

# Surface Passivation of Sn-Based Wide Band Gap Perovskite Solar Cells Using Functional Molecules

Dhruba B. Khadka<sup>1</sup>, Masatoshi Yanagida<sup>1</sup>, Roji Sahara<sup>2</sup>, and Yasuhiro Shirai<sup>1</sup>

<sup>1</sup>Photovoltaics Materials Group, Global Research Center for Environment and Energy based on Nanomaterials Science (GREEN), National Institute for Materials Science (NIMS), 1-1 Namiki, Tsukuba, Ibaraki 305-0044, Japan

<sup>2</sup>Computational Structural Materials Group, Research Center for Structural Materials, National Institute for Materials Science (NIMS), 1-2-1, Sengen, Tsukuba, Ibaraki, 305-0047 Japan

**Abstract** — Surface passivation with multifunctional molecules is a powerful approach to enhancing the performance and stability of perovskite solar cells (PSCs). This study reports the fabrication of wide-bandgap tin-based perovskite solar cells (WB-Sn-PSCs) with a bandgap of 1.68 eV, achieving a power conversion efficiency of 11.14%. Molecular surface passivation using 4-Fluorobenzohydrazide (F-BHZ) significantly improves device performance and stability by engineering both the surface and bulk properties of the WB-Sn-HP film. F-BHZ treatment strengthens electrostatic potential and molecular interactions with functional groups, mitigating surface defects and suppressing Sn<sup>2+</sup> oxidation, as confirmed by theoretical calculations. This work highlights the potential of advanced chemical engineering strategies to optimize non-toxic, lead-free perovskite solar cells, advancing their competitiveness as environmentally friendly photovoltaic solutions.

## I. INTRODUCTION

Tandem solar cells (Silicon/perovskite) have progressed by combining narrow bandgap halide perovskites ( $E_g \sim 1-1.3$  eV) as bottom cells with wide bandgap Pb-halide perovskites ( $E_g > 1.6$  eV) as top cells. [1] However, the toxicity of the lead in the halide perovskite (HP) has been an impugning factor for broader acceptance. Therefore, wide band gap tin perovskites (WB-Sn-HPs) could be one of the best choices for Pb-free Si/perovskite tandem applications.

Few works have been documented for the fabrication of WB-Sn-PSCs using additive engineering.[2] A report by Hu et al. has used an antioxidant additive such as aromatic carboxylic acid molecules in WB-Sn-HP ( $\text{FA}_{0.75}\text{MA}_{0.25}\text{SnI}_2\text{Br}$ ) resulting in a PCE of 10.35% with improved device stability.[3] Similarly, Khang and his colleagues obtained a certified record PCE of 11.70% of WB-Sn-PSCs by material engineering using 4-phenylthiosemicarbazide as a functional additive.[4] Cao et al. have reported WB-Sn-PSC using bottom passivation with potassium thiocyanate demonstrating a decent PCE of 11.17% by improving crystallization and interface engineering.[5] Although the knowledge of Pb-PSCs could be utilized in Sn-PSCs,[6], [7], [8], [9] there is still a significant gap in the progress of Sn-based PSCs compared to Pb-PSCs.[10], [11]

In this report, we documented the fabrication of WB-Sn-PSCs (FA, MA, PEA)SnI<sub>2</sub>Br) of  $E_g \sim 1.68$  combined with surface passivation using 4-Fluoro-benzohydrazide (F-

BHZ)multifunctional molecule. We achieved among the best PCE of 11.14% with inverted device configuration ITO/PEDOT:PSS/WB-Sn-HP/ICBA/BCP/Ag. The effect of surface passivation of materials properties and device physics have been explored to get insights into the underlying mechanism for the betterment of device performance.

## II. EXPERIMENTAL DETAILS

### A. Device fabrication

For the fabrication of wide-band gap tin perovskite of composition  $\text{-FA}_{0.75}\text{MA}_{0.10}\text{PEA}_{0.15}\text{SnI}_2\text{Br}$ : we prepared 0.85 M of the precursor solution by dissolving FABr (0.75 M), MABr (0.1 M), PEABr (0.15 M), SnI<sub>2</sub> (1 M), SnF<sub>2</sub> (0.1 M), Sn powder (5 mg) in the mixture of solvent (dimethylformamide and dimethyl sulfoxide (4:1)) for 2 hours.[12] PEDOT:PSS was used for the hole transport layer.[11], [13], [14] For WB-Sn-HP film deposition, the precursor was spin-coated at 6000 rpm-90 s followed by dripping 0.7 ml of CB at 64<sup>th</sup> s. Then, these as-grown films were simply placed on a hot plate at 70°C for 5min. For surface passivation, 4-Fluoro-benzo hydrazide (F-BHZ) precursor solution was spin-coated onto the WB-Sn-HP film at 5000 rpm-50 s and annealed at 70°C-5 min. Then, we deposited ICBA (18 mg/ml-CB) and BCP (2 mg/ml-IPA) by spin coating. Finally, Ag was thermally evaporated. The details have been discussed in our earlier reports.[12]

### B. Materials and device characterizations

XRD data were measured using Rigaku Smart Lab, CuK $\alpha$  radiation,  $\lambda=1.5405\text{\AA}$ . Scanning electron microscopy (SEM) images were collected by a high-resolution scanning electron microscope (SEM) at 5 kV accelerating voltage (Hitachi, S-4800). The absorption and photoluminescence (PL) spectra were measured using a micro-PL spectrometer (HORIBA, LabRamHR-PL NF(UV-NIR)). The current density–voltage (J-V) curves were measured under 1 sun with an AM1.5G spectral filter coupled with an MPPT system (Systemhouse Sunrise Corp.). Capacitance spectra (C–f) were collected using an LCR meter (IM3536, Hioki) in the dark.[15], [16]

### III. RESULTS AND DISCUSSION

To study the effect of surface passivation on WB-Sn-perovskite film with F-BHZ molecule, WB-Sn-PSC, and its device and materials properties have been investigated.

The molecules used for surface passivation are given in Fig. 1a. The F-BHZ molecule possesses higher electron density in the vicinity of hydrazide and carbonyl functionalities showing strong molecular interaction properties. Figure 1b depicts the surface treatment method of WB-Sn-HP film. The molecular interaction of the F-BHZ molecule with Sn-perovskite film has been displayed showing the interaction of Sn with oxygen in the carbonyl group as confirmed by theoretical calculations.[12]

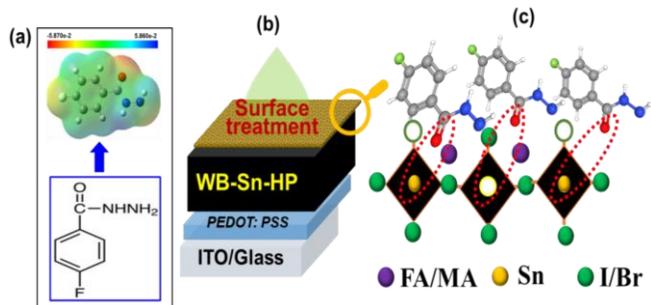


Figure 1. Electrostatic surface potential of 4-Fluorobenzohydrazide (F-BHZ) (a). Surface passivation of WB-HP with F-BHZ. (c) Schematic of interfacial interaction.

To study the effect of the surface passivation effect on the device, we fabricated WB-Sn-PSC using inverted device configuration of ITO/PEDOT:PSS/WB-Sn-HP/ST(F-BHZ)/ICBA/BCP/Ag as depicted in Figure 2a. Figure 2b shows the current density -voltage characteristics of WB-Sn-PSCs of control and optimal surface treatment (1 mg/ml- F-BHZ). Inset table 1 displays the figures of merit of respective devices. The control device demonstrated a *PCE* of 7.96% with device parameters; short circuit current density ( $J_{SC}$ )  $\sim$  12.58  $\text{mAcm}^{-2}$ , open circuit voltage ( $V_{OC}$ )  $\sim$  0.914 V, fill factor (*FF*)  $\sim$  0.692. WB-Sn-PSC with F-BHZ ( $\leq$  1 mg/ml) treated improved *PCE* to  $\sim$ 11.14% with a significant increase in  $V_{OC}$   $\sim$  1.024 V and *FF*  $\sim$  75.7%. This device result is among the best reports on WB-Sn-PSCs.[12] The material growth properties and device characteristics suggest that the compact film morphology, high crystalline quality, modulated surface chemistry, interface energy, and defect passivation.[12], [17] Figure 2c presents the effect of the concentration of F-BHZ on device performance and statistical scenario of device parameters.

Figure 3 a-c shows the surface morphology of control and surface treated WB-Sn-HP films. The WB-HP film treated with F-BHZ displays a well-covered surface with more compact and smoother grains. However, WB-Sn-HP with F-BHZ (Figure 3c) with 2mg/ml grows with unevenly distributed small crystallites,

indicating deleterious film quality. This observation aligns with the results of lower efficiency of WB-Sn-PSC treated with a higher concentration of F-BHZ ( $>1.5$  mg/ml). This decline is ascribed to the deterioration of film quality stemming from uneven film morphology induced during post-annealing with more concentrated passivated molecules.

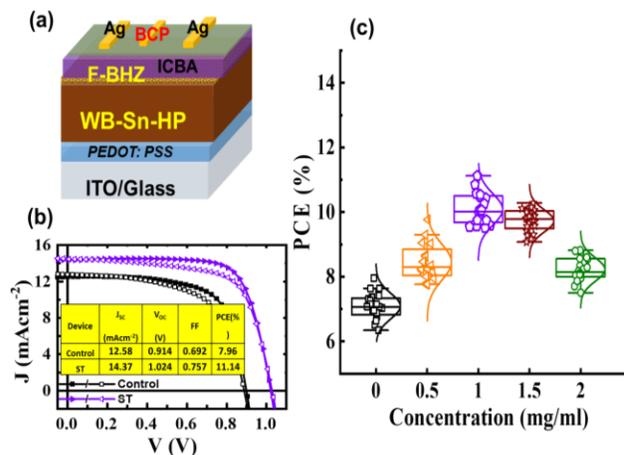


Figure 2. Effect of surface treatment: SEM image of surface treatment- (a) control, (b) PEDAI, (c) PZDI. XRD patterns (d, e) (#-2D phase,  $\delta$ -non-photoactive perovskite phase and (f) PL spectra of HP film: control, PEDAI, and PZDI treatment.

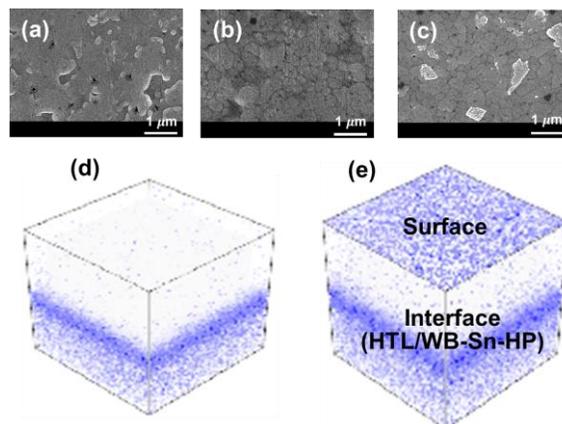


Figure 3. SEM images of WB-Sn-HP films with 4F-BHZ treatment with varying concentrations; (a) control (0 mg), (b) F-BHZ (1 mg/ml), and (c) F-BHZ (2 mg/ml), yellow circles indicate uneven crystallites growth. F-BHZ distribution 3D image reconstructed from the ToF-SIMS depth profiles; d) control and e) F-BHZ treated film.

Figure 3 d,e shows a set of 3D image profiles of F-BHZ molecules using the time-of-flight secondary ion mass spectrometry (ToF-SIMS). The F-BHZ molecule uniformly covers up the surface of the WB-Sn-HP film with a slight diffusion to the bulk through the grain boundary. This supports the suppression of Sn oxidation and defect chemistries in surface-treated WB-Sn-HP film.[12], [17]

Results of this work corroborate the improvement in device performance of WB-PSC with surface passivation of perovskite film using F-BHZ functional molecules.

#### IV. SUMMARY AND CONCLUSIONS

This report presents a surface method on Sn-based wide band gap perovskite film for the betterment of device efficiency and stability. WB-PSC (1.68 eV) with F-BHZ-treated Sn-perovskite film demonstrated enhanced device efficiency as high as 11.14% compared to the control device of PCE ~ 7.96 %. We found the modulation of the surface chemistry, interface energy by functional molecular bonding, and control of the Sn<sup>2+</sup> oxidation with uniform distribution of the F-BHZ molecule on WB-Sn-HP film analysis. This work underscores the way for developing efficient and stable WB-Sn-PSC by modulating the defect chemistry with chemical engineering.

#### ACKNOWLEDGMENT

This work was supported by The Hitachi Global Foundation, Kurata grants #1572.

#### REFERENCES

- [1] R. Lin *et al.*, ‘All-perovskite tandem solar cells with 3D/3D bilayer perovskite heterojunction’, *Nature*, vol. 620, no. 7976, pp. 994–1000, Aug. 2023, doi: 10.1038/s41586-023-06278-z.
- [2] W. J. Jang, H. W. Jang, and S. Y. Kim, ‘Recent Advances in Wide Bandgap Perovskite Solar Cells: Focus on Lead-Free Materials for Tandem Structures’, *Small Methods*, vol. 8, no. 2, p. 2300207, Feb. 2024, doi: 10.1002/smt.202300207.
- [3] F. Hu *et al.*, ‘A vertical antioxidant strategy for high performance wide band gap tin perovskite photovoltaics’, *J Mater Chem A Mater*, vol. 11, no. 9, pp. 4579–4586, 2023, doi: 10.1039/D2TA09363D.
- [4] P. Pandey *et al.*, ‘4-Phenylthiosemicarbazide Molecular Additive Engineering for Wide-Bandgap Sn Halide Perovskite Solar Cells with a Record Efficiency Over 12.2%’, *Adv Energy Mater*, vol. 14, no. 25, p. 2401188, Jul. 2024, doi: 10.1002/aenm.202401188.
- [5] J.-J. Cao *et al.*, ‘Multifunctional potassium thiocyanate interlayer for eco-friendly tin perovskite indoor and outdoor photovoltaics’, *Chemical Engineering Journal*, vol. 433, p. 133832, Apr. 2022, doi: 10.1016/j.cej.2021.133832.
- [6] D. B. Khadka, Y. Shirai, M. Yanagida, and K. Miyano, ‘Ammoniated aqueous precursor ink processed copper iodide as hole transport layer for inverted planar perovskite solar cells’, *Solar Energy Materials and Solar Cells*, vol. 210, p. 110486, 2020, doi: <https://doi.org/10.1016/j.solmat.2020.110486>.
- [7] D. B. Khadka, Y. Shirai, M. Yanagida, and K. Miyano, ‘Unraveling the Impacts Induced by Organic and Inorganic Hole Transport Layers in Inverted Halide Perovskite Solar Cells’, *ACS Appl Mater Interfaces*, vol. 11, no. 7, pp. 7055–7065, Feb. 2019, doi: 10.1021/acsami.8b20924.
- [8] D. B. Khadka, Y. Shirai, M. Yanagida, T. Noda, and K. Miyano, ‘Tailoring the Open-Circuit Voltage Deficit of Wide-Band-Gap Perovskite Solar Cells Using Alkyl Chain-Substituted Fullerene Derivatives’, *ACS Appl Mater Interfaces*, vol. 10, no. 26, pp. 22074–22082, 2018, doi: 10.1021/acsami.8b04439.
- [9] D. B. Khadka, Y. Shirai, M. Yanagida, J. W. Ryan, and K. Miyano, ‘Exploring the effects of interfacial carrier transport layers on device performance and optoelectronic properties of planar perovskite solar cells’, *J Mater Chem C Mater*, vol. 5, no. 34, pp. 8819–8827, 2017, doi: 10.1039/C7TC02822A.
- [10] D. B. Khadka *et al.*, ‘Advancing Efficiency and Stability of Lead, Tin, and Lead/Tin Perovskite Solar Cells: Strategies and Perspectives’, *Solar RRL*, vol. 7, no. 21, p. 2300535, Nov. 2023, doi: 10.1002/solr.202300535.
- [11] D. B. Khadka, Y. Shirai, M. Yanagida, and K. Miyano, ‘Passivation of the Recombination Activities with Rubidium incorporation for Efficient and Stable Sn- HaP Solar Cells’, in *2020 47th IEEE Photovoltaic Specialists Conference (PVSC)*, IEEE, Jun. 2020, pp. 0113–0116. doi: 10.1109/PVSC45281.2020.9300783.
- [12] D. B. Khadka, Y. Shirai, R. Sahara, M. Yanagida, and K. Miyano, ‘Ameliorating Defects in Wide Bandgap Tin Perovskite Solar Cells Using Fluorinated Solvent and Hydrazide’, *Small*, pp. 2410048–2410048, Dec. 2024, doi: 10.1002/sml.202410048.
- [13] D. B. Khadka, Y. Shirai, M. Yanagida, and K. Miyano, ‘Pseudohalide Functional Additives in Tin Halide Perovskite for Efficient and Stable Pb-Free Perovskite Solar Cells’, *ACS Appl Energy Mater*, vol. 4, no. 11, pp. 12819–12826, Nov. 2021, doi: 10.1021/acsae.1c02496.
- [14] D. B. Khadka, Y. Shirai, M. Yanagida, T. Masuda, and K. Miyano, ‘Enhancement in efficiency and optoelectronic quality of perovskite thin films annealed in MAI vapor’, *Sustain Energy Fuels*, vol. 1, no. 4, pp. 755–766, 2017, doi: 10.1039/C7SE00033B.
- [15] D. B. Khadka *et al.*, ‘Defect passivation in methylammonium/bromine free inverted perovskite solar cells using charge-modulated molecular bonding’, *Nat Commun*, vol. 15, no. 1, p. 882, Jan. 2024, doi: 10.1038/s41467-024-45228-9.
- [16] D. B. Khadka, Y. Shirai, M. Yanagida, T. Tadano, and K. Miyano, ‘Interfacial Embedding for High-Efficiency and Stable Methylammonium-Free Perovskite Solar Cells with Fluoroarene Hydrazine’, *Adv Energy Mater*, vol. 12, no. 38, p. 2202029, Oct. 2022, doi: 10.1002/aenm.202202029.
- [17] D. B. Khadka, Y. Shirai, M. Yanagida, T. Tadano, and K. Miyano, ‘Alleviating Defect and Oxidation in Tin Perovskite Solar Cells Using a Bidentate Ligand’, *Chemistry of Materials*, vol. 35, no. 11, pp. 4250–4258, Jun. 2023, doi: 10.1021/acs.chemmater.3c00243.