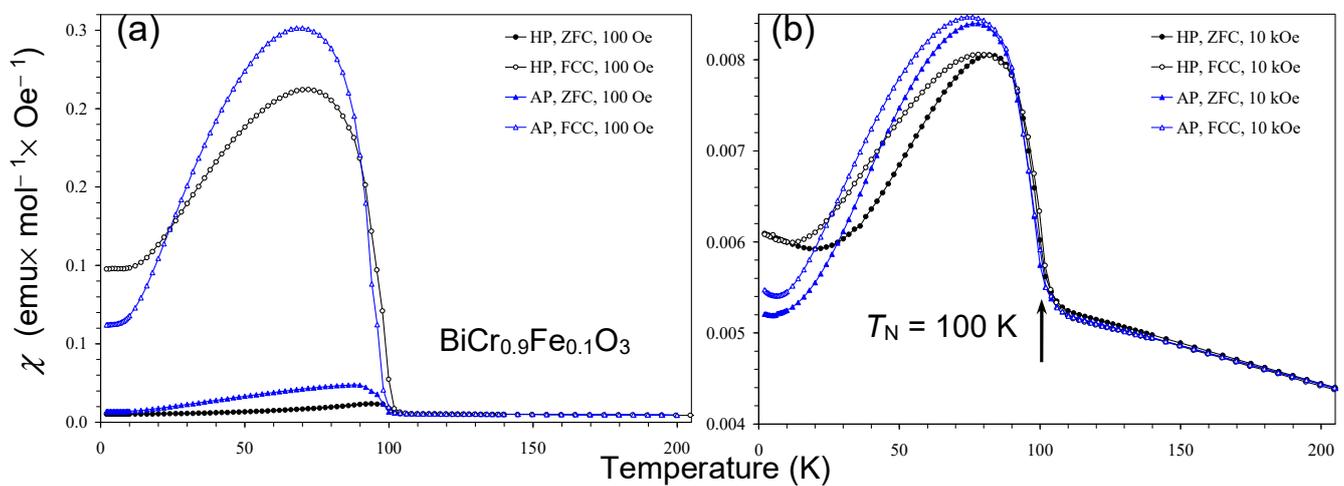


Figure 7. Magnetic properties of HP- $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$ (circles) and AP- $\text{BiCr}_{0.9}\text{Fe}_{0.1}\text{O}_3$ (triangles). Zero-field-cooled (ZFC: filled curves) and field-cooled on cooling (FCC: empty curves) curves are shown at (a) $H = 100 \text{ Oe}$ and (b) $H = 10 \text{ kOe}$.

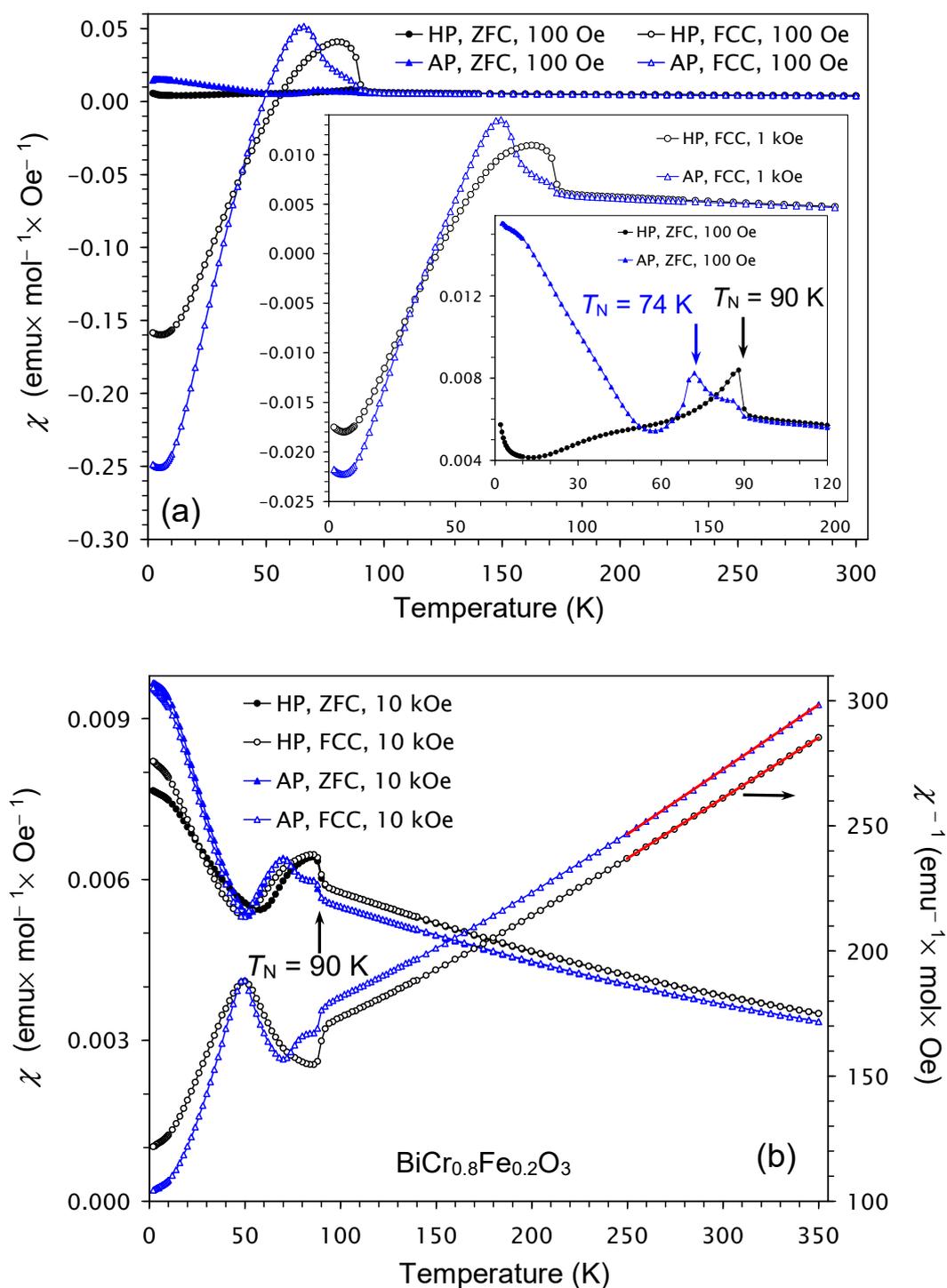


Figure 8. Magnetic properties of HP-BiCr_{0.8}Fe_{0.2}O₃ (circles) and AP-BiCr_{0.8}Fe_{0.2}O₃ (triangles). Zero-field-cooled (ZFC: filled curves) and field-cooled on cooling (FCC: empty curves) curves are shown at (a) $H = 100$ Oe and (b) $H = 10$ kOe. The main inset on panel (a) shows the FCC curves $H = 1$ kOe. The secondary inset shows the ZFC curves $H = 100$ Oe. The right-hand axis on panel (b) gives the inverse FCC χ^{-1} versus T curves at $H = 10$ kOe with Curie–Weiss fits (red lines).

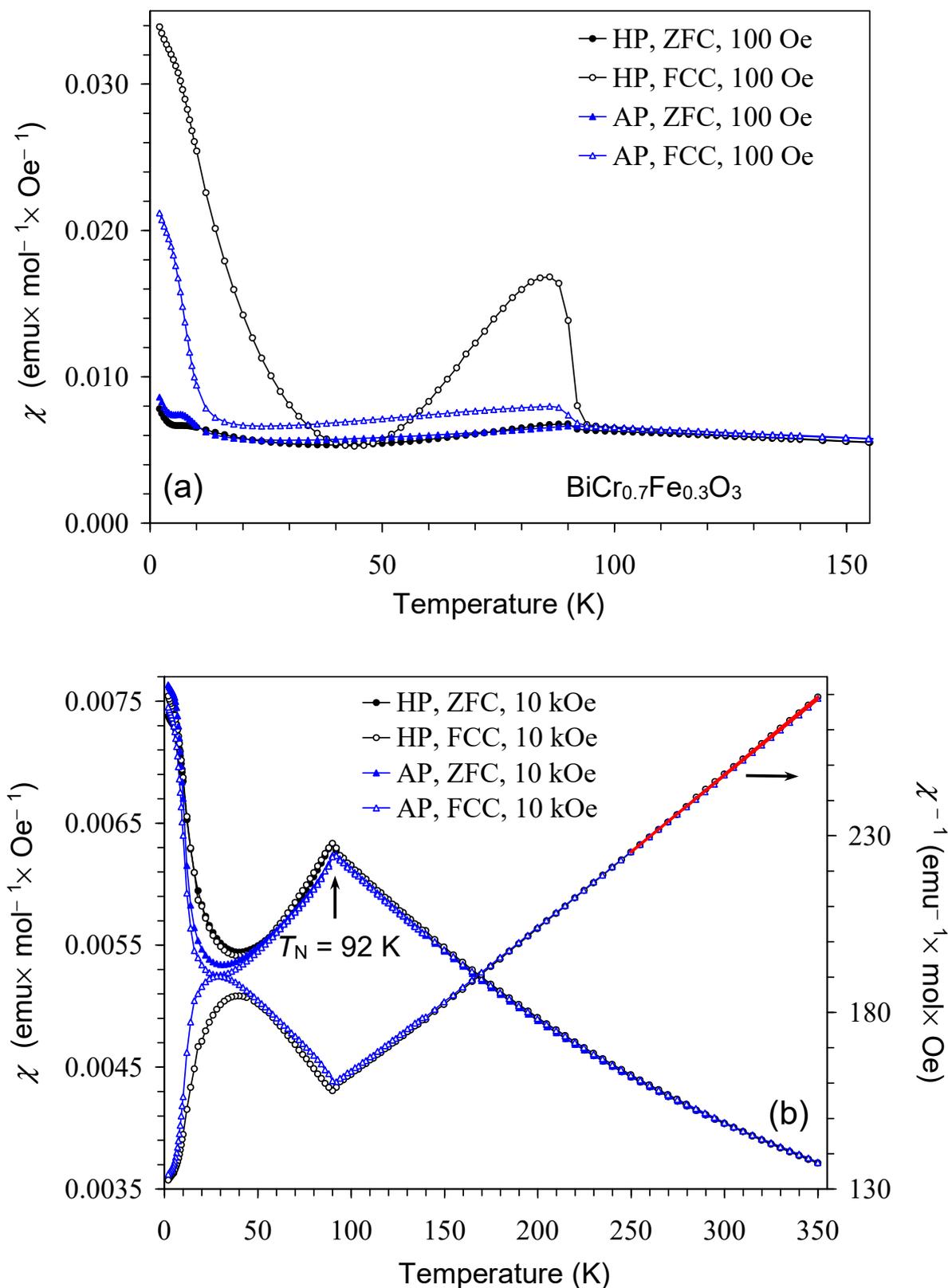


Figure 9. Magnetic properties of HP-BiCr_{0.7}Fe_{0.3}O₃ (circles) and AP-BiCr_{0.7}Fe_{0.3}O₃ (triangles). Zero-field-cooled (ZFC: filled curves) and field-cooled on cooling (FCC: empty curves) curves are shown at (a) $H = 100$ Oe and (b) $H = 10$ kOe. The right-hand axis on panel (b) gives the inverse FCC χ^{-1} versus T curves at $H = 10$ kOe with Curie-Weiss fits (red lines).

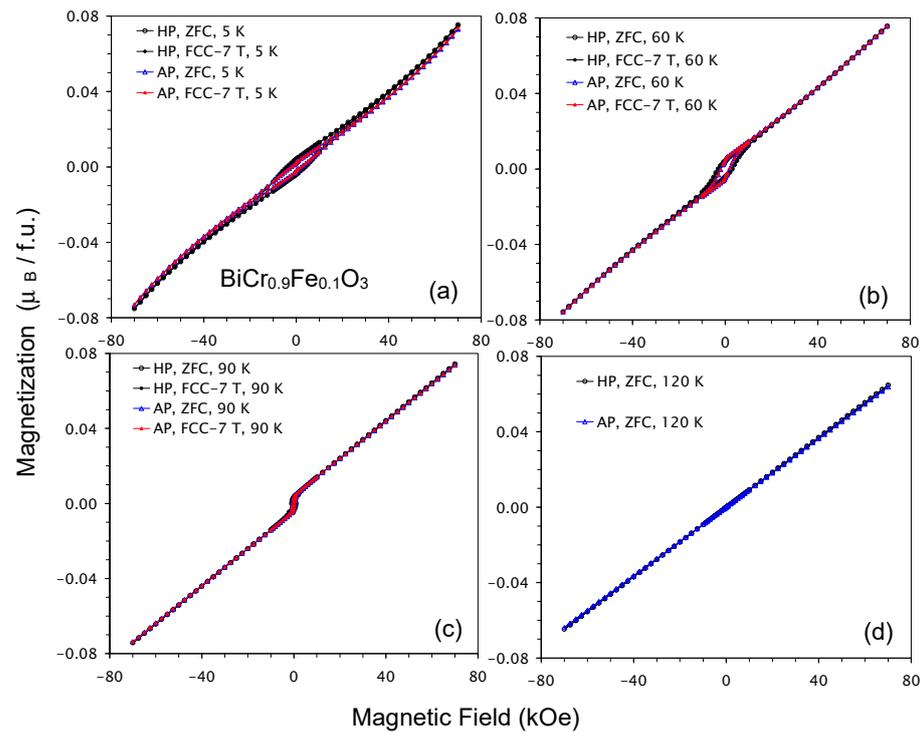


Figure 11. Comparison of magnetic properties of HP-BiCr_{0.9}Fe_{0.1}O₃ and AP-BiCr_{0.9}Fe_{0.1}O₃: M versus H curves at (a) $T = 5$ K, (b) $T = 60$ K, (c) $T = 90$ K, and (d) $T = 120$ K. M versus H curves are measured under the ZFC procedure, when samples are cooled from 300 K to measurement temperatures under $H = 0$ Oe, and under the FCC procedure (marked as FCC-7 T), when samples are cooled from 300 K to measurement temperatures under $H = 70$ kOe.

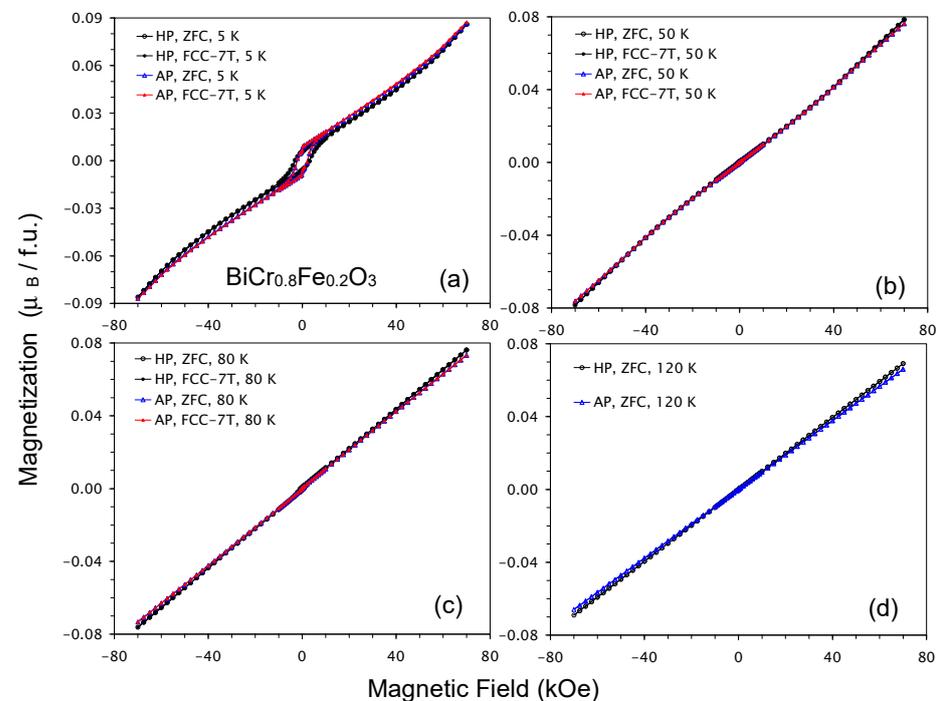


Figure 12. Comparison of magnetic properties of HP-BiCr_{0.8}Fe_{0.2}O₃ and AP-BiCr_{0.8}Fe_{0.2}O₃: M versus H curves at (a) $T = 5$ K, (b) $T = 50$ K, (c) $T = 80$ K, and (d) $T = 120$ K. M versus H curves were measured under the ZFC procedure, when samples are cooled from 300 K to measurement temperatures under $H = 0$ Oe, and under the FCC procedure (marked as FCC-7 T), when samples are cooled from 300 K to measurement temperatures under $H = 70$ kOe.

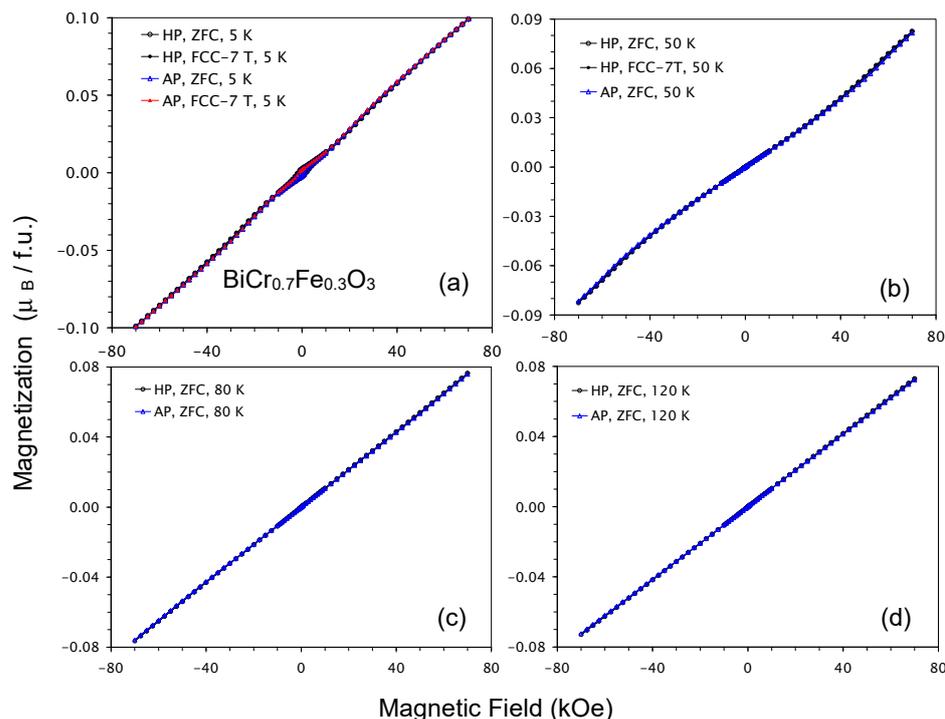


Figure 13. Comparison of magnetic properties of HP-BiCr_{0.7}Fe_{0.3}O₃ and AP-BiCr_{0.7}Fe_{0.3}O₃: M versus H curves at (a) $T = 5$ K, (b) $T = 50$ K, (c) $T = 80$ K, and (d) $T = 120$ K. M versus H curves are measured under the ZFC procedure, when samples are cooled from 300 K to measurement temperatures under $H = 0$ Oe, and under the FCC procedure (marked as FCC-7 T), when samples are cooled from 300 K to measurement temperatures under $H = 70$ kOe.

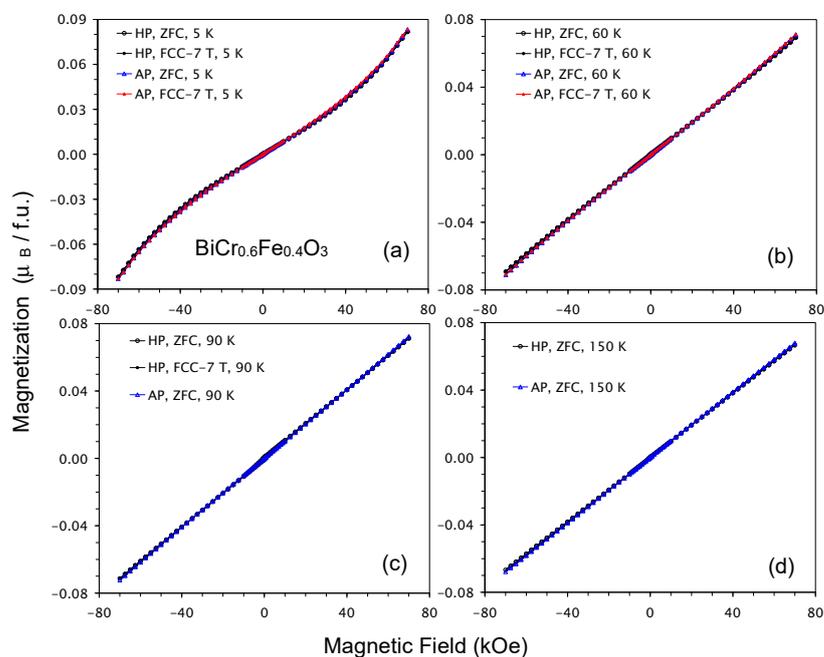


Figure 14. Comparison of magnetic properties of HP-BiCr_{0.6}Fe_{0.4}O₃ and AP-BiCr_{0.6}Fe_{0.4}O₃: M versus H curves at (a) $T = 5$ K, (b) $T = 60$ K, (c) $T = 90$ K, and (d) $T = 150$ K. M versus H curves are measured under the ZFC procedure, when samples are cooled from 300 K to measurement temperatures under $H = 0$ Oe, and under the FCC procedure (marked as FCC-7 T), when samples are cooled from 300 K to measurement temperatures under $H = 70$ kOe.

HP-BiCr_{0.8}Fe_{0.2}O₃ clearly showed one transition at $T_N = 90$ K (Figure 8) because of the relatively large fraction of the C2/ c phase. On the other hand, AP-BiCr_{0.8}Fe_{0.2}O₃ clearly

showed two transitions, the first one at $T_N = 90$ K from traces of the C2/c phase and the second one at $T_N = 74$ K from the majority of the *Pbam* phase. Traces of the C2/c phase could still be detected as this phase had a much larger FM moment. It is interesting that both HP- and AP-BiCr_{0.8}Fe_{0.2}O₃ showed a negative magnetization phenomenon [74] when the FCC curves were measured at $H = 100$ Oe and 1 kOe. Negative magnetization phenomena are often observed in different RCr_{1-x}Fe_xO₃ or RFe_{1-x}Cr_xO₃ solid solutions, where R³⁺ is a rare-earth element [75,76].

The magnetic properties of both HP- and AP-BiCr_{0.7}Fe_{0.3}O₃ were nearly identical when measured at high magnetic fields, such as $H = 10$ kOe (Figure 9b), with one AFM-like anomaly at $T_N = 92$ K. A very weak FM contribution (from spin canting) started emerging below about 20 K (Figures 9b and 13a). An extremely weak FM contribution also appeared just below $T_N = 92$ K at weak magnetic fields, such as $H = 100$ Oe (Figure 9a), and HP-BiCr_{0.7}Fe_{0.3}O₃ had a larger FM contribution in comparison to AP-BiCr_{0.7}Fe_{0.3}O₃. The weak FM contribution in HP- and AP-BiCr_{0.7}Fe_{0.3}O₃ at $H = 100$ Oe was about 10 times smaller than that of HP- and AP-BiCr_{0.9}Fe_{0.1}O₃.

Magnetic properties of both HP- and AP-BiCr_{0.6}Fe_{0.4}O₃ were nearly identical when measured at high magnetic fields, such as $H = 10$ kOe (Figure 10b), with one AFM-like anomaly at $T_N = 122$ K. No weak FM contributions appeared at lower temperatures in comparison with HP- and AP-BiCr_{0.7}Fe_{0.3}O₃ at $H = 10$ kOe. At a small magnetic field of $H = 100$ Oe, HP-BiCr_{0.6}Fe_{0.4}O₃ had a larger FM contribution in comparison to AP-BiCr_{0.6}Fe_{0.4}O₃. However, the weak FM contributions were extremely small, meaning that they could not be detected on the M versus H curves, which showed linear behavior in the vicinity of the origin between about -20 kOe and 20 kOe (Figure 14).

At high temperatures, inverse magnetic susceptibilities follow the Curie–Weiss law for all the samples (Figures 8b, 9b and 10b), and parameters of the Curie–Weiss fits are summarized in Table 2. The experimental effective magnetic moments were close to the expected, calculated ones. The Weiss temperature varied between -230 K and -280 K, resulting in a moderate frustration index of about 2.3 to 2.9. It is interesting that some M versus H curves showed noticeable upturn deviations from the linear behavior (at high magnetic fields), suggesting the presence of gradual field-induced transitions, for example, at $T = 5$ K for $x = 0.4$ (Figure 14a), $T = 50$ K for $x = 0.3$ (Figure 13b), $T = 5$ K and 50 K for $x = 0.2$ (Figure 12a,b), and $T = 5$ K for $x = 0.1$ (Figure 11a).

Table 2. Temperatures of structural transitions (T_{str}) and magnetic anomalies (T_N) and parameters of the Curie–Weiss fits and M versus H curves at $T = 5$ K for BiCr_{1-x}Fe_xO₃.

x	T_{str} (K)	T_N (K)	μ_{eff} ($\mu_B/f.u.$)	μ_{calc} ($\mu_B/f.u.$)	θ (K)	M_S ($\mu_B/f.u.$)
0.1 (HP)	450	100	3.995	4.123	-247	0.075
0.1 (AP)	450	98	4.061	4.123	-259	0.073
0.2 (HP)	480	90	4.068	4.359	-240	0.086
0.2 (AP)	478	90, 74	3.934	4.359	-227	0.087
0.3 (HP)	511	92	4.269	4.583	-264	0.099
0.3 (AP)	510	92	4.305	4.583	-273	0.099
0.4 (HP)	546	122	4.293	4.796	-279	0.083
0.4 (AP)	546	122	4.321	4.796	-279	0.083

The Curie–Weiss fits are performed between 250 and 350 K using the FCC χ^{-1} versus T data at 10 kOe. M_S is the magnetization value at $T = 5$ K and $H = 70$ kOe. μ_{calc} is calculated using $5.916\mu_B$ for Fe³⁺ and $3.873\mu_B$ for Cr³⁺. T_N values are determined from peaks on the 100 Oe FCC $d(\chi T)/dT$ versus T curves. T_{str} values are determined from peak positions on the heating DSC curves, where the anomalies on the first heating curves are assigned to the HP modification and the anomalies on the second and third heating curves are assigned to the AP modification. T_{str} corresponds to a transition to the GdFeO₃-type *Pnma* modification.

Figure 15 summarizes the temperature–composition phase diagram of the $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ system using the results of the current study and the literature data for the BiFeO_3 [6], BiCrO_3 [23,24,31], and $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ systems with high Fe content [63]. On the scale of Figure 15, T_N remains nearly the same for $0.0 \leq x \leq 0.4$ and then monotonically increases from $x = 0.4$ to $x = 1$. T_{str} increases gradually for $0.0 \leq x \leq 0.4$ and then more rapidly from $x = 0.4$ to $x = 1$.

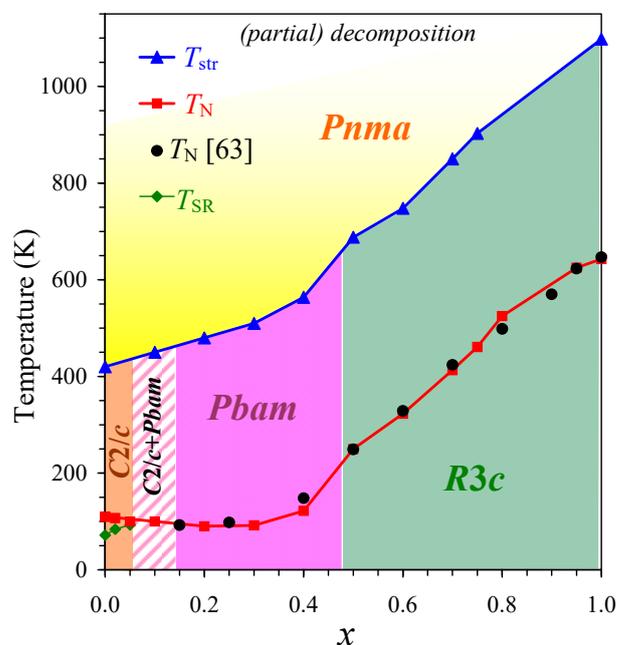


Figure 15. The temperature–composition phase diagram of the $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ system. T_{str} is the temperature of the structural transition to the $Pnma$ modification. T_N is the Néel temperature. T_{SR} is the temperature of the spin reorientation transition.

3. Materials and Methods

The as-synthesized HP modifications of $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions with $x = 0.1, 0.2, 0.3,$ and 0.4 were prepared from stoichiometric mixtures of Bi_2O_3 (Rare Metallic Co., Tokyo, Japan, 99.9999%), Fe_2O_3 (Rare Metallic Co., Tokyo, Japan, 99.9999%), and Cr_2O_3 (Rare Metallic Co., Tokyo, Japan, 99.9%). The synthesis was performed at about 6 GPa and about 1600 K for 1 h in sealed Pt capsules using a belt-type HP instrument. After annealing at 1600 K, the samples were cooled down to room temperature by turning off the heating current, and the pressure was slowly released. The AP modifications of $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ solid solutions with $x = 0.1, 0.2, 0.3,$ and 0.4 were obtained by heating HP- $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ in air at AP at 623 K for 10 min (with a heating/cooling rate of 10 K/min). We note that at $x \geq 0.5$, a different modification with space group $R3c$ is formed [62] (that is confirmed by us); therefore, compositions with $x \geq 0.5$ were not included in the present work.

X-ray powder diffraction (XRPD) data were collected at room temperature on a MiniFlex600 diffractometer (Rigaku, Tokyo, Japan) using $\text{CuK}\alpha$ radiation (2θ range of $8\text{--}100^\circ$, a step width of 0.02° , and scan speed of $2^\circ/\text{min}$). Synchrotron XRPD data of HP- $\text{BiCr}_{1-x}\text{Fe}_x\text{O}_3$ were collected at 297 K, upon heating to 550 K ($x = 0.1$ and 0.2) or to 600 K ($x = 0.3$ and 0.4), and then on cooling to 297 K using the beamline BL02B2 [77,78] of SPring-8, Japan. Intensity data were taken between 2.082° and 78.216° at a 0.006° interval in 2θ using a wavelength of $\lambda = 0.420138 \text{ \AA}$; however, data up to 50° were used in the Rietveld analysis as no detectable experimental reflections were observed above 50° . The measurement time was 300 s at 297 K and 550 K (or 600 K) and 60 s at other temperatures. The samples were placed into open Lindemann glass capillary tubes (with an inner diameter of 0.1 mm),

which were rotated during the measurements. The Rietveld analysis of all XRPD data was performed using the *RIETAN-2000* program [79]. The reported weight fractions of all the phases were calculated by the *RIETAN-2000* program [79] from the refined scale factors.

Magnetic measurements were performed on a SQUID magnetometer (Quantum Design MPMS3, San Diego, CA, USA) between 2 and 350 K in different applied fields using zero-field-cooled (ZFC) and field-cooled on cooling (FCC) procedures. Isothermal magnetization measurements, M versus H , were performed from 70 kOe to -70 kOe and from -70 kOe to 70 kOe using both ZFC and FCC procedures. In the ZFC procedure for M versus H measurements, the samples were cooled from 300 K to a measurement temperature under a zero magnetic field; in the FCC procedure, the samples were cooled from 300 K to a measurement temperature under a magnetic field of 70 kOe. The ZFC and FCC procedures for M versus H measurements were used to check the presence or absence of the exchange bias effect; no detectable difference was observed on M versus H curves measured in the ZFC and FCC procedures, suggesting the absence of the exchange bias effect.

Pieces of pellets were used in magnetic measurements. A pellet of each HP-BiCr_{1-x}Fe_xO₃ was first used to obtain magnetic properties of the HP modification; the same pellet was then transformed to AP-BiCr_{1-x}Fe_xO₃ as described above, and it was used in magnetic measurements to obtain magnetic properties of the AP modification.

Differential scanning calorimetry (DSC) curves of powder samples of HP-BiCr_{1-x}Fe_xO₃ were recorded on a Mettler Toledo DSC1 STAR^e system between 297 K and maximum 573 K in open Al capsules with a heating/cooling rate of 10 K/min. Three DSC runs were performed to check the reproducibility.

4. Conclusions

In conclusion, two modifications of the BiCr_{1-x}Fe_xO₃ perovskite solid solutions were prepared. The HP modifications (as-synthesized) were prepared by a high-pressure high-temperature method at 6 GPa. The AP modifications were obtained using a “conversion polymorphism” strategy after heating at AP above structural phase transition temperatures and cooling to room temperature. The HP and AP modifications had subtle structural differences and showed incommensurate structural modulations. Structural phase transitions to the *Pnma* modification were observed in all the samples and investigated in details. Subtle differences in magnetic properties of the HP and AP modifications were investigated and reported. In particular, the $x = 0.2$ samples demonstrated negative magnetization phenomena.

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/inorganics13030091/s1>, Figure S1: Fragments (between 4.4° and 7.4°) of experimental and calculated synchrotron powder X-ray diffraction patterns of HP-BiCr_{0.6}Fe_{0.4}O₃ and AP-BiCr_{0.6}Fe_{0.4}O₃ at $T = 297$ K; Figure S2: A magnified part of Figure 11; Figure S3: A magnified part of Figure 12; Figure S4: A magnified part of Figure 13; Figure S5: A magnified part of Figure 14; Table S1: Numerical data used to plot Figure 5; Table S2: Numerical data used to plot Figure 6.

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