

1 **Ca<sub>2</sub>CuWO<sub>6</sub>: Triclinically-distorted double perovskite with low-**  
2 **dimensional magnetic behavior**

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15

16 **Abstract**

17

18  $A_2CuB'O_6$  double perovskites with  $A = Sr$  and  $Ba$  and  $B' = W, Te,$  and  $Mo$  have received  
19 a lot of attention in the literature as material candidates for the spin-liquid ground state  
20 and frustrated spin lattices. However,  $Ca$  analogues have not been reported yet. In this  
21 work, the  $Ca_2CuWO_6$  double perovskite was synthesized by a high-pressure, high-  
22 temperature method at 6 GPa and between 1370 and 2020 K. The crystal structure was  
23 studied by synchrotron powder X-ray diffraction between 100 K and 1000 K.  $Ca_2CuWO_6$   
24 crystallizes in the triclinic space group  $P\bar{1}$  between 100 K and 1000 K with the onset of  
25 partial decomposition above 750 K. Room-temperature lattice parameters are  $a = 5.68435$   
26  $\text{\AA}$ ,  $b = 7.51261 \text{\AA}$ ,  $c = 5.49548 \text{\AA}$ ,  $\alpha = 90.0703^\circ$ ,  $\beta = 94.7634^\circ$ , and  $\gamma = 90.2311^\circ$ .  
27  $Ca_2CuWO_6$  demonstrates low-dimensional magnetic properties with the broad maximum  
28 of the magnetic susceptibility around 60 K caused by the dominant second-neighbor  
29 interactions within the deformed square planes of the  $Cu^{2+}$  ions. A long-range  
30 antiferromagnetic order takes place at  $T_N = 32$  K. A magnetic field-induced transition was  
31 observed at 28 kOe (at 2 K and 5 K), 29 kOe (at 10 K), 30 kOe (at 15 K), 32 kOe (at 20  
32 K), and 36 kOe (at 25 K). Solid solutions  $Ca_{2-x}Sr_xCuWO_6$  with  $x = 0.5, 1,$  and  $1.5$  were  
33 also prepared. The  $x = 0.5$  and  $1$  samples crystallize in the space group  $P\bar{1}$  at room  
34 temperature, while the  $x = 1.5$  sample has monoclinic symmetry (space group  $P2_1/n$ ).  
35 They also show low-dimensional magnetic behavior,  $T_N = 32\text{--}34$  K, and magnetic field-  
36 induced transitions.

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38

39 **Keywords:** double perovskites; low-dimensional magnets; antisite disorder; crystal  
40 structures; high-pressure synthesis

41

## 42 1. Introduction

43 B-site-ordered double perovskites with the stoichiometric formula of  $A_2BB'O_6$  have  
44 attracted extensive attention [1] owing to their novel magnetic and electrical properties,  
45 such as colossal magnetoresistance [2], superconductivity [3], and half-metallic behaviors  
46 [4]. Among them, double perovskites with  $A = Ca^{2+}$ ,  $Sr^{2+}$ , and  $Ba^{2+}$ ,  $B =$  divalent  $3d$   
47 magnetic or nonmagnetic transition metal cations ( $M$ ), and  $B' = W^{6+}$  ( $A_2MWO_6$ ) have  
48 been extensively studied, mainly due to their diversified structural and magnetic  
49 properties [5-7]. Such double perovskites usually have full/complete rock-salt ordered  
50 structures because of the large difference in the oxidation states and sizes of the  $M^{2+}$  and  
51  $W^{6+}$  cations [8]. Because magnetic  $M^{2+}$  cations are separated by non-magnetic  $W^{6+}$   
52 cations, such perovskites can form a frustrated square lattice of spins [9]. Their magnetism  
53 arises from super-superexchange interactions,  $M-O-(W)-O-M$ , between disconnected  
54  $MO_6$  octahedra. Typically, the  $M/WO_6$  octahedra undergo distortions and/or tilts in  
55 multiple directions to accommodate size mismatches caused by the substituted cations at  
56 the A and/or  $M$  sites.

57 A lot of different  $A_2MWO_6$  double perovskites have been synthesized, and their  
58 structure types are sensitive to chemical substitutions (summarized in Ref. [10], **TABLES**  
59 **SI and SII**, see also references [11-14] therein). The smaller ionic radius of  $Ca^{2+}$  (1.34 Å,  
60 compared to 1.44 Å for  $Sr^{2+}$  and 1.61 Å for  $Ba^{2+}$ , with the coordination number of 12 [15])  
61 results in a considerable tilting of the alternating  $MO_6$  ( $M = Mg, Ca, Mn, Fe, Co,$  and  $Ni$ )  
62 and  $WO_6$  octahedra in a monoclinic structure with the space group  $P2_1/n$  [16-20],  
63 although in the case of  $Ca_2ZnWO_6$  the space group  $Fm-3m$  was also reported [21]. The  
64 majority of the  $Sr_2MWO_6$  compounds adopt a tetragonal  $I4/m$  structure ( $M = Mg, Fe, Co,$   
65  $Ni,$  and  $Cu$ ) [22-25], though when  $M = Ca, Mn, Zn,$  or  $Cd$ , the structure changes to the  
66  $P2_1/n$  symmetry, similar to the Ca-containing compounds [25-29]. For  $Ba_2MWO_6$ , the  
67 cubic  $Fm-3m$  space group is observed for  $M = Mg, Mn, Fe, Co, Ni, Zn,$  and  $Cd$  [17,22,30-  
68 34], while  $Ba_2CaWO_6$  and  $Ba_2CuWO_6$  adopt an  $I4/m$  structure [5,35]. Furthermore, it was  
69 reported that the symmetry of  $Sr_2FeWO_6$  and  $Ba_2FeWO_6$  can depend on different  
70 synthesis procedures [6].

71 Structural phase transitions in the  $A_2MWO_6$  double perovskites have been extensively  
72 studied, particularly in the  $A = Sr$  compounds (see **TABLES SI and SII** [10]). These

73 compounds typically undergo phase transitions from low-symmetry space groups to high-  
74 symmetry space groups with increasing temperature, involving partial or complete  
75 transformations among the monoclinic, tetragonal, and cubic crystal systems. For  
76 example, in the cuprate double perovskite  $\text{Sr}_2\text{CuWO}_6$ , the Cu–O–W bond angles within  
77 the (001) planes are nonlinear indicating the octahedral tilting, and the presence of the  
78 Jahn–Teller (JT) active ions  $\text{Cu}^{2+}$  leads to significant distortions of the  $\text{CuO}_6$  octahedra  
79 [36]. With increasing temperature, the Cu–O–W bond angle of  $\text{Sr}_2\text{CuWO}_6$  changes to  
80  $180^\circ$  (at around 870 K), and the  $\text{CuO}_6$  octahedra are elongated along the fourfold axis.  
81 The room-temperature  $I4/m$  structure undergoes a transition to a higher-symmetry  
82 structure,  $I4/mmm$ . At approximately 1190 K, the long-range cooperative JT distortion  
83 disappears, leading to a first-order transition from  $I4/mmm$  to  $Fm\bar{3}m$  [25,37].

84  $\text{Sr}_2\text{CuWO}_6$  and  $\text{Ba}_2\text{CuWO}_6$  have also been intensively investigated for their interesting  
85 magnetic properties [5,9,38-40]. It was found that these two double perovskites exhibit  
86 low-dimensional antiferromagnetic (AFM) behavior which was associated with the  
87 cooperative JT effect of  $\text{Cu}^{2+}$  and the super-supereexchange interaction between  $\text{Cu}^{2+}$   
88 cations via an array of  $\text{W}^{6+}$  cations. The JT effect results in the orbital ordering and  
89 alignment of the short Cu- $O_{ab}$  bonds in the (001) planes and long Cu- $O_c$  bonds along the  
90  $c$ -axis. The half-filling of the  $3d_{x^2-y^2}$  orbitals facilitates strong magnetic interactions  
91 within the (001) planes, whereas magnetic interactions along the  $c$ -axis are much weaker.  
92 In alternating  $\text{CuO}_6$  and  $\text{WO}_6$  octahedra, the AFM ordering comes from the  $180^\circ$  Cu–O–  
93 W–O–Cu super-supereexchange interactions within the (001) planes and interplanar  
94 interactions along the  $c$ -axis [38,40]. In addition,  $\text{Sr}_2\text{CuWO}_6$  has recently been  
95 investigated as a catalytic material [41,42]. The solid solutions of  $\text{Sr}_2\text{CuWO}_6$  and  
96 isostructural  $\text{Sr}_2\text{CuTeO}_6$  develop a complex magnetically disordered quantum state  
97 [43,44] driven by quenched randomness [45,46].

98 Although all other members of  $\text{Ca}_2M\text{WO}_6$  with  $M = \text{Mg, Ca, Mn, Fe, Co, Ni, and Zn}$ ,  
99  $\text{Sr}_2M\text{WO}_6$  and  $\text{Ba}_2M\text{WO}_6$  with  $M = \text{Mg, Ca, Mn, Fe, Co, Ni, Cu, and Zn}$  have been  
100 successfully synthesized,  $\text{Ca}_2\text{CuWO}_6$  remains a missing compound in the  $A_2M\text{WO}_6$   
101 family. In this work, the  $\text{Ca}_2\text{CuWO}_6$  double perovskite was successfully synthesized by  
102 a high-pressure, high-temperature method. It has a highly distorted structure, which is  
103 different from other members of the family. It crystallizes in a triclinic space group  $P\bar{1}$

104 between 100 K and 1000 K and shows the onset of partial decomposition above 750 K.  
105 It demonstrates low-dimensional magnetic properties with a broad maximum of the  
106 magnetic susceptibility around 60 K caused by the dominant second-neighbor interactions  
107 within the deformed square planes of the  $\text{Cu}^{2+}$  ions. A long-range antiferromagnetic order  
108 takes place at  $T_N = 32$  K. The solid solutions  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  with  $x = 0.5, 1,$  and  $1.5$   
109 were also prepared. The  $x = 0.5$  and  $1$  samples crystallize in the space group  $P-1$  at room  
110 temperature, similar to the  $x = 0$  sample, whereas the  $x = 1.5$  sample crystallizes in the  
111 space group  $P2_1/n$ . The  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples also show low-dimensional magnetic  
112 behavior,  $T_N = 32-34$  K, and magnetic field-induced transitions.

113

## 114 **2. Experimental**

115  $\text{Ca}_2\text{CuWO}_6$  and  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  were prepared under high-pressure and high-  
116 temperature conditions using a belt-type high-pressure machine.  $\text{Ca}_2\text{CuWO}_6$  was  
117 prepared from a stoichiometric mixture of  $\text{Ca}_3\text{WO}_6$ ,  $\text{CuO}$ , and  $\text{WO}_3$  at 6 GPa and about  
118 1370 K and 1570 K for 2 h in Au capsules. We also prepared  $\text{Ca}_2\text{CuWO}_6$  at 6 GPa and  
119 about 2020 K for 0.5 h in Pt capsules.  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  with  $x = 0.5, 1, 1.5,$  and  $2$  were  
120 prepared from stoichiometric mixtures of  $\text{Ca}_3\text{WO}_6$ ,  $\text{Sr}_2\text{WO}_5$ ,  $\text{CuO}$ , and  $\text{WO}_3$  at 6 GPa and  
121 about 1370 K for 2 h in Au capsules. After annealing at the target temperature, the samples  
122 were cooled down to room temperature (RT) by turning off the heating current, and the  
123 pressure was slowly released. All the samples had brownish yellow color (**TABLE SIII**  
124 [10]).  $\text{Ca}_3\text{WO}_6$  was prepared from a stoichiometric mixture of  $\text{CaCO}_3$  and  $\text{WO}_3$  by  
125 annealing in air at 1270 K for 96 h with several intermediate grindings.  $\text{Sr}_2\text{WO}_5$  was  
126 prepared by the same method as  $\text{Ca}_3\text{WO}_6$  but from a stoichiometric mixture of  $\text{SrCO}_3$  and  
127  $\text{WO}_3$ .

128 Laboratory X-ray powder diffraction (XRPD) data were collected at RT on a RIGAKU  
129 MiniFlex600 diffractometer using  $\text{CuK}\alpha$  radiation (in a  $2\theta$  range of  $5-100^\circ$  with a step  
130 of  $0.02^\circ$ , and scan speed of  $3^\circ/\text{min}$ ). Synchrotron XRPD data for  $\text{Ca}_2\text{CuWO}_6$  were  
131 collected on the beamline BL02B2 of SPring-8 from 100 K to 900 K on heating in one  
132 run and from 600 K to 1000 K on heating in another run (using a fresh sample) [47]. The  
133 data were taken between  $4.002^\circ$  and  $71.25^\circ$  at  $0.006^\circ$  intervals in  $2\theta$  using a wavelength  
134 of  $\lambda = 0.61974$  Å. The data with good statistics for structural refinements were collected

135 at 100 K and 295 K (for 400 s); the measurement time was 10 s at all other temperatures.  
136 Synchrotron XRPD data of the  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples ( $x = 0.5, 1.0, \text{ and } 1.5$ ) were  
137 measured at RT with the measurement time of 100 s. The samples were filled into open  
138 Lindemann glass (or quartz, for measurements up to 1000 K) capillary tubes (inner  
139 diameter: 0.2 mm), which were rotated during the measurements. The Rietveld analysis  
140 of all XRPD data was performed using the *RIETAN-2000* program [48].

141 Magnetic measurements were performed on a SQUID magnetometer (Quantum  
142 Design, MPMS3) between 2 and 300 K in an applied field of 100 Oe and 10 kOe under  
143 both zero-field-cooled (ZFC) and field-cooled on cooling (FCC) conditions. Isothermal  
144 magnetization measurements were performed at different temperatures between  $-70$  and  
145 70 kOe. No corrections were applied to raw magnetic data. Samples weights were also  
146 not corrected (to account for non-magnetic impurities). Specific heat was measured on a  
147 Quantum Design PPMS-9T instrument on cooling at magnetic fields of 0 Oe and 90 kOe.

148 Dielectric properties of  $\text{Ca}_2\text{CuWO}_6$  were measured on a NOVOCONTROL Alpha-A  
149 High Performance Frequency Analyzer on cooling in the temperature range between 3.5  
150 K and 300 K and the frequency range between 301 Hz and 665 kHz at zero magnetic  
151 field.

152 Differential scanning calorimetry (DSC) curves of the  $\text{Ca}_2\text{CuWO}_6$  powder sample  
153 (33.29 mg) were recorded on a Mettler Toledo DSC1 STAR<sup>c</sup> system between 300 K and  
154 870 K in an open Al capsule with the heating/cooling rate of 10 K/min under  $\text{N}_2$  flow.  
155 Two DSC runs were performed to check the reproducibility. DSC measurements were  
156 also performed for the  $\text{Ca}_2\text{CuWO}_6$  powder sample (17.47 mg) between 300 K and 970 K  
157 in an open Pt capsule with the heating/cooling rate of 10 K/min under  $\text{N}_2$  flow. Laboratory  
158 XRPD data were collected after the DSC experiments to check phase compositions after  
159 heating.

160 Magnetic exchange couplings were evaluated by density-functional-theory (DFT)  
161 band-structure calculations performed in the FPLO code [49] using the Perdew-Burke-  
162 Ernzerhof (PBE) flavor of the exchange-correlation potential [50] and the experimental  
163 crystal structure of  $\text{Ca}_2\text{CuWO}_6$ . Correlation effects in the Cu  $3d$  shell were taken into  
164 account on the mean-field DFT+ $U$  level with the on-site Coulomb repulsion potential  $U_d$   
165  $= 8.5$  eV, Hund's coupling  $J_d = 1$  eV, and double-counting correction in the atomic limit

166 [51,52]. Exchange couplings were obtained from the superexchange model and,  
167 concurrently, from total energies of magnetically ordered states using the mapping  
168 procedure [53,54]. Magnetic susceptibility of the square-lattice antiferromagnet was  
169 obtained from quantum Monte-Carlo (QMC) simulations using the *loop* algorithm [55]  
170 of the ALPS package [56].

171

### 172 **3. Results and discussion**

#### 173 **3.1. Crystal structure and high-temperature behavior**

174  $\text{Ca}_2\text{CuWO}_6$  perovskite could be prepared at 6 GPa at different temperatures between  
175 1370 and 2020 K. However, the amounts of the  $\text{CaWO}_4$  impurity were different (**FIG. S1**  
176 **and TABLE SIV** [10]). For a detailed characterization, we selected a sample prepared at  
177 1570 K with the minimum amount of the  $\text{CaWO}_4$  impurity. The RT crystal structure of  
178  $\text{Ca}_2\text{CuWO}_6$  was determined from the synchrotron XRPD data. The Rietveld refinement  
179 (**FIG. 1**) confirmed that  $\text{Ca}_2\text{CuWO}_6$  crystallizes in the triclinic space group  $P\bar{1}$  which  
180 has never been observed in the other members of the  $A_2M\text{WO}_6$  family. **TABLE I** lists the  
181 structure parameters and suggests that all sites in the double perovskite structure are fully  
182 occupied. The low symmetry of the structure gives rise to four independent sites for the  
183 B cations, specifically designated as Cu1(1*e*), Cu2(1*b*), W1(1*d*), and W2 (1*g*). The crystal  
184 structure at 100 K was also refined, and the structural parameters are presented in **TABLE**  
185 **SV** (the Rietveld refinement fits are shown on **FIG. S2**) [10].

186 As shown in **FIG. 2a**, the low-symmetry crystal structure consists of alternating  
187 corner-shared  $\text{CuO}_6$  and  $\text{WO}_6$  octahedra in three crystallographic directions. The Ca  
188 atoms (Ca1 and Ca2) occupy the interstitial spaces between the octahedra. The two  
189 nonequivalent octahedra for both Cu and W sites gave two different arrangements of the  
190  $\text{CuO}_6$  (Cu1O<sub>6</sub> and Cu2O<sub>6</sub>) and  $\text{WO}_6$  (W1O<sub>6</sub> and W2O<sub>6</sub>) octahedra resulting in the two  
191 different types of alternating layers along the *b* axis. One layer consists of the Cu2O<sub>6</sub> and  
192 W1O<sub>6</sub> octahedra (**FIG. 2b**), and another layer consists of the Cu1O<sub>6</sub> and W2O<sub>6</sub> octahedra  
193 (**FIG. 2c**). The bond lengths, bond angles, octahedral distortion parameter ( $\Delta$ ), and bond  
194 valence sums (BVS) [57] for  $\text{Ca}_2\text{CuWO}_6$  at RT are summarized in **TABLE II**. The BVS  
195 values for both the Ca sites and Cu sites were close to the expected values of +2, while  
196 the values for the W1 and W2 sites were slightly higher than +6. The Cu1-O6 bonds in  
197 the Cu1O<sub>6</sub> octahedra and the Cu2-O5 bonds in the Cu2O<sub>6</sub> octahedra ( $\sim 2.4$  Å) are

198 significantly longer than the other Cu-O bonds ( $\sim 2.0$  Å) showing that the CuO<sub>6</sub> octahedra  
199 are elongated along the (-101) direction. On the other hand, the WO<sub>6</sub> octahedra are  
200 slightly compressed along (-101). The highly distorted CuO<sub>6</sub> octahedra with the sizable  
201 octahedral distortion parameter ( $\Delta(\text{CuO}_6)$ ) can be ascribed to the JT effect of the Cu<sup>2+</sup>  
202 ions. The Cu-O-W angles between the CuO<sub>6</sub> and WO<sub>6</sub> octahedra in Ca<sub>2</sub>CuWO<sub>6</sub> are  
203 smaller than those of Sr<sub>2</sub>CuWO<sub>6</sub> and Ba<sub>2</sub>CuWO<sub>6</sub> [5] as expected because the ion size of  
204 Ca<sup>2+</sup> is smaller than Sr<sup>2+</sup> and Ba<sup>2+</sup> (1.34 Å for Ca<sup>2+</sup>, 1.44 Å for Sr<sup>2+</sup>, and 1.61 Å for Ba<sup>2+</sup>,  
205 with a coordination number of 12) [15]. Therefore, the combination of the largest tilts  
206 (among A<sub>2</sub>CuWO<sub>6</sub>) and strong JT effects could result in a strong structural distortion and  
207 triclinic symmetry.

208 We note that the crystal structure of Ca<sub>2</sub>CuWO<sub>6</sub> was predicted from the first-principles  
209 calculations in the framework of the Materials Project, where two models were given in  
210 space group *Cm* [58] and *P2<sub>1</sub>/c* [59]. However, the experimental structure does not match  
211 with these two predictions (**FIGs. S3** [10]).

212 **FIG. 3** depicts the temperature dependence of the unit-cell parameters and volume  
213 based on the in situ synchrotron XRPD measurements. The *a* lattice parameter first  
214 decreases slightly from 100 to 225 K, then gradually increases from 225 to 450 K, and  
215 starts to decline when the temperature exceeds 450 K. However, both *b* and *c* lattice  
216 parameters monotonically increase on heating from 100 to 900 K. Therefore, Ca<sub>2</sub>CuWO<sub>6</sub>  
217 demonstrates anisotropic thermal expansion above 450 K. The values of  $\alpha$  and  $\gamma$  exhibit  
218 a monotonically decreasing trend with increasing temperature, while the value of  $\beta$   
219 initially shows a slight increase before also following a consistent downward trend (**FIG.**  
220 **3c**). Concurrently, the unit-cell volume (*V*) demonstrates a monotonically increasing trend  
221 without any anomalies (**FIG. 3d**) with the volumetric coefficient of thermal expansion of  
222  $3.61(14) \times 10^{-5} \text{ K}^{-1}$  between 500 K and 850 K. As illustrated in **FIGs. S4a-S4c** [10], the  
223 diffraction peak positions shift slightly towards lower angles across the entire temperature  
224 range due to heating, which can be attributed to thermal expansion effects. The refined  
225 weight fraction of the CaWO<sub>4</sub> impurity started increasing above 750 K along with the  
226 broadening of the corresponding reflections, and very broad new diffraction peaks  
227 appeared, which can be assigned to Ca<sub>3</sub>WO<sub>6</sub> impurity. Above 850 K, diffraction peaks  
228 from the CuO impurity could be detected as well. Therefore, the Ca<sub>2</sub>CuWO<sub>6</sub> double

229 perovskite partially decomposes above about 750 K. Despite this partial decomposition,  
 230 all lattice parameters continue to change monotonically up to 900 K. Compared to the as-  
 231 synthesized, original sample, the diffraction patterns (measured at RT) of the samples  
 232 obtained after in-situ synchrotron XRPD heating experiments to 900 K and 1000 K  
 233 exhibit significant changes (**FIG. S5** [10]). The CaWO<sub>4</sub> impurity content in the sample  
 234 heated to 900 K increased from 3.3 wt% to 8.6 wt% (**FIG. S6a** [10]), and small amounts  
 235 of Ca<sub>3</sub>WO<sub>6</sub> and CuO could be detected. While in the sample heated to 1000 K, the CaWO<sub>4</sub>  
 236 content rose to 24.9 wt%, the sample also contained 40.5 wt% Ca<sub>3</sub>WO<sub>6</sub>, and broad peaks  
 237 from Cu<sub>2</sub>O could be identified (**FIG. S6b** [10]). No noticeable anomalies were observed  
 238 in the DSC curves. Therefore, the DSC measurements and high-temperature in situ  
 239 synchrotron XRPD studies suggest that Ca<sub>2</sub>CuWO<sub>6</sub> double perovskite (partly)  
 240 decomposes above 750 K and does not undergo any structural phase transitions between  
 241 100 K and 750 K, thus preserving its triclinic symmetry in a broad temperature range.

242

### 243 **3.2. Magnetic susceptibility**

244 The temperature dependence of the magnetic susceptibilities for Ca<sub>2</sub>CuWO<sub>6</sub> measured  
 245 under applied magnetic fields of 10 kOe and 100 Oe is illustrated in **FIG. 4** and **FIG. S7**  
 246 [10], respectively. As shown in **FIG. 4**, a broad maximum of the magnetic susceptibility  
 247 is observed at around 60 K originating from low-dimensional magnetism – a  
 248 characteristic feature commonly observed in the A<sub>2</sub>CuWO<sub>6</sub> double perovskites. The  
 249 temperature at which these maxima occur serves as an indicator of the magnitude of the  
 250 predominant AFM exchange interactions. With decreasing temperature, a long-range  
 251 AFM order takes place below  $T_N$  (Néel temperature) = 32 K, as evidenced by peaks in  
 252 the ZFC and FCC  $d\chi/dT$  versus  $T$  curves (inset of **FIG. 4** and **FIG. S7** [10]). The  $T_N$  value  
 253 for Ca<sub>2</sub>CuWO<sub>6</sub> was slightly higher than the  $T_N$  values of 28 K and 24 K reported for  
 254 Ba<sub>2</sub>CuWO<sub>6</sub> [60] and Sr<sub>2</sub>CuWO<sub>6</sub> [39], respectively.

255 The inverse magnetic susceptibilities exhibit the behavior consistent with the Curie–  
 256 Weiss law at elevated temperatures above approximately 200 K. The FCC data, measured  
 257 under an applied magnetic field of 10 kOe, were accurately fitted to the following Curie–  
 258 Weiss equation within the temperature range of 200 K to 295 K:

259

$$\chi(T) = \mu_{\text{eff}}^2 N(3k_B(T - \theta))^{-1}$$

260 Here  $\mu_{\text{eff}}$  is the effective magnetic moment,  $N$  is Avogadro's number,  $k_{\text{B}}$  is Boltzmann's  
 261 constant, and  $\theta$  is the Curie–Weiss temperature. The observed experimentally determined  
 262 effective magnetic moment,  $\mu_{\text{eff}} = 1.776(3) \mu_{\text{B}}$ , aligns closely with the spin-only  
 263 theoretical value,  $\mu_{\text{cal}} = g\sqrt{S(S+1)} \approx 1.732 \mu_{\text{B}}$ , calculated for the  $\text{Cu}^{2+}$  ions, where  $S =$   
 264  $\frac{1}{2}$  and  $g = 2$ . This agreement confirms that  $\text{Cu}^{2+}$  is the sole contributor to the magnetic  
 265 properties of  $\text{Ca}_2\text{CuWO}_6$ . The  $\mu_{\text{eff}}$  value is comparable to those of  $\text{Ba}_2\text{CuWO}_6$  [61] and  
 266  $\text{Sr}_2\text{CuWO}_6$  [5,40]. The negative Weiss temperature ( $\theta = -88.6(12)$  K) confirms the  
 267 dominant AFM interactions (**TABLE III**).

268 The isothermal magnetization curve ( $M$  versus  $H$ ) for  $\text{Ca}_2\text{CuWO}_6$  at  $T = 5$  K is  
 269 illustrated in **FIG. S8** [10]. There was almost no hysteresis near the origin (zero magnetic  
 270 field) suggesting a purely AFM state. Furthermore, as shown in **FIG. 5 and Table SVI**  
 271 [10], the  $M$  versus  $H$  curves at different temperatures indicate a magnetic field-induced  
 272 transition at 28 kOe (at 2 K and 5 K), 29 kOe (at 10 K), 30 kOe (at 15 K), 32 kOe (at 20  
 273 K), and 36 kOe (at 25 K).

274 While the low-symmetry crystal structure of  $\text{Ca}_2\text{CuWO}_6$  allows for multiple  
 275 nonequivalent superexchange pathways, our DFT calculations (**Sec. 3.5**) show that the  
 276 dominant interactions connect second-neighbor  $\text{Cu}^{2+}$  ions, which build deformed square  
 277 lattices associated with the Cu1 and Cu2 sites, respectively (**Table IV**). These  
 278 interpenetrating square lattices are connected by the weaker nearest-neighbor couplings  
 279 (**FIG. 8**) that lead to a moderate frustration akin to the  $J_1$ - $J_2$  square lattice model [62].  
 280 The experimental magnetic susceptibility of  $\text{Ca}_2\text{CuWO}_6$  can be then described using the  
 281 high-temperature series expansion (HTSE) for the  $J_1$ - $J_2$  square lattice model [63]. The  
 282 respective fit extends down to 80 K and returns  $J_1 = 18$  K,  $J_2 = 65$  K ( $g = 2.10$ ,  $\chi_0 =$   
 283  $-5.3 \times 10^{-5}$  emu/mol), see **FIG. 4**. Alternatively, one could neglect the weaker nearest-  
 284 neighbor couplings and use QMC simulations for the nonfrustrated square-lattice model  
 285 with  $J = 63$  K ( $g = 1.90$ ,  $\chi_0 = 8.5 \times 10^{-5}$  emu/mol) that satisfactorily describes the data  
 286 down to  $T_{\text{N}}$ , albeit showing a slightly different curvature at 100–200 K (**FIG. 4**). The  
 287 resulting  $J$  value should be seen as an effective coupling within the deformed square  
 288 lattices of Cu1 and Cu2.

289

### 290 **3.3. Heat capacity**

291 Specific heat data plotted as  $C_p/T$  versus  $T$  are shown in **FIG. 6**. A very weak and sharp  
 292 anomaly was observed at  $T_N = 32$  K at  $H = 0$  Oe confirming the transition toward a long-  
 293 range AFM ordered state. The fact that this transition is almost unaffected by the magnetic  
 294 field of 90 kOe (inset of **FIG. 6**) gives additional evidence that  $\text{Ca}_2\text{CuWO}_6$  is a robust  
 295 antiferromagnet. In magnetic insulators, the specific heat capacity is primarily governed  
 296 by lattice contributions at higher temperatures and magnetic contributions at lower  
 297 temperatures. The lattice contribution was estimated using the experimental specific heat  
 298 of the  $\text{Ca}_2\text{CaWO}_6$  double perovskite without magnetic cations (between 2 and 46 K) and  
 299 the experimental specific heat of the  $\text{Ca}_2\text{MnWO}_6$  double perovskite (between 48 and 80  
 300 K) as described in Ref. [64]. We calculated the magnetic entropy,  $S_m$ , by integrating the  
 301 difference between  $C_p$  and the lattice contribution. The expected magnetic entropy for  
 302 spin- $1/2$  is  $S_m = R \ln 2 \approx 5.76 \text{ J K}^{-1} \text{ mol}^{-1}$ , where  $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$  is the universal gas  
 303 constant. The calculated  $S_m$  of  $\text{Ca}_2\text{CuWO}_6$  at  $T = 80$  K is  $2.8 \text{ J K}^{-1} \text{ mol}^{-1}$ , about half of  
 304 the expected value. The reduced value of  $S_m$  in  $\text{Ca}_2\text{CuWO}_6$  could be caused by the  
 305 overestimation of the lattice contribution. However, the precise estimation of the lattice  
 306 contribution is difficult as  $\text{Ca}_2\text{CuWO}_6$  crystallizes in a different space group and in a  
 307 highly distorted structure in comparison with other members of the family. Moreover,  
 308 with the dominant exchange coupling of 63 K, a partial release of the magnetic entropy  
 309 also above 80 K can be expected. The magnetic entropy below  $T_N$  was about  $1.2 \text{ J K}^{-1}$   
 310  $\text{mol}^{-1}$ . In other words, more than half of the entropy (estimated in a way described above)  
 311 is released above  $T_N$ , thus corroborating the low-dimensional nature of the  $\text{Ca}_2\text{CuWO}_6$   
 312 magnetism.

313

### 314 **3.4. Dielectric properties**

315 **FIG. 7** shows the temperature dependence of the dielectric constant, which was nearly  
 316 frequency-independent between 3.5 and 100 K. No anomalies, kinks, or peaks were  
 317 observed in the dielectric constant at  $T_N = 32$  K. The high dielectric constant in the low-  
 318 frequency region at higher temperatures was primarily caused by a Maxwell-Wagner  
 319 contribution. Therefore, no magnetodielectric effects were detected in the  $\text{Ca}_2\text{CuWO}_6$   
 320 double perovskite in comparison with the  $(\text{Ca}_{2-x}\text{Mn}_x)\text{MnWO}_6$  and  $\text{Ca}_2\text{MnWO}_6$  double  
 321 perovskites [64].

322

### 323 3.5. Magnetic model

324 Triclinic symmetry of  $\text{Ca}_2\text{CuWO}_6$  with two nonequivalent Cu sites renders the  
325 magnetic model of this material more complex compared to  $\text{Sr}_2\text{CuWO}_6$ . Therefore, we  
326 used DFT calculations to assess individual exchange couplings via two complementary  
327 methods, superexchange theory and mapping analysis. **FIG. 8** shows the metallic band  
328 structure of  $\text{Ca}_2\text{CuWO}_6$ , which is calculated on the PBE level and does not include  
329 correlations that are responsible for opening the band gap. This band structure can be used  
330 for identifying relevant electronic states of the material. The Fermi level is crossed by  
331 two bands with the predominant Cu  $d_{x^2-y^2}$  character that corresponds to the highest-  
332 energy crystal-field level of  $\text{Cu}^{2+}$  in the distorted octahedral coordination. Each of the two  
333 bands is formed by Cu1 and Cu2, respectively, with only a weak hybridization. This  
334 separation of the orbital characters indicates that the magnetic sublattices of Cu1 and Cu2  
335 are largely decoupled. Indeed, the hopping parameters extracted from the Wannier fit [65]  
336 of the two  $d_{x^2-y^2}$  bands reveal stronger couplings within each sublattice and much  
337 weaker couplings between the sublattices (**Table IV**). Each sublattice represents a  
338 deformed square lattice (rectangular lattice) lying in the (10-1) plane of the crystal  
339 structure. Whereas the Cu1 sublattice is almost regular, the Cu2 sublattice shows a sizable  
340 difference between the couplings along the [010] and [101] directions within the plane.  
341 The leading interplane hopping is about 5 meV, at least one order of magnitude smaller  
342 than the in-plane hoppings. Therefore,  $\text{Ca}_2\text{CuWO}_6$  should be understood as a quasi-2D  
343 antiferromagnet, in agreement with the experiment.

344 Our mapping analysis confirms these results. The Cu1 and Cu2 sites form  
345 interpenetrating rectangular lattices with the dominant interactions  $J_{21a}$ ,  $J_{21b}$ ,  $J_{22a}$ , and  $J_{22b}$ .  
346 These rectangular lattices are connected by the weaker frustrated couplings  $J_{1a}$  and  $J_{1b}$   
347 (**Table IV**). Overall, the magnetic model of  $\text{Ca}_2\text{CuWO}_6$  can be interpreted as a distorted  
348 version of the  $J_1$ - $J_2$  frustrated square lattice. Previous studies showed that such  
349 deformations almost do not influence the magnetic susceptibility above  $T_N$  [66].  
350 Therefore, the fitted values of  $J_1$  and  $J_2$  should be seen as the averaged nearest-neighbor  
351 and second-neighbor couplings, respectively:  $J_1 = (J_{1a}+J_{1b})/2$  and  $J_2 =$   
352  $(J_{21a}+J_{21b}+J_{22a}+J_{22b})/4$ . From the DFT results (**Table IV**), one finds  $J_1 = 9$  K and  $J_2 = 59$   
353 K in a good agreement with the fitted values of 18 K and 65 K, respectively.

354 Despite its triclinic distortion,  $\text{Ca}_2\text{CuWO}_6$  features the same mechanism of the  
355 exchange couplings as the other  $\text{Cu}^{2+}$  double perovskites. The second-neighbor couplings  
356 are amplified by the low-lying  $d$ -states of  $\text{W}^{6+}$ , similar to  $\text{Sr}_2\text{CuWO}_6$  and at odds with  
357  $\text{Sr}_2\text{CuTeO}_6$  where the  $p$ -states of  $\text{Te}^{6+}$  facilitate nearest-neighbor couplings ( $J_1$ ) and  
358 suppress second-neighbor couplings ( $J_2$ ) [9,67].

359

### 360 **3.6. $\text{Sr}^{2+}$ doping**

361 Solid solutions  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  with  $x = 0.5, 1,$  and  $1.5$  were also prepared to study  
362 the doping effects in  $\text{Ca}_2\text{CuWO}_6$  and its evolution toward the extensively studied  
363  $\text{Sr}_2\text{CuWO}_6$  [5,25,36,38-40,68-71]. XRPD data for the  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples with  $x =$   
364  $0.5, 1,$  and  $1.5$  are illustrated in **FIG. S9** [10] and compared with  $\text{Ca}_2\text{CuWO}_6$  ( $x = 0$ ) and  
365  $\text{Sr}_2\text{CuWO}_6$  ( $x = 2$ ), prepared under similar synthesis conditions. With the increase in the  
366  $\text{Sr}^{2+}$  concentration ( $x$ ), the main impurity changes from  $\text{CaWO}_4$  (for  $x = 0, 0.5,$  and  $1$ ) to  
367 the high-pressure modification of  $\text{Sr}_3\text{W}_2\text{O}_9$  [72] (for  $x = 1.5$  and  $2$ ). Our analysis showed  
368 that the  $x = 0.5$  and  $1$  samples retained the space group  $P-1$ . However, the application of  
369 this triclinic model to the  $x = 1.5$  sample resulted in the following refined angles,  $\alpha =$   
370  $89.997(2)^\circ$  and  $\gamma = 89.998(2)^\circ$ , suggesting that the symmetry is higher. Therefore, the  
371 structure of the  $x = 1.5$  sample was refined in the space group  $P2_1/n$ , which is a direct  
372 supergroup of  $P-1$  with the same cell dimensions. (**FIGs. S10-S12 and TABLEs SVII -**  
373 **SX** [10]). We also note that only a few extremely weak reflections with  $h + k + l = 2n + 1$   
374 were observed experimentally in the  $x = 1.5$  sample. In other words, the structure of the  
375  $x = 1.5$  sample can also be well refined in the  $I2/m$  model (**FIG. S12 and TABLEs SVIII**  
376 **and SX** [10]). We further emphasize that the  $P2_1/n$  symmetry is usually assigned for  
377 monoclinic members of the  $A_2MWO_6$  family, but the  $P2_1/n$  model has a monoclinic angle  
378  $\beta$  between the  $\sqrt{2}a_p$  and  $2a_p$  lattice parameters (where  $a_p$  is the parameter of the cubic  
379 perovskite subcell). On the other hand, the  $P2_1/n$  model of the  $x = 1.5$  sample has its  
380 monoclinic angle  $\beta$  between the  $\sqrt{2}a_p$  and  $\sqrt{2}a_p$  lattice parameters. No DSC anomalies  
381 were found in the  $x = 1.5$  sample between  $300\text{ K}$  and  $870\text{ K}$  suggesting the absence of  
382 structural phase transitions. The compositional dependence of the cell parameters and  
383 volume obtained from the synchrotron XRPD data is presented in **FIG. S13** [10]. The  $b$   
384 and  $c$  parameters increased linearly, whereas the  $a$  parameter decreased gradually with

385 increasing  $x$ . Additionally, the  $\alpha$  and  $\gamma$  parameters decreased linearly with increasing  $x$ ,  
386 whereas the  $\beta$  parameter remained almost constant. The volume  $V$  increased linearly as  
387 expected following the incorporation of the larger  $\text{Sr}^{2+}$  ions. The effect of  $\text{Sr}^{2+}$  doping on  
388 the lattice parameters of  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  was similar to the effect of temperature on the  
389 lattice parameters of  $\text{Ca}_2\text{CuWO}_6$ .

390 The magnetic susceptibility and isothermal magnetization curves for the  
391  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples with  $x = 0.5, 1, \text{ and } 1.5$  are given on **FIGs. S14-S16** [10], and  
392 their Curie–Weiss fit parameters are summarized in **TABLE III**. They also showed the  
393 quasi-2D magnetic behavior and magnetic field-induced transitions (**FIGs. S17-S22** [10]).  
394 The  $T_N$  remained nearly the same in the  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples with  $x = 0.5$  ( $T_N = 32$   
395 K), 1 ( $T_N = 33$  K), and 1.5 ( $T_N = 34$  K) (**FIGs. S23-S24** [10]). This observation is  
396 somewhat unexpected, considering that the  $T_N$  of  $\text{Sr}_2\text{CuWO}_6$  ( $x = 2$ ) was reported to be  
397 24 K [39]. Therefore, the symmetry lowering in the  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples with  $0 \leq x$   
398  $\leq 1.5$  enhances  $T_N$ . The magnetic field corresponding to the field-induced transition at  
399 different temperatures also increases noticeably. For example, at 25 K, the corresponding  
400 magnetic field is 36 kOe for  $x = 0$ , and it increased to 42 kOe for  $x = 0.5$  and 49 kOe for  
401  $x = 1$ .

402 In simple  $\text{ABO}_3$  perovskites with direct B–O–B exchange interactions, magnetic phase  
403 transition temperatures are highly sensitive to the value of B–O–B angles and, therefore,  
404 to the A-cation size. Magnetic phase transition temperatures usually decrease with  
405 decreasing the size of the A cations [73]. However, in the  $\text{A}_2\text{MWO}_6$  double perovskites  
406 with the  $M\text{--O--(W)--O--M}$  super-superexchange interactions, the opposite tendency is  
407 observed wherein the more distorted samples with  $\text{A} = \text{Ca}$  have higher magnetic phase  
408 transition temperatures (**TABLE SI** [10]). Our results for  $\text{Ca}_2\text{CuWO}_6$  are in agreement  
409 with this tendency.

410

#### 411 **4. Conclusions**

412 In conclusion,  $\text{Ca}_2\text{CuWO}_6$  double perovskite – a missing member in the family of the  
413  $\text{A}_2\text{MWO}_6$  double perovskites – was successfully prepared by a high-pressure, high-  
414 temperature method.  $\text{Ca}_2\text{CuWO}_6$  has a highly distorted structure and different symmetry  
415 in comparison with other members of the family. It crystallizes in a triclinic space group

416  $P-1$  between 100 K and 1000 K; however, it shows partial decomposition above 750 K  
417 upon heating in air. Magnetic susceptibility data and density-functional calculations  
418 consistently show a quasi-2D antiferromagnetic behavior, with the deformed square  
419 lattice of the  $\text{Cu}^{2+}$  ions dominated by the second-neighbor exchange couplings of about  
420 60 K. A long-range antiferromagnetic order takes place at  $T_N = 32$  K, and a magnetic  
421 field-induced transition was observed below  $T_N$ . The solid solutions  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  with  
422  $x = 0.5$  and 1 also crystallize in the space group  $P-1$  at room temperature, while the  $x =$   
423 1.5 sample has a  $P2_1/n$  structure.  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  samples show similar quasi-2D  
424 magnetic behavior,  $T_N = 32-34$  K, and magnetic field-induced transitions.

425

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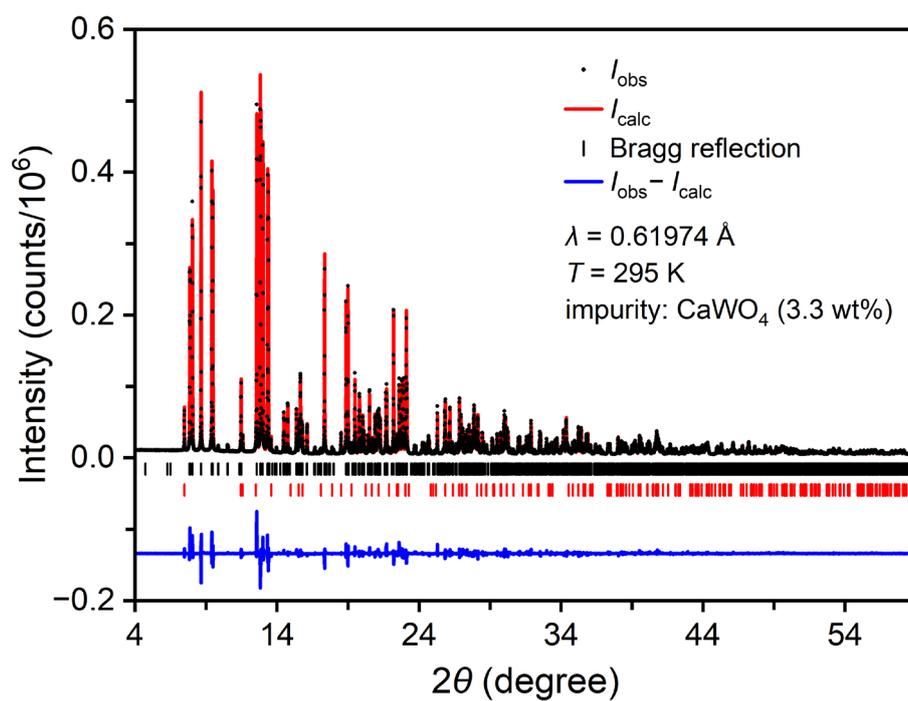
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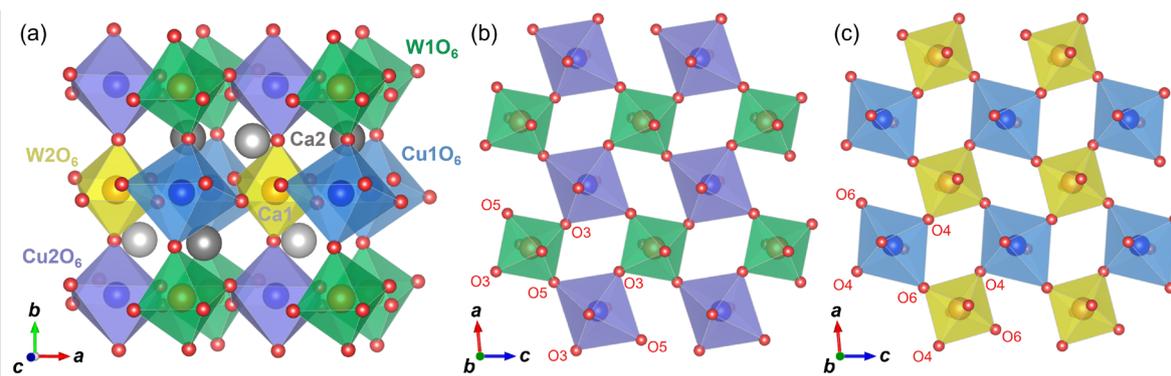
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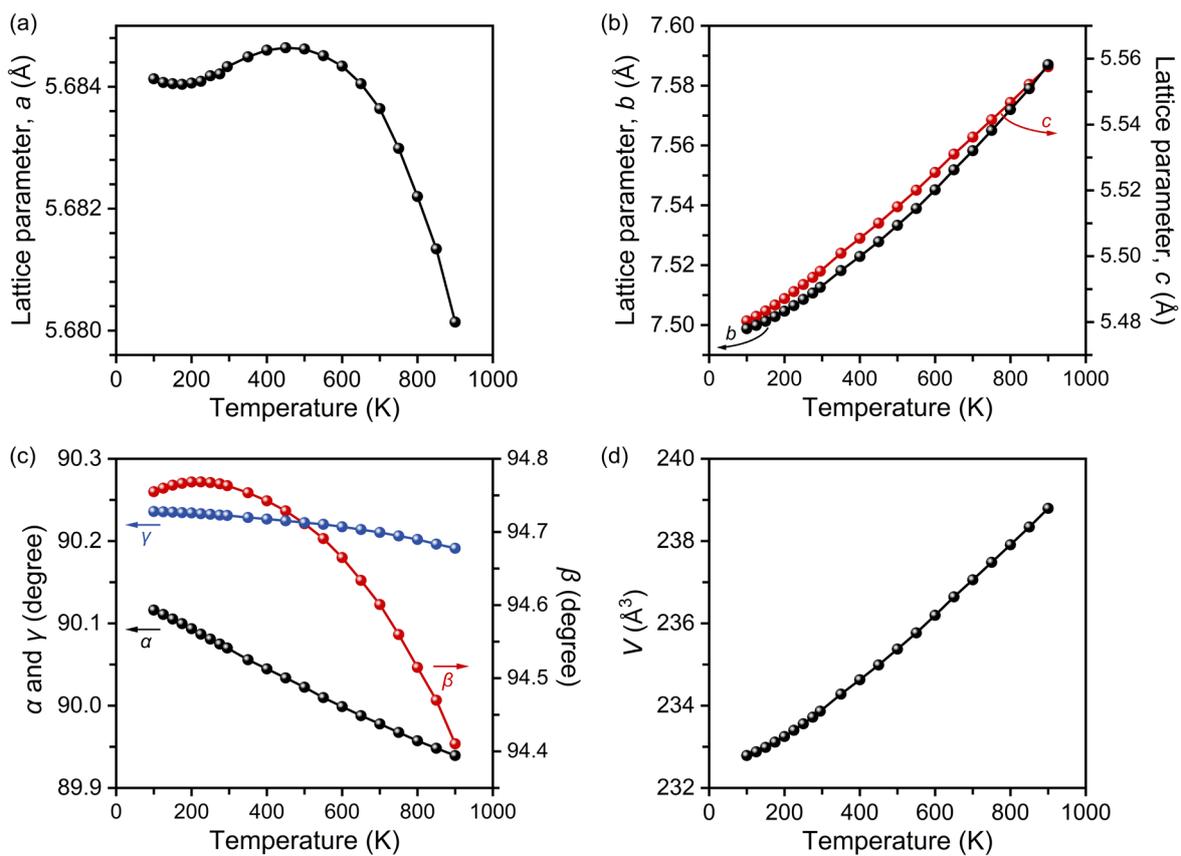
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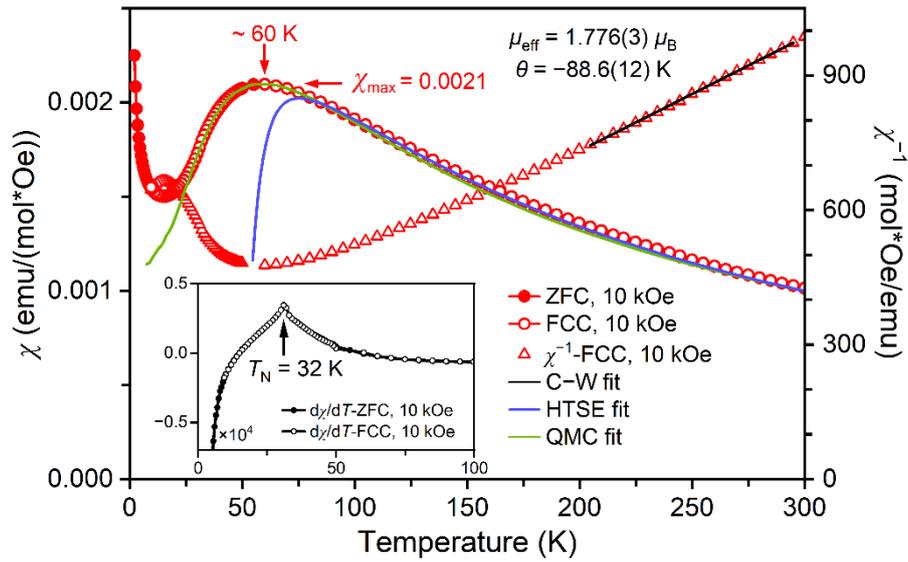
646 **FIG. 1.** Fragments of the experimental (black circles), calculated (red line), and  
 647 difference (blue line at the bottom) synchrotron X-ray powder diffraction (XRPD)  
 648 patterns of  $\text{Ca}_2\text{CuWO}_6$  at  $T = 295 \text{ K}$ . The tick marks show possible Bragg reflection  
 649 positions of the main perovskite phase (the first black row) and  $\text{CaWO}_4$  impurity (the  
 650 second red row).



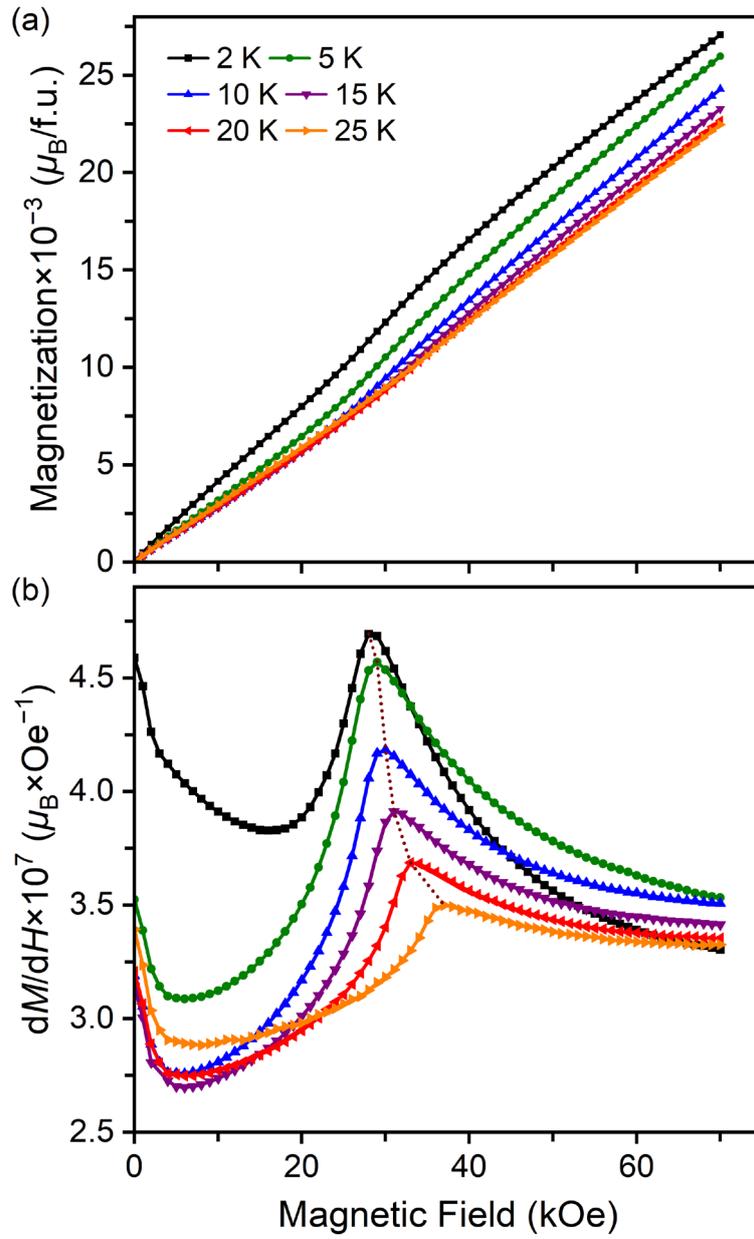
651 **FIG. 2.** (a) Fragments of the  $\text{Ca}_2\text{CuWO}_6$  crystal structure with the  $P-1$  space group in a  
 652 polyhedral presentation. The  $\text{Cu1O}_6$ ,  $\text{Cu2O}_6$ ,  $\text{W1O}_6$ , and  $\text{W2O}_6$  octahedra are shown.  
 653 The Ca atoms are presented by large grey spheres. Two types of octahedral  
 654 arrangements along the  $b$  axis with (b)  $\text{Cu2O}_6$ - $\text{W1O}_6$  and (c)  $\text{Cu1O}_6$ - $\text{W2O}_6$  types.



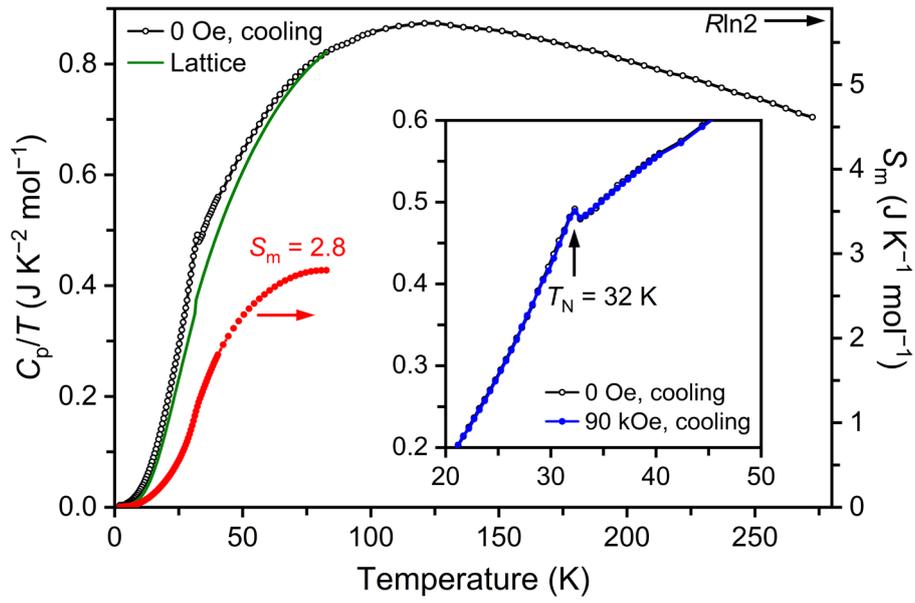
655 **FIG. 3.** Unit-cell parameters ( $a$ ,  $b$ ,  $c$ ,  $\alpha$ ,  $\beta$ , and  $\gamma$ ) and volume ( $V$ ) as a function of  
656 temperature for  $\text{Ca}_2\text{CuWO}_6$  from synchrotron XRPD data measured on heating.



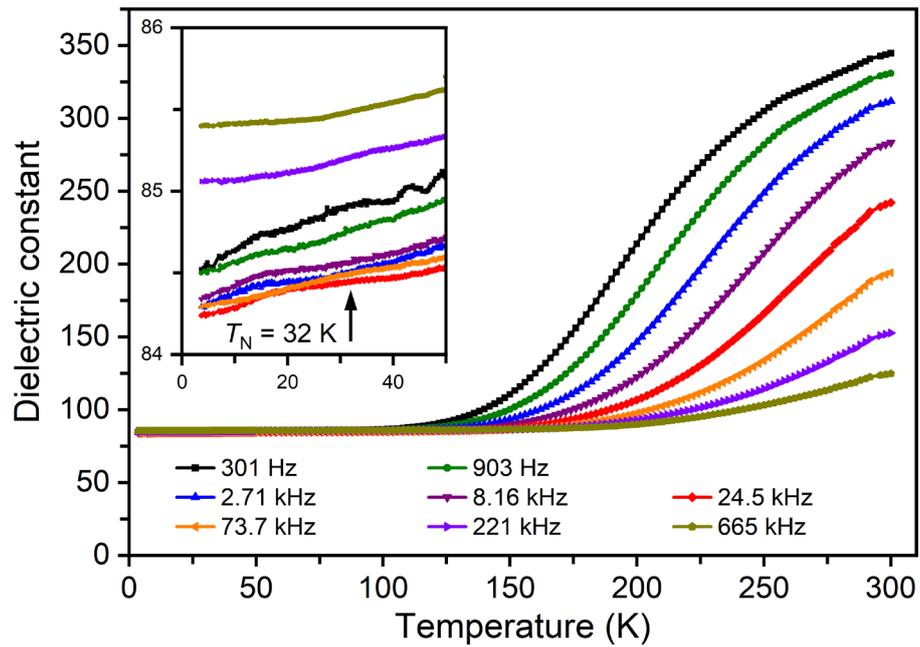
657 **FIG. 4.** ZFC (filled red circles) and FCC (empty red circles) dc magnetic susceptibility  
658 ( $\chi = M/H$ ) curves for  $\text{Ca}_2\text{CuWO}_6$  measured at 10 kOe. The black line shows the Curie–  
659 Weiss fit (C–W fit) of the FCC  $\chi^{-1}$  versus  $T$  curve. The HTSE and QMC fits, as  
660 described in the text, are also shown. The inset presents the  $d\chi/dT$  versus  $T$  curves.  
661



662 **FIG. 5.** (a)  $M$  versus  $H$  curves and (b) derivative of magnetization curves ( $dM/dH$   
 663 versus  $H$ ) for  $\text{Ca}_2\text{CuWO}_6$  at different temperatures. The dotted line is drawn for eye.

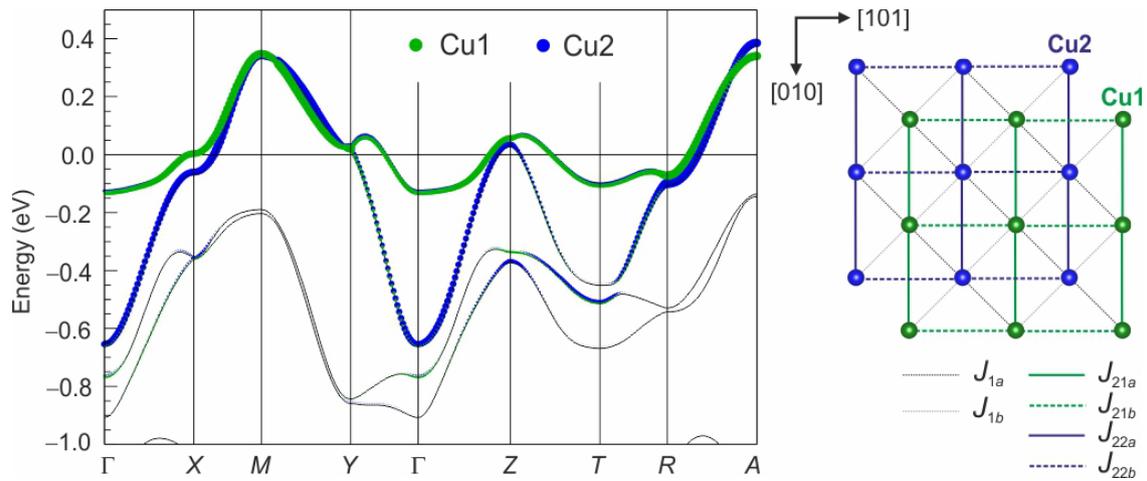


664 **FIG. 6.** Specific heat data,  $C_p/T$  (empty circles; measured on cooling at  $H = 0$  Oe) and  
 665 magnetic entropy,  $S_m$  (filled red circles), of  $\text{Ca}_2\text{CuWO}_6$  as a function of  $T$  are shown in  
 666 the left and right  $y$  axes, respectively. The green line shows the estimated lattice  
 667 contribution (see the text). The inset presents the  $C_p/T$  versus  $T$  measured on cooling at  
 668  $H = 0$  Oe (empty circles) and 90 kOe (filled blue circles) in the vicinity of  $T_N$ .



669 **FIG. 7.** Frequency-dependent (from 301 Hz to 665 kHz) dielectric constant of  
 670  $\text{Ca}_2\text{CuWO}_6$  as a function of temperature measured at  $H = 0$  Oe on heating. The inset  
 671 shows a zoomed part in the vicinity of  $T_N$ .

672



673

674 **FIG. 8.** Left: PBE band structure of  $\text{Ca}_2\text{CuWO}_6$  with the band weights due to the Cu1  
 675 and Cu2  $d_{x^2-y^2}$  orbitals. The Fermi level is at zero energy. Right: spin lattice of  
 676  $\text{Ca}_2\text{CuWO}_6$  (see Table IV for the list of the exchange couplings).

677 **TABLE I.** Structure parameters of  $\text{Ca}_2\text{CuWO}_6$  at 295 K from synchrotron X-ray powder  
678 diffraction data.

site	WP	$g$	$x$	$y$	$z$	$B_{\text{iso}} (\text{\AA}^2)$
Ca1	$2i$	1	0.0547(5)	0.2522(3)	1.0021(5)	0.71(5)
Ca2	$2i$	1	0.5519(5)	0.2497(3)	0.5150(5)	0.77(5)
Cu1	$1e$	1	0.5	0.5	0	1.12(8)
Cu2	$1b$	1	0	0	0.5	0.11(6)
W1	$1d$	1	0.5	0	0	0.44(2)
W2	$1g$	1	0	0.5	0.5	0.23(2)
O1	$2i$	1	0.4720(12)	0.2454(11)	0.0867(13)	0.25(14)
O2	$2i$	1	0.9736(13)	0.2590(11)	0.4270(14)	0.59(15)
O3	$2i$	1	0.3119(11)	0.0467(9)	0.7016(13)	0.59(14)
O4	$2i$	1	0.7059(12)	0.5521(10)	0.3037(13)	0.92(15)
O5	$2i$	1	0.2283(12)	0.9438(9)	0.1658(13)	0.96(15)
O6	$2i$	1	0.8353(12)	0.4563(10)	0.7766(12)	0.73(13)

679 Synchrotron powder X-ray diffraction ( $\lambda = 0.61974 \text{ \AA}$ );  $d$  range used in the refinement:  
680 from 0.532  $\text{\AA}$  to 8.879  $\text{\AA}$ . Crystal system: triclinic; space group:  $P-1$  (No. 2, origin choice  
681 1),  $Z = 2$ . Molecular weight: 423.5384 g/mol.  $a = 5.68435(1) \text{ \AA}$ ,  $b = 7.51261(2) \text{ \AA}$ ,  $c =$   
682  $5.49548(1) \text{ \AA}$ ,  $\alpha = 90.0703(2)^\circ$ ,  $\beta = 94.7634(2)^\circ$ ,  $\gamma = 90.2310(2)^\circ$ , and  $V = 233.8677(8)$   
683  $\text{\AA}^3$ ;  $R_{\text{wp}} = 5.94\%$ ,  $R_{\text{p}} = 4.50\%$ ,  $R_{\text{I}} = 2.26\%$ , and  $R_{\text{F}} = 1.19\%$ . WP = Wyckoff position.  $g$  is  
684 the occupation factor.

685 **TABLE II.** Selected bond lengths (Å), angles (deg), octahedral distortion parameters  
 686 ( $\Delta$ ), and bond valence sums (BVS) for  $\text{Ca}_2\text{CuWO}_6$  at  $T = 295$  K.

Ca1–O1	2.379(8)	Ca2–O1	2.360(8)
Ca1–O2	2.418(9)	Ca2–O2	2.485(8)
Ca1–O3	2.768(8)	Ca2–O3	2.670(8)
Ca1–O4	2.696(8)	Ca2–O3	2.336(8)
Ca1–O5	2.651(8)	Ca2–O4	2.719(8)
Ca1–O5	2.311(8)	Ca2–O4	2.362(8)
Ca1–O6	2.554(8)	Ca2–O5	2.531(8)
Ca1–O6	2.282(8)	Ca2–O6	2.581(8)
BVS (Ca1)	2.06	BVS (Ca2)	1.99
Cu1–O1 $\times 2$	1.980(8)	Cu2–O2 $\times 2$	1.991(8)
Cu1–O4 $\times 2$	2.008(7)	Cu2–O3 $\times 2$	2.040(7)
Cu1–O6 $\times 2$	2.375(7)	Cu2–O5 $\times 2$	2.374(7)
BVS (Cu1)	2.01	BVS (Cu2)	1.92
$\Delta$ (Cu1O <sub>6</sub> )	$7.2 \times 10^{-3}$	$\Delta$ (Cu2O <sub>6</sub> )	$6.4 \times 10^{-3}$
W1–O5 $\times 2$	1.903(7)	W2–O2 $\times 2$	1.857(8)
W1–O1 $\times 2$	1.914(8)	W2–O6 $\times 2$	1.879(7)
W1–O3 $\times 2$	1.915(7)	W2–O4 $\times 2$	1.946(7)
BVS (W1)	6.17	BVS (W2)	6.49
$\Delta$ (W1O <sub>6</sub> )	$8.1 \times 10^{-6}$	$\Delta$ (W2O <sub>6</sub> )	$4.0 \times 10^{-4}$
Cu1–O1–W1	149.3(5)	Cu2–O2–W2	155.0(5)
Cu1–O4–W2	146.5(5)	Cu2–O3–W1	146.3(4)
Cu1–O6–W2	150.3(4)	Cu2–O5–W1	148.1(4)

687  $\text{BVS} = \sum_{i=1}^N v_i$ ,  $v_i = \exp[(R_0 - l_i)/B]$ ,  $N$  is the coordination number,  $l_i$  is a bond length,  $B$   
 688  $= 0.37$ ,  $R_0(\text{Ca}^{2+}) = 1.967$ ,  $R_0(\text{Cu}^{2+}) = 1.679$ ,  $R_0(\text{W}^{6+}) = 1.921$ .  $\Delta = (1/N) \sum_{i=1}^N [(l_i -$   
 689  $l_{\text{av}})/l_{\text{av}}]^2$ , where  $l_{\text{av}} = (1/N) \sum_{i=1}^N l_i$  is the average distance.

690 **TABLE III.** Lattice parameters of  $\text{Ca}_{2-x}\text{Sr}_x\text{CuWO}_6$  ( $x = 0, 0.5, 1, 1.5,$  and  $2$ ) at  $T = 295$   
691 K from the synchrotron<sup>a</sup> and laboratory<sup>b</sup> X-ray powder diffraction data, the Néel  
692 temperatures ( $T_N$ ), and main parameters extracted from the magnetic susceptibility  
693 curves: position ( $T_{\text{max}}$ ) and height ( $\chi_{\text{max}}$ ) of the susceptibility maximum, the effective  
694 moment and Curie-Weiss temperature (from the Curie–Weiss fits), as well as  $J_1$  and  $J_2$   
695 (from the HTSE fits above 80 K for  $x = 0$ , above 90 K for  $x = 0.5, 1, 1.5,$  and above 100  
696 K for  $x = 2$ ; up to 400 K for all samples).

$x$	$0^a$	$0.5^a$	$1^a$	$1.5^a$	$2^b$
space group	$P-1$	$P-1$	$P-1$	$P2_1/n$	$I4/m$
$a$ (Å)	5.68435(1)	5.68310(3)	5.67677(3)	5.67514(2)	5.4284(1)
$b$ (Å)	7.51261(2)	7.56989(3)	7.62931(4)	7.68165(3)	$= a$
$c$ (Å)	5.49548(1)	5.56063(3)	5.62368(3)	5.65900(3)	8.4155(2)
$\alpha$ (°)	90.0703(2)	89.9575(6)	89.8954(5)	90	90
$\beta$ (°)	94.7634(2)	94.8187(4)	94.7683(3)	94.8384(3)	90
$\gamma$ (°)	90.2310(2)	90.2037(4)	90.1552(5)	90	90
$V$ (Å <sup>3</sup> )	233.8677(8)	238.374(2)	242.717(2)	245.822(2)	247.986(9)
$\mu_{\text{eff}}$ ( $\mu_B/\text{f.u.}$ )	1.776(3)	1.693(4)	1.689(3)	1.907(9)	1.649(9)
$\theta$ (K)	−88.6(12)	−83.6(9)	−101.2(11)	−163(4)	−155(4)
$T_N$ (K)	32	32	33	34	24 [34]
$\chi_{\text{max}}$ (emu/(mol*Oe))	0.0021	0.0020	0.0017	0.0017	0.0013
$T_{\text{max}}$ (K)	60	55	65	72	72
$T_{\text{max}}/ \theta $	0.677	0.658	0.642	0.442	0.465
$J_1$ (K)	18	15	9	6	6
$J_2$ (K)	65	68	78	85	97

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699 **TABLE IV.** Magnetic couplings in  $\text{Ca}_2\text{CuWO}_6$ : interatomic distances  $d_i$ , hopping  
700 parameters  $t_i$  (from the Wannier fits), and exchange couplings  $J_i$  (from the mapping  
701 analysis).

		$d_i$ (Å)	$t_i$ (meV)	$J_i$ (K)
$J_{1a}$	Cu1–Cu2	5.323	43	9
$J_{1b}$	Cu1–Cu2	5.343	43	9
$J_{21a}$	Cu1–Cu1	7.513	78	54
$J_{21b}$	Cu1–Cu1	7.571	80	51
$J_{22a}$	Cu2–Cu2	7.513	95	80
$J_{22b}$	Cu2–Cu2	7.571	79	50

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