

Supporting Information (Online Material) for

Anisotropic Thermal Expansion and a Second-order Charge Order Transition in the Ferrimagnetic $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ Perovskite with Triple A-site Cation Ordering

Alexei A. Belik*

*Research Center for Materials Nanoarchitectonics (MANA), National Institute for
Materials Science (NIMS), Namiki 1-1, Tsukuba, Ibaraki 305-0044, Japan*

E-mail: Alexei.Belik@nims.go.jp

Table S1. Selected Bond Lengths (l (Å) < 2.8 Å), Bond Angles (deg), Bond Valence Sums, BVS, and Distortion Parameters of MnO₆, Δ, in Dy₂CuZnMn₄O₁₂ at 100 K ^a

Dy1–O5 ×2	2.289(9)	Dy2–O2 ×2	2.336(9)
Dy1–O3 ×2	2.340(8)	Dy2–O4 ×2	2.378(9)
Dy1–O2 ×2	2.400(9)	Dy2–O5 ×2	2.418(9)
Dy1–O1 ×4	2.656(7)	Dy2–O1 ×4	2.641(7)
BVS(Dy1 ³⁺)	+3.39	BVS(Dy2 ³⁺)	+3.18
Cu–O1 ×4	1.951(3)	Zn–O3 ×2	1.965(9)
BVS(Cu ²⁺)	+1.92	Zn–O4 ×2	2.044(9)
		BVS(Zn ²⁺)	+1.79
Mn1–O2 ×2	1.890(2)	Mn2–O5 ×2	1.901(2)
Mn1–O3 ×2	1.963(3)	Mn2–O4 ×2	1.932(3)
Mn1–O1 ×2	2.129(7)	Mn2–O1 ×2	1.946(8)
BVS(Mn1 ³⁺)	+3.30	BVS(Mn2 ⁴⁺)	+3.76
Δ(Mn1–O)	25.0×10 ⁻⁴	BVS(Mn2 ³⁺)	+3.84
		Δ(Mn2–O)	1.0×10 ⁻⁴
Mn1–O1–Mn2 ×2	144.84(9)	Mn2–O4–Mn2	139.94(9)
Mn1–O2–Mn1	147.53(9)	Mn2–O5–Mn2	145.03(9)
Mn1–O3–Mn1	134.91(9)		

^a $BVS = \sum_{i=1}^N \nu_i$, $\nu_i = \exp[(R_0 - l_i)/B]$, N is the coordination number, $B = 0.37$, $R_0(\text{Dy}^{3+}) = 2.036$, $R_0(\text{Cu}^{2+}) = 1.679$, $R_0(\text{Zn}^{2+}) = 1.704$, $R_0(\text{Mn}^{4+}) = 1.753$, and $R_0(\text{Mn}^{3+}) = 1.76$.

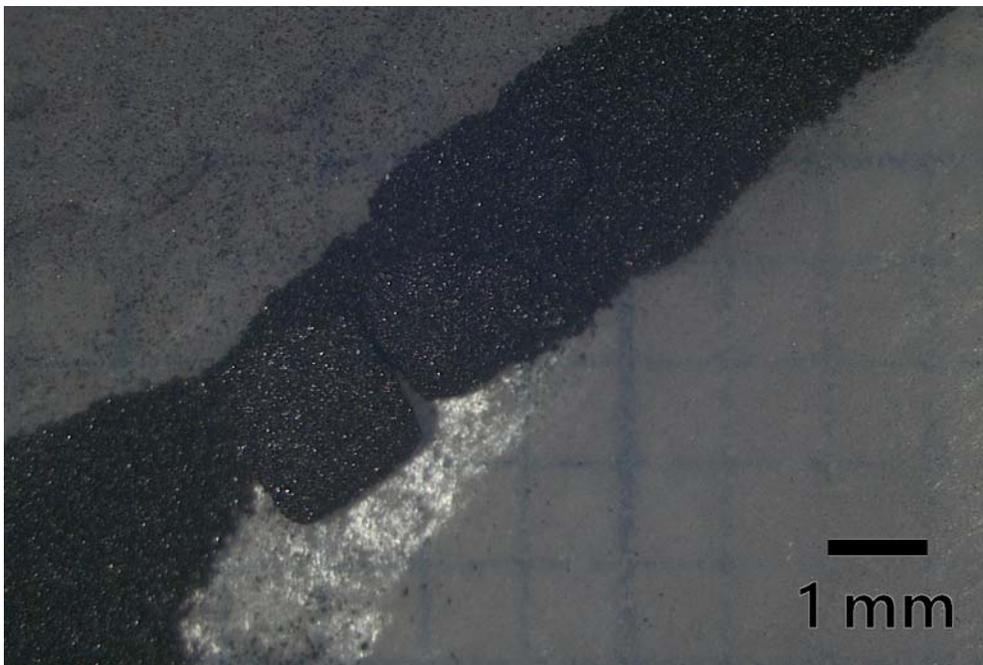
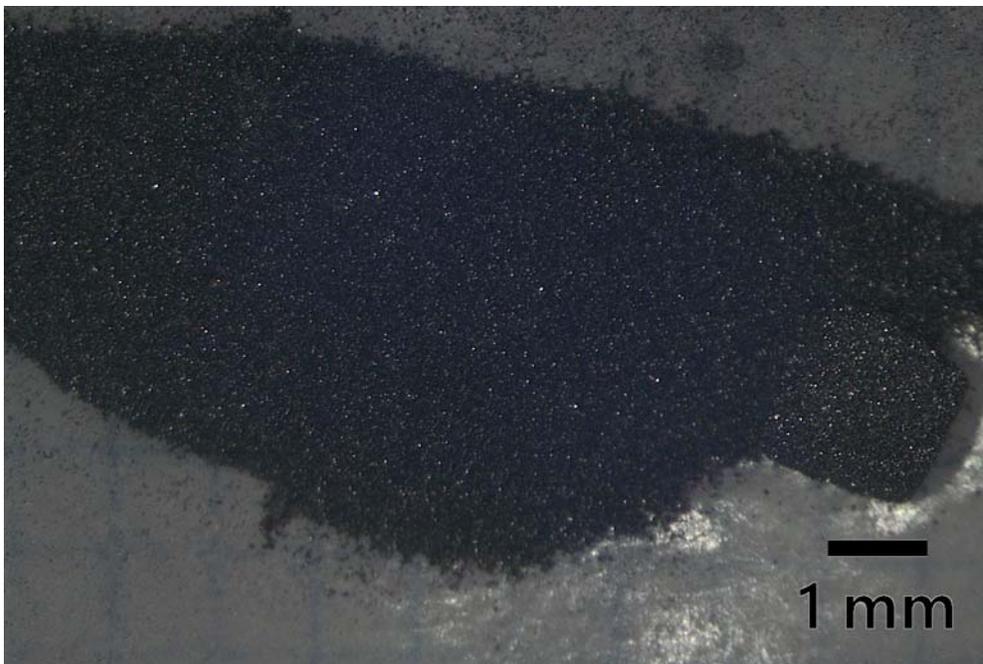


Figure S1. Photographs of the as-synthesized $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ sample. This figure shows that the as-synthesized sample was a highly fragile pellet, which fell into powder when touched.

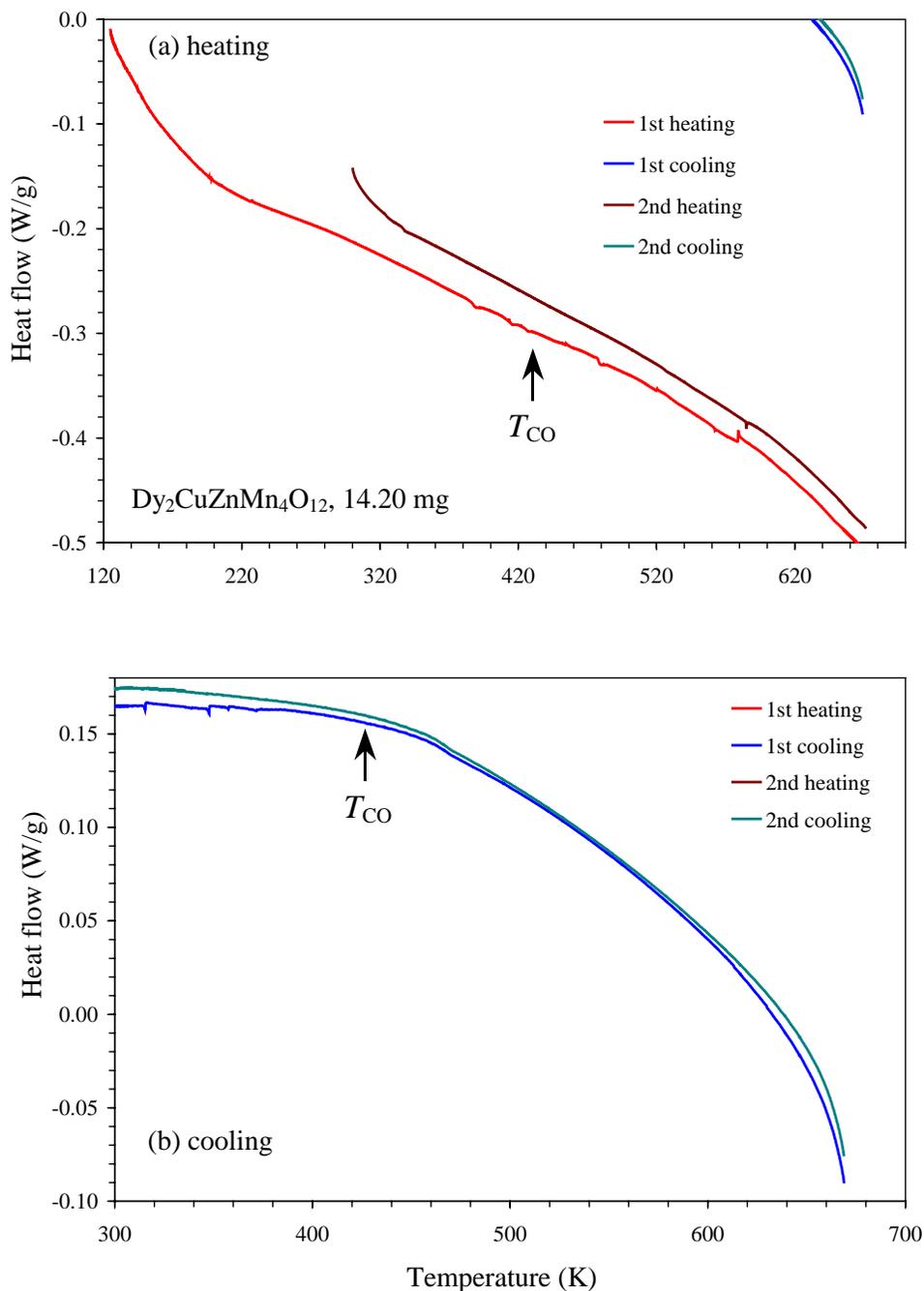


Figure S2. Differential scanning calorimetry (DSC) curves of a powder sample of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ (14.20 mg) on (a) heating and (b) cooling. The arrows show the charge-order phase transition temperature determined by high-temperature synchrotron X-ray powder diffraction – no anomalies were detected at this temperature by DSC.

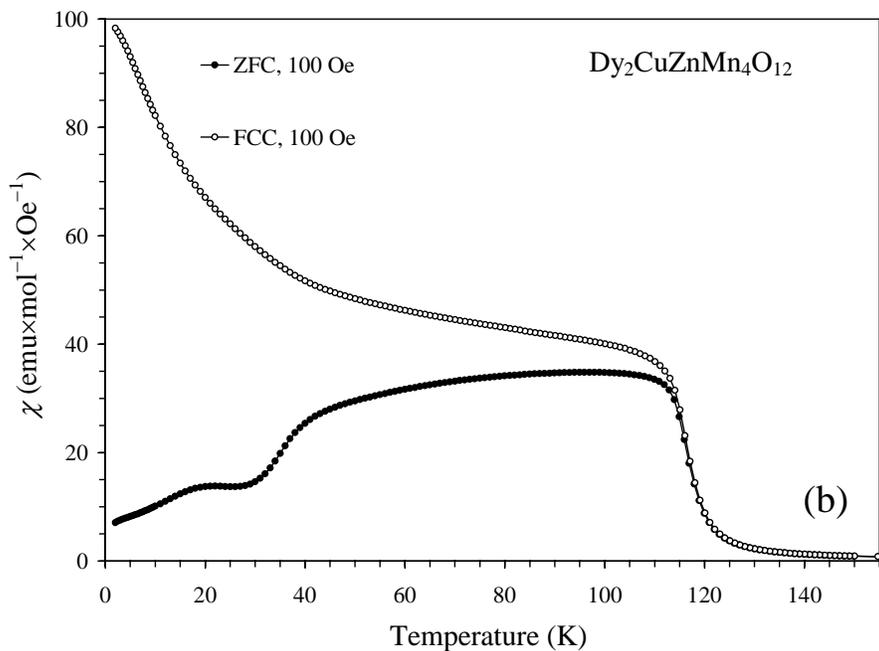
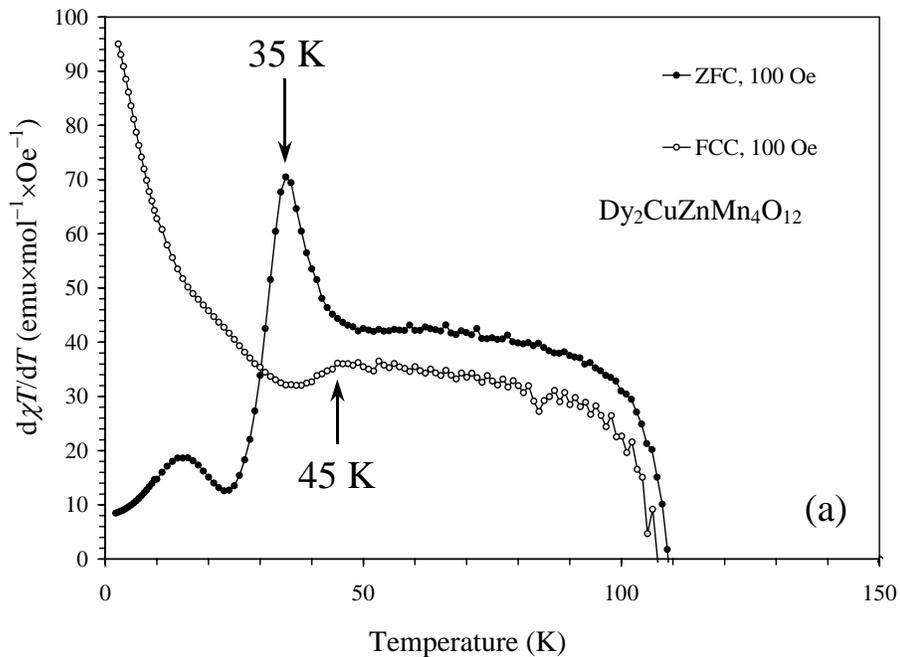


Figure S3. (a) A fragment of the ZFC and FCC $d\chi T/dT$ versus T curves of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at $H = 100$ Oe to emphasize anomalies near 35–45 K. (b) The ZFC and FCC χ versus T curves of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at $H = 100$ Oe below 155 K.

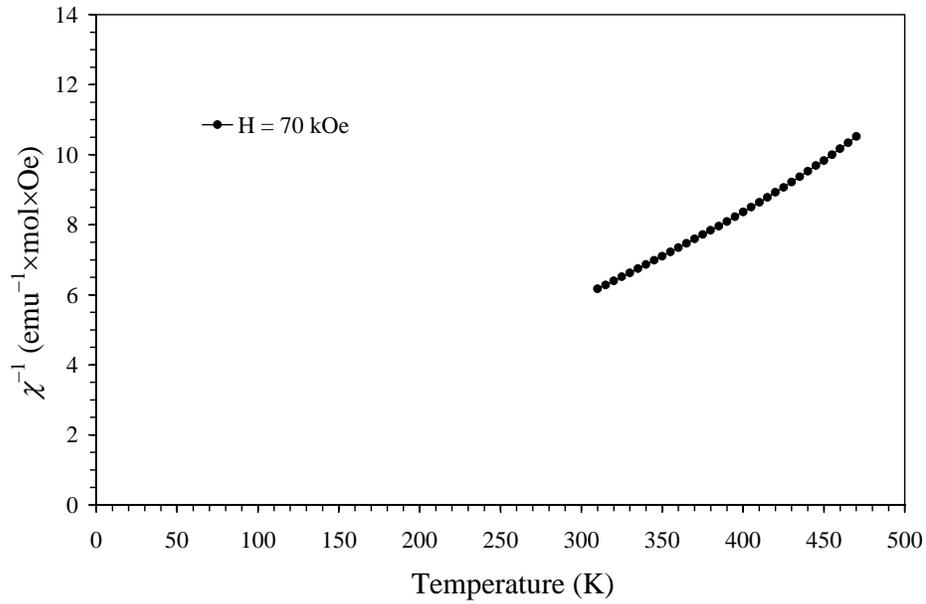


Figure S4. (Raw/uncorrected) high-temperature inverse χ^{-1} versus T curve of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at $H = 70$ kOe measured from 300 K to 470 K. A small deviation from the linear behavior at high temperatures is caused by a temperature-independent contribution from a sample holder. There were no detectable anomalies near $T_{\text{CO}} = 425$ K.

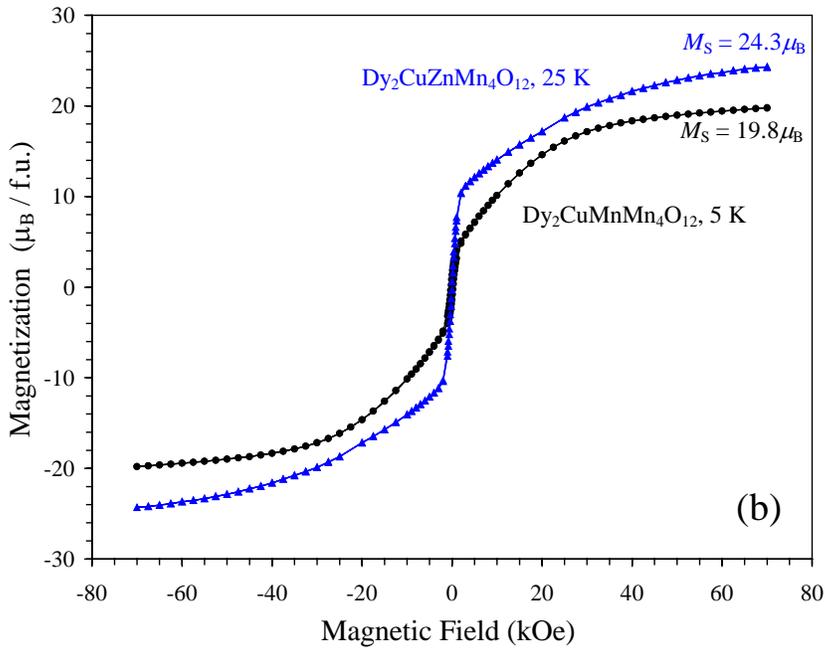
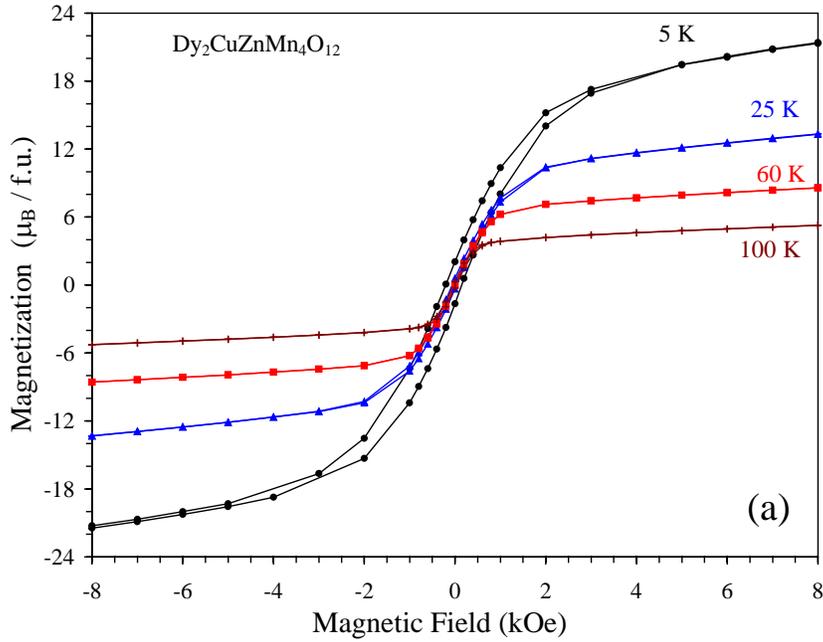


Figure S5. (a) A zoomed part of the M versus H curves of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at $T = 5$ K, 25 K, 60 K, and 100 K. (b) Comparison of the M versus H curves of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at $T = 25$ K (blue triangles) and $\text{Dy}_2\text{CuMnMn}_4\text{O}_{12}$ at $T = 5$ K (black circles) – the curves were qualitatively similar.

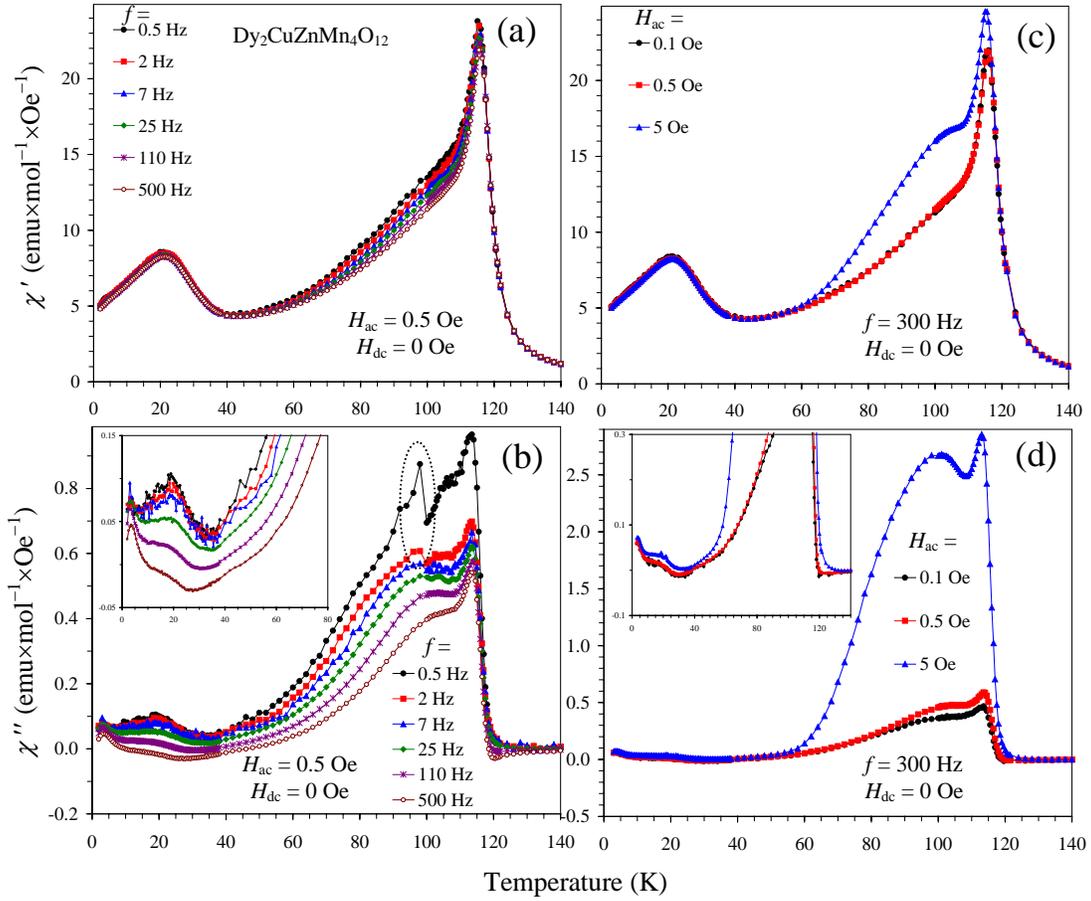


Figure S6. (a) The χ' versus T and (b) χ'' versus T curves of $\text{Dy}_2\text{CuZnMn}_4\text{O}_{12}$ at different frequencies measured with $H_{\text{ac}} = 0.5$ Oe and $H_{\text{dc}} = 0$ Oe. (c) The χ' versus T and (d) χ'' versus T curves at different $H_{\text{ac}} = 0.1, 0.5,$ and 5 Oe and one frequency of 300 Hz ($H_{\text{dc}} = 0$ Oe). Insets show zoomed parts of the corresponding figures. The dotted oval on panel (b) emphasizes (most probably) measurement artifacts because the measurement steps changed exactly at this temperature range.