

Synthesis and high-temperature properties of *medium-entropy* (Ti,Ta,Zr,Nb)C using the spark plasma consolidation of carbide powders

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Abstract

In this study, we explore a simple and effective method for producing a quaternary solid-solution of carbides, a medium-entropy (Ti_{1/4}Ta_{1/4}Zr_{1/4}Nb_{1/4})C. Using commercially-available carbide powders with an equimolar ratio and performing spark plasma consolidation at 1927 °C and 1977 °C, we demonstrated that the phase formation and high-temperature flexural strength can be controlled using the processing conditions. The flexural strength and fracture toughness at room temperature were reached an average of 560 MPa and 3.2 MPa m^{1/2}, respectively. The high-temperature performance of these ceramics was analyzed and compared with the available data for the group IV and V transition metal carbide monoliths.

Keywords: transition metal carbides; high-entropy ceramics; medium-entropy ceramics; flexural strength; high-temperature materials.

1 Introduction

Recent interest in devices capable of withstanding high temperatures concerns various industries from aerospace with advanced thermal protection materials to energy-related where an increase in energy efficiency is derived from an increase in the working temperatures. Energy efficiency is now being prioritized as new materials are being developed for green energy sources such as solar energy [1–3]. For these purposes, a class of ultra-high temperature ceramics (UHTC) is being used as the main backbone for future high-temperature material development. This is not surprising since monolithic additive-free carbides and diborides of

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1 the transition metals of the IV and V groups possess a unique combination of high melting
2 points, high elastic moduli, and high-temperature strength [4–7].

3 To further increase the potential application of these compounds, further development is
4 required as room and high-temperature performance of these monoliths is being constantly
5 improved by the composite approach [8,9] or creating solid solutions [10]. The latter approach
6 was spurred on by the development of multi-metal solid-solution compounds, analogous to the
7 recent advances in metallurgy as these compounds were defined as high-entropy ceramics [11–
8 13], while the equimolar solid-solution with four principle metal elements can be defined as
9 medium-entropy ceramics.

10 Various synthesis methods for these carbides, borides, and silicides have been developed in the
11 last two years [11–14]. However, because of the slow atomic mobility of transition metal
12 carbides of the IV and V groups, their consolidation or combined consolidation/synthesis
13 [11,13–16] requires a temperature range typical for the classical UHTCs (i.e., >1800 °C).
14 Without careful control of the consolidation conditions, such high-temperatures cause grain
15 growth and hence further embrittlement of these ceramics.

16 Despite reports of the high-hardness and high Young’s modulus of high-entropy carbides,
17 studies related to strength or toughness at room or high-temperature are somewhat sparse
18 [14,16–19].

19 A recent report of the ternary high-entropy carbide prototype $(\text{Ta}_{1/3}\text{Zr}_{1/3}\text{Nb}_{1/3})\text{C}$ [14] showed
20 an interesting high-temperature flexural behavior, suggesting that solid-solution strengthening
21 might lead to further improvement of the high-temperature properties. In this regard, further
22 studies, which investigate the addition of a 4th or 5th metal atom to the (Ta,Zr,Nb)C system may
23 clarify the potential in the development of carbides in particular, and the UHTC in general.

24 In the present study to obtain a quaternary medium-entropy carbide, the TiC, TaC, ZrC, and
25 NbC were combined at the mixing stage. The formation of a solid-solution between carbides
26 was observed during the spark plasma consolidation at 1927 °C and 1977 °C. The present
27 investigation will examine the possibility of using these ceramics as a structural material at
28 elevated temperatures (i.e., 1600–1800 °C). In particular, the effect of the processing conditions
29 on the lattice parameters and high-temperature flexural strength were the main focus of this
30 study.

31 **2 Materials and Methods**

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Commercially available TiC, TaC, ZrC, and NbC (Wako Pure Chemicals, Osaka, Japan) powders were used as the starting materials. The raw powders were mixed in equimolar ratios TiC:TaC:ZrC:NbC 1:1:1:1. Powder mixtures were prepared by wet-chemical mixing in alcohol with low-temperature drying (at ~100 °C) to remove moisture. The resultant powder was screened using a 60-mesh screen. For simplicity, a quaternary equimolar mixture was denoted as TTZN. Based on the previous results for (Ta,Zr,Nb)C [14], a dwell time of 10 and 40 min at 1977 °C and 1927 °C was used: TTZN-10 and TTZN-40, respectively. These conditions were used in order to obtain ceramics with similar grain sizes.

The spark plasma sintering (SPS) experiments were conducted using ‘Dr. Sinter’ 1050 (Sumitomo, Japan) unit using a 30-mm die. The schedule for the TTZN carbide specimens prepared in this study had four major steps: (1) heating to 700 °C in four minutes following a five-minute dwell, (2) heating to 1500 °C in 10 minutes with a five-minute dwell; (3) five-minute ramp to 1927 °C or 1977 °C, with a dwell of 40 or 10 minutes. The last step included cooling down to 600 °C in 15 minutes. Steps (1) and (2) were performed in a vacuum; during the dwell at 1400 °C, the SPS chamber was backfilled with argon. At the end of the dwell at stage (2), the pressure was increased from 8 to 12 kN, while reaching 1927 °C the pressure was increased from 12 to 32 kN. The pressure of 32 kN was maintained during the consolidation and cooling stages. Argon gas at the flow rate of 2 L/min was used.

An X-ray diffraction (XRD) analysis (D8 Advance, Bruker, Karlsruhe, Germany) was performed on the polished surfaces of the bars before the flexural tests using the Cu-K α radiation. The intensity data were collected over the 2 θ range of 20°– 130°, in steps of 0.02– 0.05°, using a sampling time of 10 s for each step. In some instances, a sampling time of 40 s was used in the attempt to refine peaks with 2 thetas higher than 100°. The software used for the refinement was TOPAS (TOPAS Ver. 4.0, Bruker AXS, Germany). Instrumental broadening was determined using a NIST 660b LaB₆ standard is run under the same conditions for each carbide sample [20]. The computation of the lattice parameters for the NaCl-type carbides was performed using TOPAS or using refinement code developed by Lutterotti et al. [21]. Lattice parameters of the carbides were determined with an accuracy of 0.0001 Å.

The structural characteristics of the TTZN ceramics and were studied using scanning electron microscopy (SEM, JCM-6000, JEOL) with secondary (SE) and backscattered electrons (BSE mode).

The three-point flexural strength was determined using rectangular blocks (2×2×25 mm) using strength testing equipment were previously described in detail [22,23]. A span of 16 mm was used. Measurements were performed with a loading speed of 0.5 mm/min. Four to six samples

1 were tested at room temperature originated from ceramic tiles that were free of macroscopic
2 cracks. In some instances, specimens were tested using the four-point method (a 20–10 mm
3 configuration). The standard deviation was taken as the measurement accuracy. Tests at
4 elevated temperatures were performed in argon. The heating procedure for the high-
5 temperature flexure tests at 1600 °C and 1800 °C are described in detail elsewhere [23].
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8 We evaluated the elastic modulus (E_f) for the tests at room temperature from the linear portion
9 of the load-displacement curve using the procedure described in ASTM E111–04.
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11 The fracture toughness of the ceramics was evaluated using specimen bending testing which
12 contained a single edge through-thickness notch following ASTM C1421–10. Details of the
13 testing configuration and the notch profile are presented in ref. [8]. Two tests with loading rates
14 of 0.05 and 0.5 mm/min were performed.
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16 Hardness was determined by an MMT-7 Vickers hardness tester (Matsuzawa MMT-7;
17 Matsuzawa SEIKI Co., Ltd., Tokyo, Japan), using load 9.8 N with a dwell time of 15 s
18 following the standard procedure (ASTM C 1327–15).
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27 **3 Results and Discussion**

28 *3.1 Structure analysis of medium-entropy carbide ceramics*

29 **Figure 1** shows a summary of the XRD analysis performed within the present study. One can
30 see that in the case of the TTZN–40 (**Fig. 1 (b)**), a single-phase ceramic was formed. In the
31 case of the shorter dwell time (TTZN–10), the presence of two phases was suggested based on
32 the refinement results (**Table 1**). In this case, two phases were identified as quaternary carbides
33 with lattice parameters of $a = 4.4816 \text{ \AA}$ and $a = 4.4687 \text{ \AA}$. The lattice parameter of the
34 quaternary TTZN–40 phase in **Fig. 1 (b)** was estimated to be 4.4687 \AA . In this case structural
35 refinement presented in **Table 1** for the phase designated as TTZN–10_2 is valid for both the
36 TTZN–10 and TTZN–40 specimens.
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38 The difference in phases between the TTZN–10 and TTZN–40 ceramics is more obvious
39 during the analysis of the (442) peak (**Fig. 1 (c)**). For clarity, the $K\alpha_2$ peaks were not removed
40 from the data presented in **Fig. 1**. For the case of the TTZN–40, both the $K\alpha_1$ and $K\alpha_2$ peaks
41 are visible even using a 0.05° step size, while for the TTZN–10, a 0.02° step size and 40 s
42 sampling time does not allow separating these peaks. Thus it is postulated [20] that at least two
43 carbide phases are present for the TTZN–10 ceramic. Furthermore, the phase (TTZN–10_1 in
44 **Table 1**) with the lattice parameter $a = 4.4816 \text{ \AA}$ suffers from the severe contribution of strain
45 and misorientation [24].
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1 The difference between the phases based on the refinement results, and taking into account the
2 results of the EDX analysis (see below) can be understood based on the occupancy of the Ti
3 atom in the quaternary solid-solution (**Table 1, Fig. 2**). The theoretical density of the TTZN
4 specimens was estimated as 8.353 and 8.408 g/cm³. The bulk densities of the SPSed TTZN
5 ceramics are 8.31 g/cm³ (10 min dwell at 1977 °C) and 8.35 g/cm³ (40 min dwell at 1927 °C).
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7 Another detail was derived during the XRD analysis – a quasi-linear correlation between the
8 reported lattice parameters and the theoretical density for the medium-entropy/high-entropy
9 carbide ceramics (see **Fig. 1 (d)**) [14–16,19,25–27]. One can see a clear difference between the
10 medium-entropy (i.e. <5 principle elements), high-entropy [14,15,25–27] and binary carbides
11 [28,29] based on the two lines obtained using the regression analysis in Origin 7.5 (OriginLab
12 Corp.). This difference serves as an indication of the order/disorder in the complex solid
13 solution, but further confirmation based on compounds with a similar NaCl-type structure, e.g.,
14 nitrides, is required.
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18 The SEM micrographs for the ceramic bulks after the flexural tests at room temperature (**Figs.**
19 **3,4**) illustrate the difference between the TTZN–10 and TTZN–40 ceramics that were
20 consolidated using different conditions. The TTZN–10 ceramic consists of at least two carbide
21 phases, while the TTZN–40 was found to be a single-phase ceramic. In the case of the TTZN–
22 10 ceramic (**Figure 3**), it was possible to identify the Ti-rich (Ti – 30 mol.%; Ta, Zr, Nb – 23
23 mol.%) and equimolar solid-solution phases (Ta,Ti, Zr, Nb – 25 mol.%) using EDX. The
24 titanium-rich quaternary carbide phase occupied up to 32 vol.% which is in rough agreement
25 with 42.5 vol.% based on the XRD refinement.
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29 In terms of the fractographic analysis (using ImageJ, OrientationJ), after the flexural tests at
30 room temperature, a mixed fracture mode was observed for TTZN–10 and TTZN–40. For the
31 TTZN–10 ceramic, one can see the larger contribution of the transgranular fracture mode
32 compared to the TTZN–40 ceramic. Nevertheless, in both cases, one can observe the presence
33 of microcracks even after the tests at 1800 °C [14]. The mean grain size for the TTZN–10 and
34 TTZN–40 ceramics was 19±4 μm and 15±4 μm, respectively. In some instances fine grains
35 were observed for both ceramics with a grain size between 8 and 12 μm (up to 10 vol.%).
36 Grains exceeding 25 μm were extremely rare. Pores had a spherical shape and size between 1
37 and 2 μm. The total porosity based on the fractographic analysis was less than 1 vol.%.
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57 *3.2 Mechanical properties at room temperature*

58 Hardness of the bulk TTZN–10 and TTZN–40 ceramics was within 27.6±2.3 GPa, comparable
59 to that reported in refs. [14,18], but somehow higher than in [16]. In terms of the development
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1 of quaternary or quinary high-entropy carbides as a structural material for the high-
2 temperature applications, further optimization of the processing may be required. For the
3 TTZN ceramics, within the eight specimens attempted, only five were free of macroscopic
4 cracks [14,30]. Thus, it is thought that micro- and macrocracking is the reason for the absence
5 of the flexural strength data for these compounds. Because, even within the present study, only
6 specimens free of macroscopic cracks were used for the flexural tests (~40% of the specimens
7 had macroscopic cracks after SPS). Further optimization of the consolidation process is a
8 mandatory step to decrease the grain size (to at least 10 μm) in order to control the mechanical
9 performance and cracking issue [30,31].

10 For the crack-free specimens of the TTZN-40 ceramic, the flexural strength and fracture
11 toughness at room temperature were 544 ± 35 MPa and 3.2 ± 0.2 MPa $\text{m}^{1/2}$, respectively. For the
12 TTZN-10 ceramic, a slightly higher average strength was evaluated as 560 ± 49 MPa, while the
13 toughness was 3.3 ± 0.2 MPa $\text{m}^{1/2}$. In absolute values, the strength of the $(\text{Ti}_{1/4}\text{Ta}_{1/4}\text{Zr}_{1/4}\text{Nb}_{1/4})$
14 carbide ceramics are on the same level as reported for the TaC or TaB₂ ceramics [32-38].
15 However, because the formation of the medium-entropy or high-entropy ceramics allows
16 controlling the bulk density, the values of the specific strength were higher than those for the
17 TaC ceramics. Furthermore, one can see from Fig. 5 [14,16,31-48] that TTZN-10 has a
18 slightly higher absolute or specific strength compared to TTZN-40. This can be explained in
19 terms of the presence of the second phase that might act as strengthening elements, and, of
20 course, due to the formation of the solid solution and local stresses associated with the lattice
21 distortion.

22 3.3 Analysis of high-temperature properties

23 As a rule for the monolithic single phase ceramic, the presence of segregated impurities or
24 secondary phases may greatly influence the flexural strength at room temperature or at elevated
25 temperature. Thus in the following analysis, the data for the additive-free ceramics are being
26 used. Nevertheless, it should be noted that, if possible, the grain size reported by the authors of
27 the original study was mentioned. These actions were made to minimize the possibility for
28 incorrect data interpretation during analysis that is provided below.

29 It should be noted that the data for the UHTC-based ceramic composites and monolithic UHTC
30 may be differently interpreted [1,2,4]. The addition of a even minor amount of secondary
31 reinforcement phase (SiC, WC, WB, MoSi₂, etc) to the UHTC matrix will have its own
32 toughening or strengthening advantages [1-4]. Hence, when the monolithic carbide/diboride is
33 being improved by creation of a solid-solution and by adding a reinforcement phase, the

strengthening effect is more difficult to explain [4]. In the case of the studies of the monolithic (circa free of SiC, B₄C or MoSi₂, etc., additives) carbide or boride ceramics [4,31], it may be suggested that the strength improvement associated with the solid-solution formation for these monolithic ceramics can be (i) lost upon reheating to higher temperatures, (ii) maintained up to a higher temperature region, or (iii) can be suppressed by the contribution of local plastic deformation [14, 31] before the significant decrease in strength would occur due to macroscopic plastic deformation.

In this regard, one of the main goals of the present study was to analyze the high-temperature behavior of the medium-entropy (Ti_{1/4}Ta_{1/4}Zr_{1/4}Nb_{1/4}) carbide compared with data for other monolithic carbides (Figs. 6–12) [14, 31–62]. Our previous attempt for the ternary (Ta_{1/3}Zr_{1/3}Nb_{1/3}) carbide [14] suggested that the behavior of the ternary ceramic is comparable with data for the pure NbC [48,49] or NbC–ZrC [50] solid solution as their behavior is quite similar at elevated temperatures. The only difference was the onset of the significant contribution of the macroscopic plastic deformation, which was slightly higher for the NbC–ZrC case [50].

The results of the present study may provide an additional explanation for these observations. First, for the TTZN–10 or TTZN–40 ceramics, a decrease in strength was noticeable at 1000 °C, which can be explained as an effect of the reheating. The degree of strength degradation or strength vs the temperature dependence was different for the two medium-entropy carbide ceramics. This becomes clear after examination of **Figure 6 (b)**, in which the data for the low-temperature region of the TTZN–10 ceramic are approximated using the dashed line. One can see that in terms of the flexural strength at 1600 °C, the (Ta_{1/3}Zr_{1/3}Nb_{1/3}) carbide [14] and TTZN–10 have similar strength. Nevertheless, the (Ta,Zr,Nb) carbide shows a gradual increase in strength. Because this trend is similar to that for the NbC–ZrC data [50] (see **Fig. 7** [14,36,40,41,48–51]), it was concluded in [14] that this may be due to the governing role of the solid-solution strengthening with respect to the flexural strength behavior of the single-phase medium-entropy (Ta,Zr,Nb) carbide.

Data for the temperature dependence of the individual monolithic carbides of the IV and V groups with the NaCl-type lattice are summarized in **Figures 8–12** [36,40–44,47–49,51–62]. Although HfC was not used in the present study, we attempted to analyze the data for the monolithic hafnium carbide since we consider the addition of HfC to study the (Ta,Zr,Nb,Hf)C, (Ti,Zr,Nb,Hf)C and (Ti,Ta,Zr,Nb,Hf)C carbide ceramics.

The high-temperature flexural behavior of titanium carbide (**Figure 8**) [40–43,52] can be summarized as follows: a slight variation in strength before 1400–1500 °C (brittle to ductile

1 transition temperature, BDTT), followed by a rapid strength decrease due to activation of the
2 macroscopic plastic deformation [31,63]. TaC also follows a similar trend (see **Figure**
3 **11**[48,49,51,52,58]), but a study [59] reported a deep minimum at 1400 °C, followed by an
4 increase in strength. Data for the BDTT of tantalum carbide showed quite a wide range from
5 1550 °C to 2300 °C [49,61,64,65]. Similar to the flexural strength, the BDTT is sensitive to a
6 carbon deficiency [65], method of fracture, consolidation or annealing conditions, and strain
7 rates [64].
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12 The data for ZrC [44,47,51,53,54] and NbC [48,49,51,52,58] are slightly different, and they
13 are fully discussed in the study of ref. [31]. For these ceramics, the transition between brittle
14 and ductile fracture is associated with a bell-shaped strength dependence (i.e., with a clear
15 maximum). The peak strength for these ceramics is usually observed in the vicinity of the
16 BDTT, but the position of the peak is quite sensitive to (a) grain size and (b) consolidation
17 conditions, i.e., a higher consolidation temperature allows shifting the peak strength to higher
18 temperatures. Flexural strength data for HfC are scarce [55–57]. Within the available sources,
19 two studies indicate a dependence typical for TiC, but one study [57] reported a gradual
20 increase in strength up to 2200 °C. As noted above, data of the flexural strength of binary
21 carbides are limited by [50], in which the NbC–ZrC solid-solution follows a trend similar to
22 the individual NbC or ZrC. Based on these results, one can underline the importance of studies
23 about binary carbides and carbides featuring HfC, as an increase in strength after 2000 °C for
24 the monolithic HfC reported in refs. [55] and [57] looks quite interesting from theoretical and
25 practical viewpoints.
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29 The shaded strength window highlighted in **Fig. 7** [14,36,40,41,48–51] serves as an indication
30 of an abnormal strength behavior and perhaps is best suited for the practical applications of
31 these compounds. Typical loading curves at 1600 °C and 1800 °C suggest that a fully elastic
32 behavior is observed for the TTZN ceramics at 1600 °C, while a deviation from elastic behavior
33 serves as an indication of the ongoing plastic deformation at 1800 °C [21,35,63]. The yield
34 stress instead of fracture stress was used for the analysis in **Fig. 6** of specimens tested at
35 1800 °C.
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39 As noted above, both data sets in this study show a gradual decrease in strength, thus one can
40 interpret this situation in terms of the magnitude of the local stresses stored and released upon
41 reheating to high temperatures. Because of the relatively high flexural strength at room
42 temperature observed for the TTZN carbides within this study (>500 MPa, comparable to bulk
43 TaC [32,34,36]), it is suggested that the thermal history of the specimens during consolidation
44 allowed activation accumulation of a higher lattice strain, which was relaxed at the higher
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1 temperatures. Microcracking was reported to be visible even at 2000 °C for the (Ta,Zr,Nb)
2 carbide [14]. Because the appearance of microcracks was noticed at temperatures above
3 1600 °C for (Ti,Ta,Zr,Nb) carbides within the present study, it may be suggested that only
4 high-temperature annealing above >2000 °C may help one to understand the exact mechanism
5 of the stress relaxation. Furthermore, for the binary carbide systems, a miscibility gap may
6 differ from 1880 °C to 2050 °C for the TiC–HfC system [66], while the miscibility gap for the
7 ternary (TiC)_{0.27}(HfC)_{0.41}(WC)_{0.32} system was studied at 1540 °C [66,67]. As noted in ref. [67],
8 the binary, ternary and multi-carbides of the metals of the IV and V groups (excluding VC)
9 addition of NbC, in particular to the ternary system may expand the single-phase limit after a
10 2000 °C/12h annealing.

11 Another important observation that was made during the high-temperature flexural strength is
12 that Young’s modulus degradation for the medium-entropy TTZN carbides was not as severe
13 as for the monolithic carbides. Following the analysis of ref. [68], **Figure 13** shows the
14 evolution of the specific stiffness (based on E_f values) and absolute values of Young’s modulus
15 at room temperature and at 1600 °C [4,5,68]. Data for the (Ti,Ta,Zr,Nb)C carbides favorably
16 agree with the modulus of the TaC data, but as noted previously due to a sufficient decrease in
17 the theoretical density, the TTZN ceramics have a higher specific stiffness. HEC stands for
18 data on elastic moduli at room-temperature data from ref. [19] (data for carbides with valence
19 electron concentration between 8.6 and 9 were used). Furthermore, a slight deviation from the
20 equimolar ratio observed for the TTZN–10 ceramic phases (**Fig. 1, Table 1**) may also
21 contribute to the higher modulus values as was noted that for some binary carbide systems, the
22 maximum in modulus or creep-resistance was observed for the 40/60 values rather than the
23 50/50 value, as the former concentration is believed to be responsible for the maximum in the
24 distortion of the crystal lattice [4,69].

25 This underlines the fact that for medium-entropy or high-entropy carbide ceramics, the effect
26 of the processing conditions (at least for an *in situ* approach [13,14]) can be tailored to govern
27 the solid-solution strengthening and to manufacture carbide ceramics with a high strength and
28 high stiffness up to 1600 °C. The contribution of various factors can be further optimized but
29 required an in-depth knowledge of the carbides behavior for ternary, quaternary, and
30 quintenary equimolar carbide systems.

31 To understand the effect of the solid-solution, it is suggested that high-entropy ceramics with
32 different grain sizes should be manufactured similar to study [16]. This underlines the
33 importance of further consolidation kinetic studies of these ceramics. These studies and high-

1
2 temperature characterization of the equimolar (Ta,Zr,Nb,Hf)C, (Ti,Zr,Nb,Hf)C and
3 (Ti,Ta,Zr,Nb,Hf)C carbide ceramics are considered as a crucial step in the ongoing research.
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5 6 **4 Summary and conclusions**

7 Bulk solid-solution or medium-entropy carbide ceramics have been obtained by two different
8 thermal histories using the spark plasma sintering method. Phase analysis via lattice parameter
9 measurements by X-ray diffraction showed the processing conditions such as dwell time and
10 consolidation temperature, allow one to produce single-phase or multi-phase carbide ceramics
11 with the lattice parameters of the phases of $a = 4.4816 \text{ \AA}$ and $a = 4.4687 \text{ \AA}$.
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13 The mechanical performance at room temperature for the (Ti,Ta,Zr,Nb) carbides is comparable
14 to the best data for the TiC or TaC monolithic ceramics, reaching 600 MPa. This may be
15 explained in terms of a solid-solution strengthening mechanism. In terms of the specific
16 flexural strength or specific stiffness, medium-entropy carbide bulks are superior to the
17 majority of the monolithic carbides, except the TiC. Titanium carbide is known to have a brittle
18 to ductile transition temperature around 1200–1300 °C, which does not allow using it as per se
19 high-temperature ceramics. Importantly, at elevated temperatures, medium-entropy
20 (Ti,Ta,Zr,Nb) carbides possess the strength of 300 or 500 MPa at 1600 °C where they fracture
21 in an elastic manner. The strength at high temperatures depends on the consolidation
22 conditions. Based on the results of this study, consolidation using a higher temperature but
23 shorter dwell time is recommended.
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Tables

Table 1. Results of structure refinement of carbide phases for TTZN–10 ceramic.

Designation	TTZN–10_1	TTZN–10_2 ^Y
Lattice parameter, Å	4.4816	4.4687
Ti occupancy	0.2400	0.2755
Ta occupancy	0.2582	0.2410
Zr occupancy	0.2568	0.2417
Nb occupancy	0.2569	0.2416
C occupancy*	1	1
Rwp / GOF	4.7 / 2.2	

* fixed to 0.99 during structure refinement

^Y main phase for the TTZN–40 ceramic

Figure Captions

Fig. 1. XRD patterns for TTZN carbide ceramic after spark plasma consolidation. (a) shows a refinement procedure for the TTZN–10 ceramics. (b) provide refinement of the TTZN–40 specimen. The lattice parameters for two solid solutions were: (i) $a = 4.4963 \text{ \AA}$ and (ii) $a = 4.4682 \text{ \AA}$. For the volume fraction of the two phases was 42.5 and 57.5 %. (c) shows a refinement of the (442) peak. For clarity, the $K\alpha_2$ peaks were not removed for (a)–(c). The bars indicate the allowed Bragg reflections for the $Fm-3m$ structure. (d) shows relation between the bulk density and lattice parameters for the monolithic, binary, medium-entropy, and high-entropy carbides (using an XRD database survey) [14–16,25–29]. The difference between a general density vs lattice trend (dashed line) and that for medium-entropy/high-entropy carbides (dotted line) may indicate a more severe lattice distortion.

Fig. 2. Visual representation of medium-entropy carbide phases derived during phase refinement using a $2 \times 2 \times 1$ supercell ($C_{16}Nb_4Ta_4Ti_4Zr_4$). An equimolar solid-solution (Nb,Zr)C via (PDF #65-8790, [28]) was visualized as a reference ($C_{16}Nb_8Zr_8$).

Fig. 3. SEM micrographs of medium-entropy carbides after flexural tests at ambient temperature: (a,b) TTZN–10, (c–e) TTZN–40. (e) shows typical microcracks observed during fracture. In order to observe homogeneity of distribution of metals with different atomic numbers in medium-entropy carbide (b,d) are provided in the BSE mode.

Fig. 4. SEM micrographs of medium-entropy carbide after flexural strength tests at elevated temperatures: (b,c) at $1000 \text{ }^\circ\text{C}$, (d-g) and (h-j) $1600 \text{ }^\circ\text{C}$ and $1800 \text{ }^\circ\text{C}$, respectively. (c, e, g, i, k) are taken in BSE mode. (a) provides a statistical variation between transgranular and intergranular fracture for all specimens tested within the present study. Dashed circles in (k) show locations of microcracks. (d, e, h, i) are TTZN–40, while (b, c, f, g, j, k) are TTZN–10.

Fig. 5. Effect of the grain size and composition of selected high-strength carbide and diboride ceramics on normalized strength values. Dashed lines provide a slope verified for tantalum diboride within study [37]. Note that the data for quaternary medium-entropy carbides in this study comport the trend for TaB_2 or TaC , but at the same time show an opposite trend, as the highest strength at room temperature has been observed for ceramics with the grain size of $19 \pm 4 \text{ }\mu\text{m}$ (TTZN–10 ceramic, $566 \pm 12 \text{ MPa}$). This serves as a direct confirmation of the solid-solution strengthening effect. Data on binary and ternary carbides were collected in [14]. Data of ref. [16] show that decrease in grain size allows increasing strength up to by 20%.

Fig. 6. Effect of temperature on the flexural strength of TTZN carbides prepared in this study. (a) provides variation in strength for TTZN–10 ceramic, while (b) shows data for TTZN–40

1 ceramic and for ternary (Ta,Zr,Nb)C [14]. Dashed lines in (a) and (b) show the linear regression
2 obtained from data in (a). (c) shows typical loading curves for TTZN-40 ceramics at 1600 °C
3 and 1800 °C.
4

5 **Fig. 7.** Effect of temperature on the flexural strength of TTZN carbides and selected transition
6 metal carbide monoliths [14, 36, 40, 41, 48–51] (see **Figures 8–12** [36,40–44,46–49,51–62]
7 for details). Argon was used during the high-temperature flexural test for all the reported data.
8 The closed symbols indicate that the strength was measured using a four-point setup and the
9 open symbols show the results of the three-point flexural strength tests. Shaded area underlines
10 sufficient increase in strength between 25 °C and 1600 °C obtained for ceramics with carbide
11 solid-solid solutions, see **Figure 6 (b)** for clarity.
12

13 **Fig. 8.** Effect of temperature on the flexural strength of monolithic titanium carbide ceramics
14 [40–44,52]. The semi-closed symbols indicate that the strength was measured using a four-
15 point setup and the open symbols show the results of the three-point flexural strength tests. If
16 available, average grain size (G.S.) values are presented
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18 **Fig. 9.** Effect of temperature on the flexural strength of monolithic zirconium carbide ceramics
19 [44,46,51,53,54]. The semi-closed symbols indicate that the strength was measured using a
20 four-point setup and the open symbols show the results of the three-point flexural strength tests.
21 If available, the average grain size (G.S.) values are presented.
22

23 **Fig. 10.** Effect of temperature on the flexural strength of monolithic hafnium carbide ceramics
24 [55–57]. All reported data were collected using three-point flexural strength test. If available,
25 the average grain size (G.S.) values are presented.
26

27 **Fig. 11.** Effect of temperature on the flexural strength of monolithic niobium carbide ceramics
28 [48,49,51,52,58]. All reported data were collected using three-point flexural strength test. If
29 available, the average grain size (G.S.) values are presented.
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31 **Fig. 12.** Effect of temperature on the flexural strength of monolithic tantalum carbide ceramics
32 [36, 59–62]. All reported data were collected using three-point flexural strength test. If
33 available, the average grain size (G.S.) values are presented.
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35 **Fig. 13.** Effect of ceramic composition on elastic moduli and specific stiffness of commonly
36 used high-temperature ceramics. Values for stiffness at 1600 °C were estimated using approach
37 provided in ref. [68], while the data for carbide ceramics show the Young's modulus corrected
38 to zero porosity. HE – stands for high-entropy carbides based on data of ref. [19], while TTZ
39 stands for $(\text{Ta}_{1/3}\text{Zr}_{1/3}\text{Nb}_{1/3})\text{C}$ [14]. Note, that the quaternary medium-entropy carbides have a
40 minor modulus decrease at 1600 °C.
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Figures

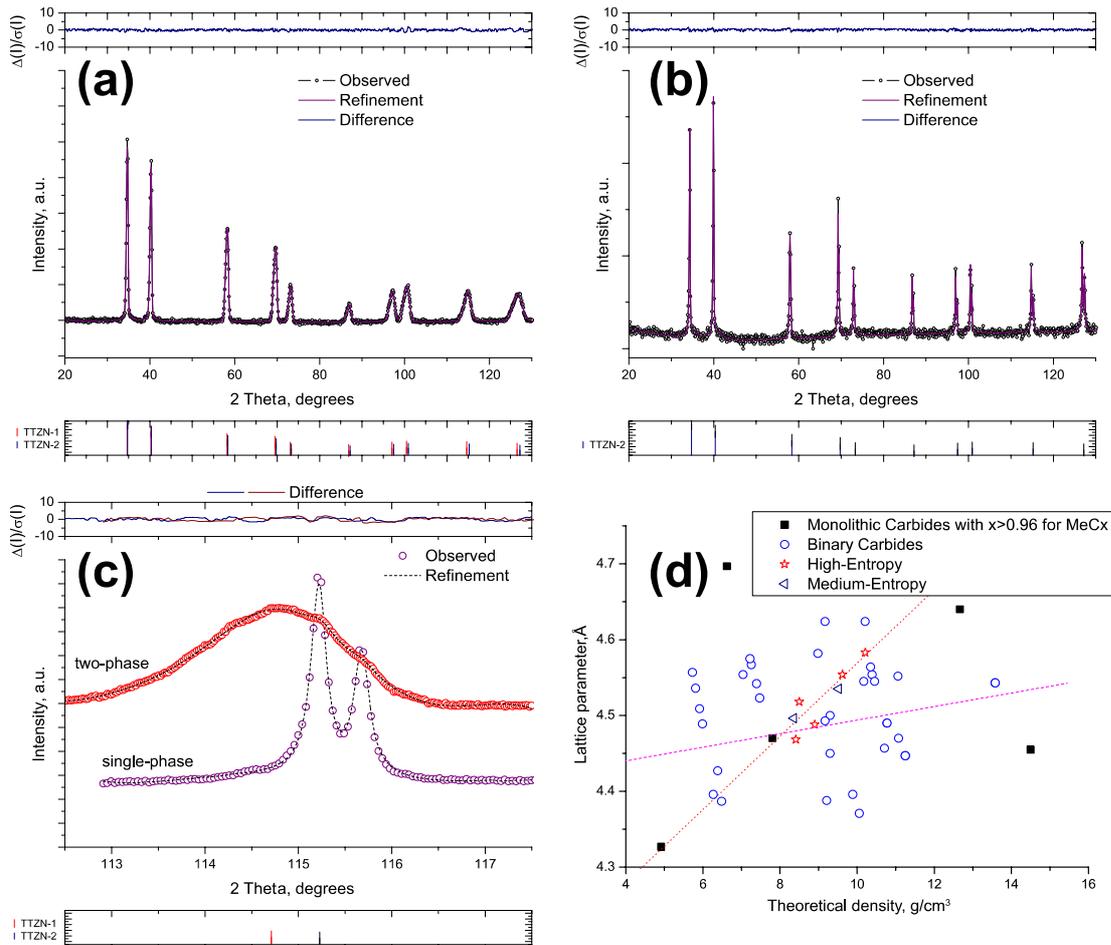
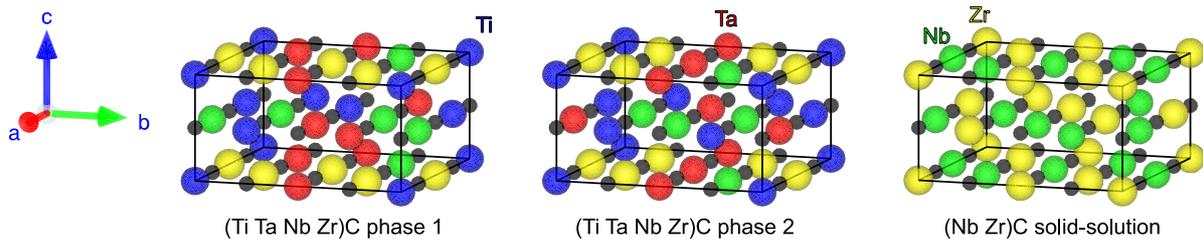
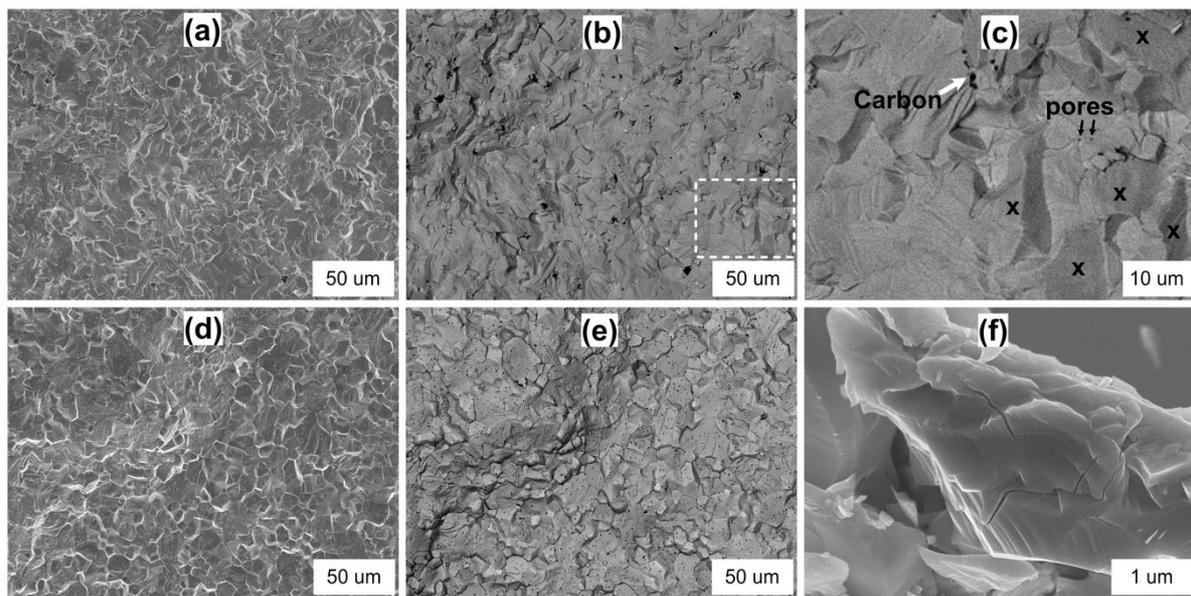


Fig. 1. XRD patterns for TTZN carbide ceramic after spark plasma consolidation. (a) shows a refinement procedure for the TTZN–10 ceramics. (b) provide refinement of the TTZN–40 specimen. The lattice parameters for two solid solutions were: (i) $a = 4.4963 \text{ \AA}$ and (ii) $a = 4.4682 \text{ \AA}$. For the volume fraction of the two phases was 42.5 and 57.5 %. (c) shows a refinement of the (442) peak. For clarity, the $K\alpha_2$ peaks were not removed for (a)–(c). The bars indicate the allowed Bragg reflections for the $Fm-3m$ structure. (d) shows relation between the bulk density and lattice parameters for the monolithic, binary, medium-entropy, and high-entropy carbides (using an XRD database survey) [14–16,25–29]. The difference between a general density vs lattice trend (dashed line) and that for medium-entropy/high-entropy carbides (dotted line) may indicate a more severe lattice distortion.



11 **Fig. 2.** Visual representation of medium-entropy carbide phases derived during phase
12 refinement using a 2x2x1 supercell ($C_{16}Nb_4Ta_4Ti_4Zr_4$). An equimolar solid-solution
13 (Nb,Zr)C via (PDF #65-8790, [28]) was visualized as a reference ($C_{16}Nb_8Zr_8$).
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38 **Fig. 3.** SEM micrographs of medium-entropy carbides after flexural tests at ambient
39 temperature: (a–c) TTZN–10, (d–f) TTZN–40. (f) shows typical microcracks observed during
40 fracture. In order to observe homogeneity of distribution of metals with different atomic
41 numbers in medium-entropy carbide (b,d,c) are provided in the BSE mode. (c) shows presence
42 of carbon, pores and second solid-solution carbide phase (marked by x).
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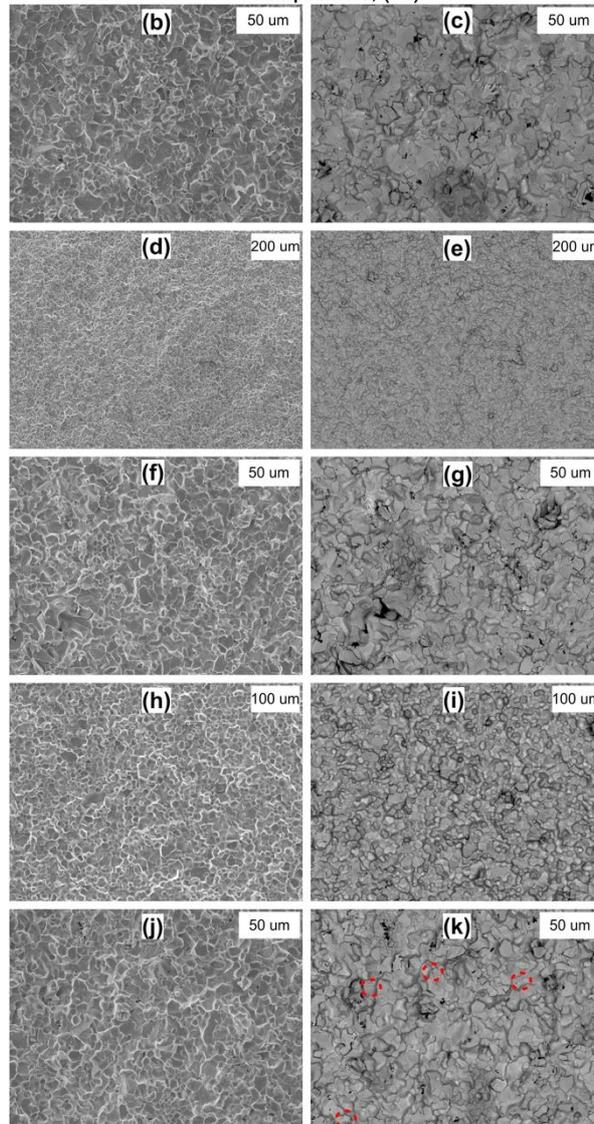
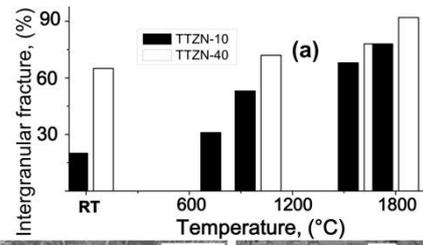


Fig. 4. SEM micrographs of medium-entropy carbide after flexural strength tests at elevated temperatures: (b,c) at 1000 °C, (d-g) and (h-j) 1600 °C and 1800 °C, respectively. (c, e, g, i, k) are taken in BSE mode. (a) provides a statistical variation between transgranular and intergranular fracture for all specimens tested within the present study. Dashed circles in (k) show locations of microcracks. (d, e, h, i) are TTZN-40, while (b, c, f, g, j, k) are TTZN-10.

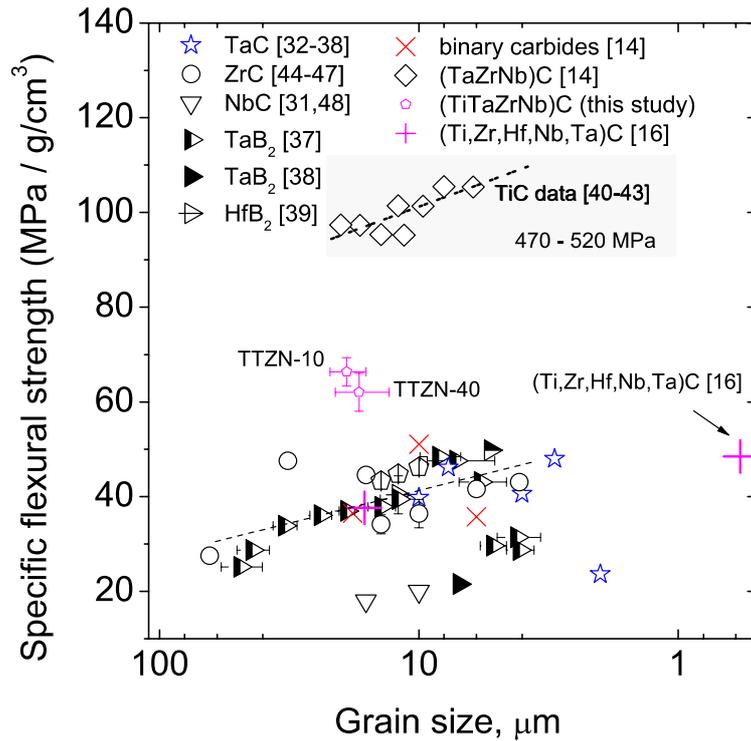


Fig. 5. Effect of the grain size and composition of selected high-strength carbide and diboride ceramics on normalized strength values. Dashed lines provide a slope verified for tantalum diboride within study [37]. Note that the data for quaternary medium-entropy carbides in this study comport the trend for TaB₂ or TaC, but at the same time show an opposite trend, as the highest strength at room temperature has been observed for ceramics with the grain size of 19±4 μm (TTZN–10 ceramic, 566±12 MPa). This serves as a direct confirmation of the solid-solution strengthening effect. Data on binary and ternary carbides were collected in [14]. Data of ref. [16] show that decrease in grain size allows increasing strength up to by 20%.

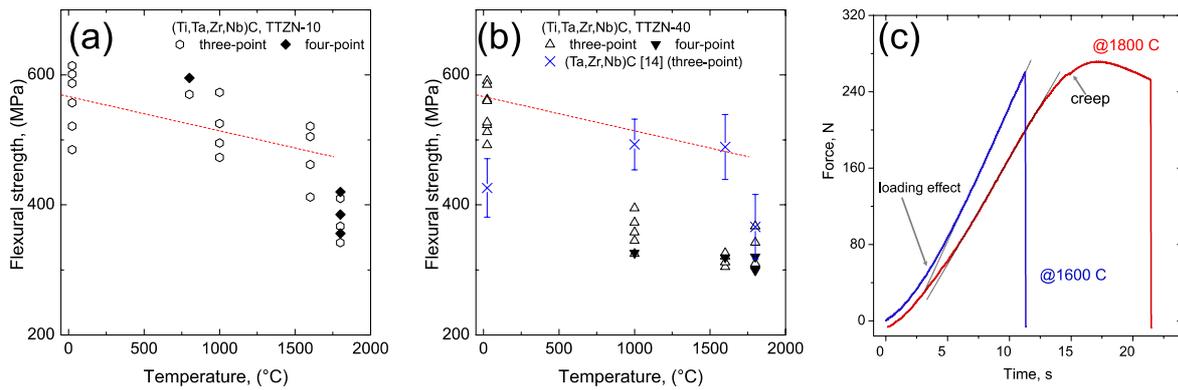


Fig. 6. Effect of temperature on the flexural strength of TTZN carbides prepared in this study. (a) provides variation in strength for TTZN-10 ceramic, while (b) shows data for TTZN-40 ceramic and for ternary (Ta,Zr,Nb)C [14]. Dashed lines in (a) and (b) show the linear regression obtained from data in (a). (c) shows typical loading curves for TTZN-40 ceramics at 1600 °C and 1800 °C.

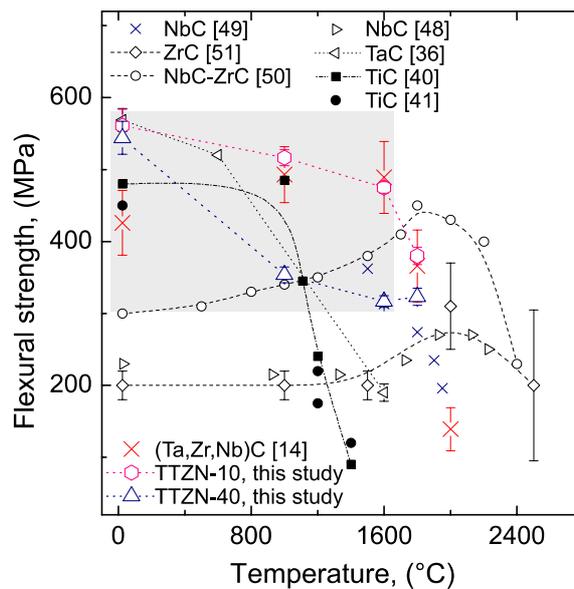


Fig. 7. Effect of temperature on the flexural strength of TTZN carbides and selected transition metal carbide monoliths [14, 36, 40, 41, 48–51] (see **Figures 8–12** [36,40–44,46–49,51–62] for details). Argon was used during the high-temperature flexural test for all the reported data. The closed symbols indicate that the strength was measured using a four-point setup and the open symbols show the results of the three-point flexural strength tests. Shaded area underlines sufficient increase in strength between 25 °C and 1600 °C obtained for ceramics with carbide solid-solid solutions, see **Figure 6 (b)** for clarity.

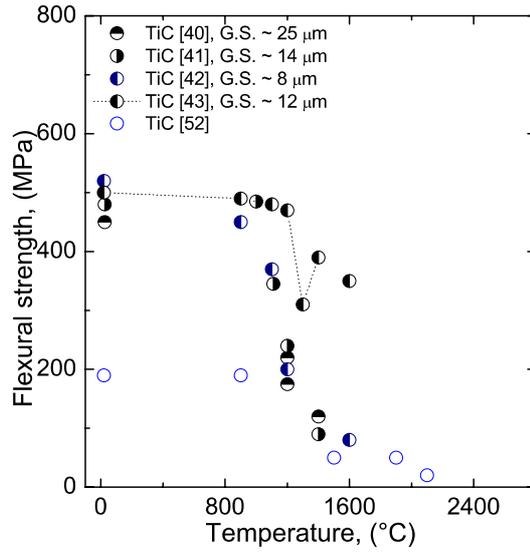


Fig. 8. Effect of temperature on the flexural strength of monolithic titanium carbide ceramics [40–44,52]. The semi-closed symbols indicate that the strength was measured using a four-point setup and the open symbols show the results of the three-point flexural strength tests. If available, average grain size (G.S.) values are presented

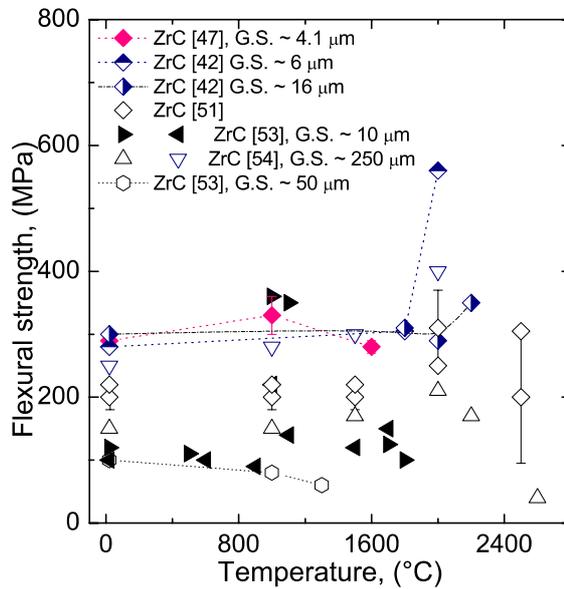


Fig. 9. Effect of temperature on the flexural strength of monolithic zirconium carbide ceramics [44,46,51,53,54]. The semi-closed symbols indicate that the strength was measured using a four-point setup and the open symbols show the results of the three-point flexural strength tests. If available, the average grain size (G.S.) values are presented.

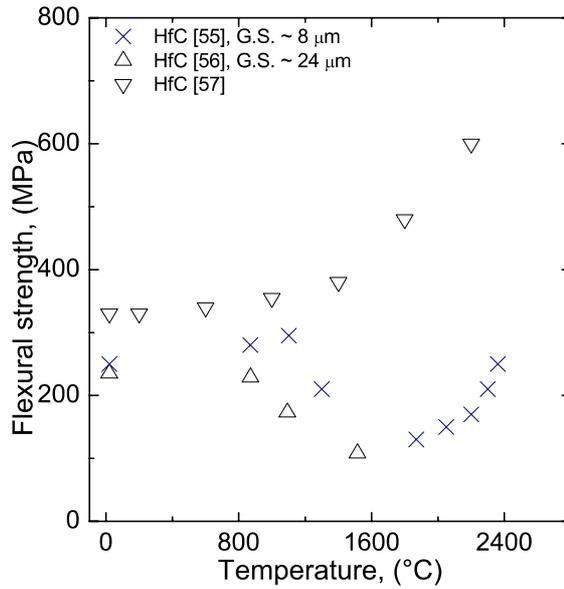


Fig. 10. Effect of temperature on the flexural strength of monolithic hafnium carbide ceramics [55–57]. All reported data were collected using three-point flexural strength test. If available, the average grain size (G.S.) values are presented.

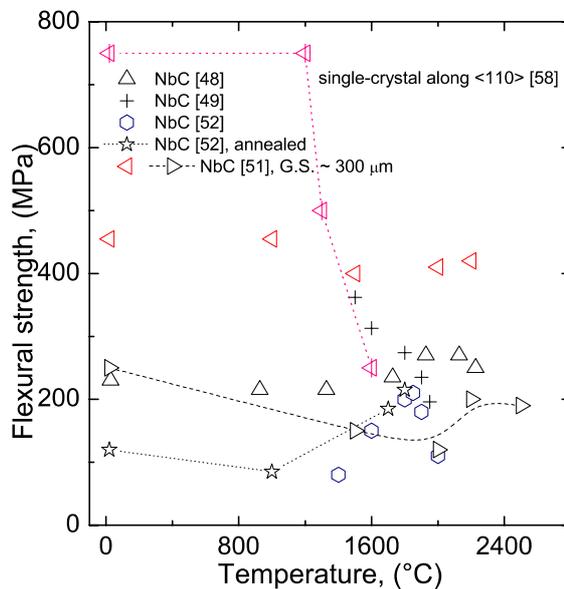


Fig. 11. Effect of temperature on the flexural strength of monolithic niobium carbide ceramics [48,49,51,52,58]. All reported data were collected using three-point flexural strength test. If available, the average grain size (G.S.) values are presented.

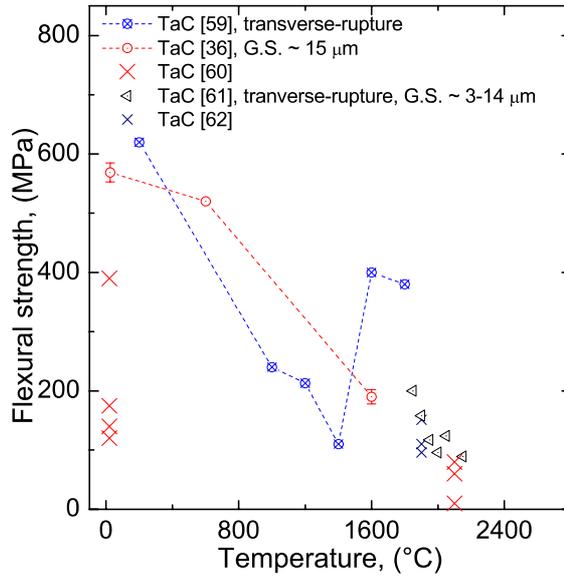


Fig. 12. Effect of temperature on the flexural strength of monolithic tantalum carbide ceramics [36, 59–62]. All reported data were collected using three-point flexural strength test. If available, the average grain size (G.S.) values are presented.

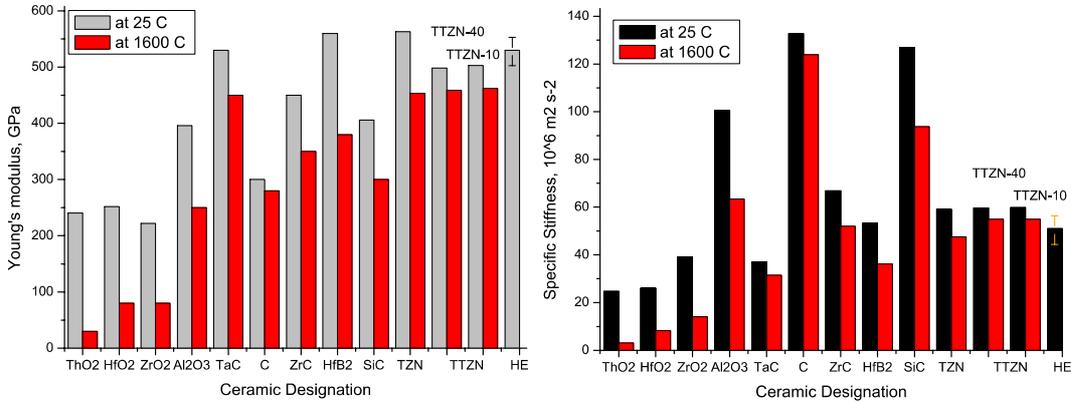
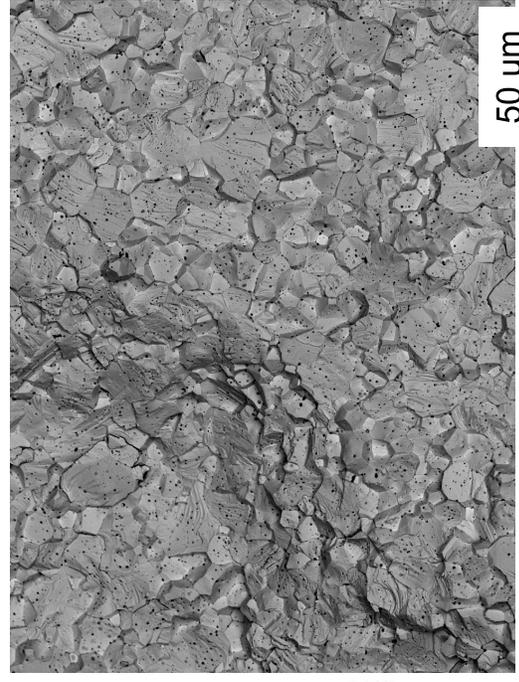
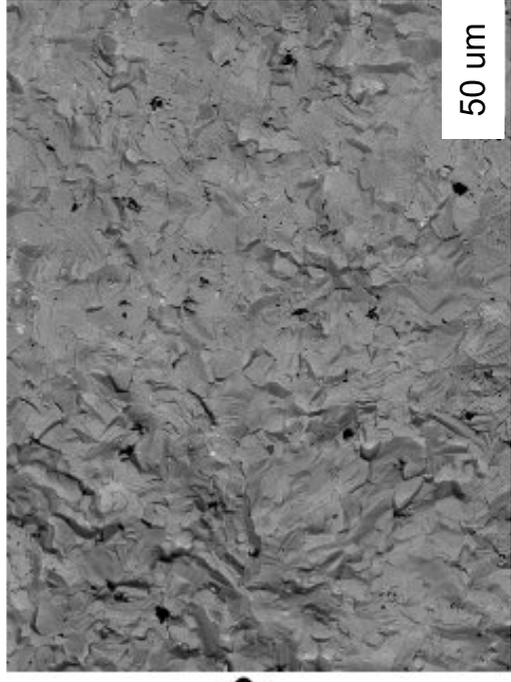


Fig. 13. Effect of ceramic composition on elastic moduli and specific stiffness of commonly used high-temperature ceramics. Values for stiffness at 1600 °C were estimated using approach provided in ref. [68], while the data for carbide ceramics show the Young’s modulus corrected to zero porosity. HE – stands for high-entropy carbides based on data of ref. [19], while TTZ stands for $(\text{Ta}_{1/3}\text{Zr}_{1/3}\text{Nb}_{1/3})\text{C}$ [14]. Note, that the quaternary medium-entropy carbides have a minor modulus decrease at 1600 °C.



strength 320 MPa @ 1600°C



strength 475 MPa @ 1600°C

Spark plasma sintering at 1927°C



Spark plasma sintering at 1977°C

